

## **Right to Know** Health Hazardous Substance Fact Sheet

**ACETIC ACID** Common Name:

Synonyms: Glacial Acetic Acid; Ethanoic Acid; Ethylic Acid

Chemical Name: Acetic Acid

Date: July 2007 Revision: March 2016

## **Description and Use**

Acetic Acid is a colorless liquid with a strong vinegar-like odor. It can also be an ice-like solid below 62°F (17°C). It is used in making drugs, dves, plastics, food additives and insecticides.

#### ► ODOR THRESHOLD = 0.48 to 1.0 ppm

▶ The range of accepted odor threshold values is guite broad. Caution should be used in relying on odor alone as a warning of potentially hazardous exposures.

#### **Reason for Citation**

- ▶ Acetic Acid is on the Right to Know Hazardous Substance List because it is cited by OSHA, ACGIH, DOT, NIOSH, NFPA and EPA.
- ▶ This chemical is on the Special Health Hazard Substance List.

#### SEE GLOSSARY ON PAGE 5.

#### **FIRST AID**

#### **Eye Contact**

▶ Immediately flush with large amounts of cool water. Continue for at least 30 minutes, occasionally lifting upper and lower lids. Remove contact lenses, if worn, while rinsing. Immediate medical attention is necessary.

## **Skin Contact**

▶ Quickly remove contaminated clothing. Immediately wash contaminated skin with large amounts of soap and water. Seek medical attention immediately.

#### Inhalation

- ▶ Remove the person from exposure.
- ▶ Begin rescue breathing (using universal precautions) if breathing has stopped and CPR if heart action has stopped.
- ▶ Transfer promptly to a medical facility.
- ▶ Medical observation is recommended for 24 to 48 hours after breathing overexposure, as pulmonary edema may be delayed.

#### **EMERGENCY NUMBERS**

Poison Control: 1-800-222-1222 CHEMTREC: 1-800-424-9300 NJDEP Hotline: 1-877-927-6337

National Response Center: 1-800-424-8802

CAS Number: 64-19-7 RTK Substance Number: 0004 **DOT Number:** UN 2789

#### **EMERGENCY RESPONDERS >>>> SEE BACK PAGE**

Hazard Summary			
Hazard Rating	NJDHSS	NFPA	
HEALTH	-	3	
FLAMMABILITY	-	2	
REACTIVITY	-	0	

**CORROSIVE COMBUSTIBLE** POISONOUS GASES ARE PRODUCED IN FIRE CONTAINERS MAY EXPLODE IN FIRE

Hazard Rating Key: 0=minimal; 1=slight; 2=moderate; 3=serious;

- ▶ Acetic Acid can affect you when inhaled.
- ► Contact can severely irritate and burn the skin and eyes leading to eye damage.
- ▶ Inhaling Acetic Acid can irritate the nose and throat.
- ▶ Inhaling Acetic Acid can irritate the lungs. Higher exposures may cause a build-up of fluid in the lungs (pulmonary edema), a medical emergency.
- ▶ Repeated exposure can cause thickening and cracking of the skin.

## **Workplace Exposure Limits**

OSHA: The legal airborne permissible exposure limit (PEL) is 10 ppm averaged over an 8-hour workshift.

NIOSH: The recommended airborne exposure limit (REL) is **10 ppm** averaged over a 10-hour workshift and 15 ppm, not to be exceeded during any 15-minute work period.

ACGIH: The threshold limit value (TLV) is 10 ppm averaged over an 8-hour workshift and 15 ppm as a STEL (short-term exposure limit).

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## **Determining Your Exposure**

- ▶ Read the product manufacturer's Material Safety Data Sheet (MSDS) and the label to determine product ingredients and important safety and health information about the product mixture.
- ► For each individual hazardous ingredient, read the New Jersey Department of Health Hazardous Substance Fact Sheet, available on the RTK website (<a href="http://nj.gov/health/workplacehealthandsafety/right-to-know/">http://nj.gov/health/workplacehealthandsafety/right-to-know/</a>) or in your facility's RTK Central File or Hazard Communication Standard file.
- ➤ You have a right to this information under the New Jersey Worker and Community Right to Know Act, the Public Employees Occupational Safety and Health (PEOSH) Act if you are a public worker in New Jersey, and under the federal Occupational Safety and Health Act (OSHA) if you are a private worker.
- ► The New Jersey Right to Know Act requires most employers to label chemicals in the workplace and requires public employers to provide their employees with information concerning chemical hazards and controls. The federal OSHA Hazard Communication Standard (29 CFR 1910.1200) requires private employers to provide similar information and training to their employees.

This Fact Sheet is a summary of available information regarding the health hazards that may result from exposure. Duration of exposure, concentration of the substance and other factors will affect your susceptibility to any of the potential effects described below.

#### **Health Hazard Information**

#### **Acute Health Effects**

The following acute (short-term) health effects may occur immediately or shortly after exposure to **Acetic Acid**:

- ► Contact can severely irritate and burn the skin and eyes leading to eye damage.
- ▶ Inhaling Acetic Acid can irritate the nose and throat.
- ▶ Inhaling Acetic Acid can irritate the lungs causing coughing and/or shortness of breath. Higher exposures may cause a build-up of fluid in the lungs (pulmonary edema), a medical emergency, with severe shortness of breath.

#### **Chronic Health Effects**

The following chronic (long-term) health effects can occur at some time after exposure to **Acetic Acid** and can last for months or years:

#### Cancer Hazard

According to the information presently available to the New Jersey Department of Health, Acetic Acid has been tested and has not been shown to cause cancer in animals.

#### Reproductive Hazard

▶ According to the information presently available to the New Jersey Department of Health, **Acetic Acid** has been tested and has not been shown to affect reproduction.

#### Other Effects

- Acetic Acid can cause bronchitis to develop with cough, phlegm and/or shortness of breath.
- ▶ Repeated exposure can cause thickening and cracking of the skin, particularly the skin of the hands.

#### Medical

## **Medical Testing**

For those with frequent or potentially high exposure (half the PEL or greater), the following are recommended before beginning work and at regular times after that:

▶ Lung function tests

If symptoms develop or overexposure is suspected, the following is recommended:

► Chest x-ray

Any evaluation should include a careful history of past and present symptoms with an exam. Medical tests that look for damage already done are <u>not</u> a substitute for controlling exposure.

Request copies of your medical testing. You have a legal right to this information under the OSHA Access to Employee Exposure and Medical Records Standard (29 CFR 1910.1020).

#### Mixed Exposures

▶ Because smoking can cause heart disease, as well as lung cancer, emphysema, and other respiratory problems, it may worsen respiratory conditions caused by chemical exposure. Even if you have smoked for a long time, stopping now will reduce your risk of developing health problems.

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## **Workplace Controls and Practices**

Very toxic chemicals, or those that are reproductive hazards or sensitizers, require expert advice on control measures if a less toxic chemical cannot be substituted. Control measures include: (1) enclosing chemical processes for severely irritating and corrosive chemicals, (2) using local exhaust ventilation for chemicals that may be harmful with a single exposure, and (3) using general ventilation to control exposures to skin and eye irritants. For further information on workplace controls, consult the NIOSH document on Control Banding at <a href="https://www.cdc.gov/niosh/topics/ctrlbanding/">www.cdc.gov/niosh/topics/ctrlbanding/</a>.

The following work practices are also recommended:

- ▶ Label process containers.
- ▶ Provide employees with hazard information and training.
- ▶ Monitor airborne chemical concentrations.
- Use engineering controls if concentrations exceed recommended exposure levels.
- ▶ Provide eye wash fountains and emergency showers.
- Wash or shower if skin comes in contact with a hazardous material
- ▶ Always wash at the end of the workshift.
- Change into clean clothing if clothing becomes contaminated.
- ▶ Do not take contaminated clothing home.
- ▶ Get special training to wash contaminated clothing.
- ▶ Do not eat, smoke, or drink in areas where chemicals are being handled, processed or stored.
- Wash hands carefully before eating, smoking, drinking, applying cosmetics or using the toilet.

## **Personal Protective Equipment**

The OSHA Personal Protective Equipment Standard (29 CFR 1910.132) requires employers to determine the appropriate personal protective equipment for each hazard and to train employees on how and when to use protective equipment.

The following recommendations are only guidelines and may not apply to every situation.

#### **Gloves and Clothing**

- Avoid skin contact with Acetic Acid. Wear personal protective equipment made from material which can not be permeated and/or degraded by this substance. Safety equipment suppliers/manufacturers can provide recommendations on the most protective glove/clothing material for your operation.
- ➤ Safety equipment manufacturers recommend *Neoprene* and *Butyl Rubber* for gloves and DuPont *Tychem*® *CPF4*, *Responder*®, *TK*, *Reflector*® and CHEMFAB *Challenger*® 4000 as protective materials for clothing.
- ▶ All protective clothing (suits, gloves, footwear, headgear) should be clean, available each day, and put on before work.

#### **Eye Protection**

- Wear indirect-vent, impact and splash resistant goggles when working with liquids.
- Contact lenses should not be worn when working with this substance.
- ► Wear a face shield along with goggles when working with corrosive, highly irritating or toxic substances.

#### **Respiratory Protection**

*Improper use of respirators is dangerous.* Such equipment should only be used if the employer has a written program that takes into account workplace conditions, requirements for worker training, respirator fit testing, and medical exams as described in the OSHA Respiratory Protection Standard (29 CFR 1910.134).

- ► Where the potential exists for exposure over 10 ppm, use a NIOSH approved full facepiece respirator with an organic vapor cartridge. Increased protection is obtained from full facepiece powered-air purifying respirators.
- ▶ If while wearing a filter or cartridge respirator you can smell, taste, or otherwise detect **Acetic Acid**, or if while wearing particulate filters abnormal resistance to breathing is experienced, or eye irritation occurs while wearing a full facepiece respirator, leave the area immediately. Check to make sure the respirator-to-face seal is still good. If it is, replace the filter or cartridge. If the seal is no longer good, you may need a new respirator.
- ▶ Be sure to consider all potential exposures in your workplace. You may need a combination of filters, prefilters or cartridges to protect against different forms of a chemical (such as vapor and mist) or against a mixture of chemicals.
- ▶ Where the potential exists for exposure over 100 ppm, use a NIOSH approved supplied-air respirator with a full facepiece operated in a pressure-demand or other positive-pressure mode. For increased protection use in combination with an auxiliary self-contained breathing apparatus operated in a pressure-demand or other positive-pressure mode.

## Fire Hazards

If employees are expected to fight fires, they must be trained and equipped as stated in the OSHA Fire Brigades Standard (29 CFR 1910.156).

- ▶ Acetic Acid is a COMBUSTIBLE LIQUID.
- ▶ Use dry chemical, CO<sub>2</sub>, water spray, alcohol-resistant foam or other foaming agents.
- ▶ POISONOUS GASES ARE PRODUCED IN FIRE.
- ► CONTAINERS MAY EXPLODE IN FIRE.
- Use water spray to keep fire-exposed containers cool and disperse vapors.
- Vapor is heavier than air and may explode if ignited in an enclosed space.

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## **Spills and Emergencies**

If employees are required to clean-up spills, they must be properly trained and equipped. The OSHA Hazardous Waste Operations and Emergency Response Standard (29 CFR 1910.120) may apply.

If Acetic Acid is spilled or leaked, take the following steps:

- Evacuate personnel and secure and control entrance to the area.
- ▶ Eliminate all ignition sources.
- Absorb liquids in vermiculite, dry sand, earth, or a similar material and deposit in sealed containers.
- ▶ Ventilate and wash area after clean-up is complete.
- ▶ Use water spray to disperse vapors.
- ▶ It may be necessary to contain and dispose of **Acetic Acid** as a HAZARDOUS WASTE. Contact your state Department of Environmental Protection (DEP) or your regional office of the federal Environmental Protection Agency (EPA) for specific recommendations.

## **Handling and Storage**

Prior to working with **Acetic Acid** you should be trained on its proper handling and storage.

- ► Acetic Acid reacts violently with OXIDIZING AGENTS (such as PERCHLORATES, PEROXIDES, PERMANGANATES, CHLORATES, NITRATES, CHLORINE, BROMINE and FLUORINE) and STRONG BASES (such as SODIUM HYDROXIDE and POTASSIUM HYDROXIDE).
- Acetic Acid attacks many METALS forming flammable and explosive Hydrogen gas.
- Acetic Acid is incompatible with CHROMIC ACID; SODIUM PEROXIDE; NITRIC ACID; ACETONE; and AMMONIUM NITRATE.
- ► Store in tightly closed containers in a cool, well-ventilated area away from HEAT and MOISTURE.
- Sources of ignition, such as smoking and open flames, are prohibited where Acetic Acid is used, handled, or stored in a manner that could create a potential fire or explosion hazard.

# Occupational Health Information Resources

The New Jersey Department of Health and Occupational Health Service, offers multiple services in occupational health. These services include providing informational resources, educational materials, public presentations, and industrial hygiene and medical investigations and evaluations.

#### For more information, please contact:

New Jersey Department of Health Right to Know

PO Box 368

Trenton, NJ 08625-0368 Phone: 609-984-2202 Fax: 609-984-7407 E-mail: rtk@doh.nj.gov

Web address:

http://nj.gov/health/workplacehealthandsafety/right-to-

know/

The Right to Know Hazardous Substance Fact Sheets are not intended to be copied and sold for commercial purposes.

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#### **GLOSSARY**

**ACGIH** is the American Conference of Governmental Industrial Hygienists. They publish guidelines called Threshold Limit Values (TLVs) for exposure to workplace chemicals.

**Boiling point** is the temperature at which a substance can change its physical state from a liquid to a gas.

A **carcinogen** is a substance that causes cancer.

The **CAS number** is unique, identifying number, assigned by the Chemical Abstracts Service, to a specific chemical.

**CFR** is the Code of Federal Regulations, which are the regulations of the United States government.

A **combustible** substance is a solid, liquid or gas that will burn.

A **corrosive** substance is a gas, liquid or solid that causes destruction of human skin or severe corrosion of containers.

**DEP** is the New Jersey Department of Environmental Protection.

**DOT** is the Department of Transportation, the federal agency that regulates the transportation of chemicals.

**EPA** is the Environmental Protection Agency, the federal agency responsible for regulating environmental hazards.

**ERG** is the Emergency Response Guidebook. It is a guide for emergency responders for transportation emergencies involving hazardous substances.

A fetus is an unborn human or animal.

A **flammable** substance is a solid, liquid, vapor or gas that will ignite easily and burn rapidly.

The **flash point** is the temperature at which a liquid or solid gives off vapor that can form a flammable mixture with air.

**IARC** is the International Agency for Research on Cancer, a scientific group.

**Ionization Potential** is the amount of energy needed to remove an electron from an atom or molecule. It is measured in electron volts.

**IRIS** is the Integrated Risk Information System database maintained by federal EPA. The database contains information on human health effects that may result from exposure to various chemicals in the environment.

**LEL** or **Lower Explosive Limit** is the lowest concentration of a combustible substance (gas or vapor) in the air capable of continuing an explosion.

mg/m<sup>3</sup> means milligrams of a chemical in a cubic meter of air. It is a measure of concentration (weight/volume).

A **mutagen** is a substance that causes mutations. A **mutation** is a change in the genetic material in a body cell. Mutations can lead to birth defects, miscarriages, or cancer.

**NFPA** is the National Fire Protection Association. It classifies substances according to their fire and explosion hazard.

**NIOSH** is the National Institute for Occupational Safety and Health. It tests equipment, evaluates and approves respirators, conducts studies of workplace hazards, and proposes standards to OSHA.

**NTP** is the National Toxicology Program which tests chemicals and reviews evidence for cancer.

**OSHA** is the federal Occupational Safety and Health Administration, which adopts and enforces health and safety standards.

**PEOSHA** is the New Jersey Public Employees Occupational Safety and Health Act, which adopts and enforces health and safety standards in public workplaces.

**Permeated** is the movement of chemicals through protective materials.

**PIH** is a DOT designation for chemicals which are Poison Inhalation Hazards.

**ppm** means parts of a substance per million parts of air. It is a measure of concentration by volume in air.

A **reactive** substance is a solid, liquid or gas that releases energy under certain conditions.

**STEL** is a Short Term Exposure Limit which is usually a 15-minute exposure that should not be exceeded at any time during a work day.

A **teratogen** is a substance that causes birth defects by damaging the fetus.

**UEL** or **Upper Explosive Limit** is the highest concentration in air above which there is too much fuel (gas or vapor) to begin a reaction or explosion.

**Vapor Density** is the ratio of the weight of a given volume of one gas to the weight of another (usually *Hydrogen*), at the same temperature and pressure.

The **vapor pressure** is a measure of how readily a liquid or a solid mixes with air at its surface. A higher vapor pressure indicates a higher concentration of the substance in air and therefore increases the likelihood of breathing it in.



## **Right to Know Hazardous Substance Fact Sheet**

Emergency Responders Quick Reference

Common Name: ACETIC ACID

Synonyms: Glacial Acetic Acid; Ethanoic Acid; Ethylic Acid

CAS No: 64-19-7

Molecular Formula: CH<sub>3</sub> COOH or C<sub>2</sub>H<sub>4</sub>O<sub>2</sub>

RTK Substance No: 0004

Description: Colorless liquid with vinegar odor

HAZARD DATA			
Hazard Rating	Firefighting	Reactivity	
3 - Health	Use dry chemical, CO <sub>2</sub> , water spray, alcohol- resistant foam or other foaming agent.	Reacts violently with OXIDIZING AGENTS (such as	
2 - Fire	POISONOUS GASES ARE PRODUCED IN FIRE.	PERCHLORATES, PEROXIDES, PERMANGANATES, CHLORATES, NITRATES, CHLORINE, BROMINE and	
0 - Reactivity	CONTAINERS MAY EXPLODE IN FIRE. Use water spray to cool containers and disperse	FLUORINE) and STRONG BASES (such as SODIUM HYDROXIDE and POTASSIUM HYDROXIDE).	
<b>DOT#</b> : UN 2789	vapors.	Acetic Acid attacks many METALS forming flammable	
<b>ERG Guide #:</b> 132	Vapor is heavier than air and may explode if ignited in an enclosed space.	and explosive <i>Hydrogen gas</i> .	
Hazard Class: 8		Incompatible with CHROMIC ACID; SODIUM PEROXIDE;	
(Corrosive)		NITRIC ACID; ACETONE; and AMMONIUM NITRATE.	

#### SPILL/LEAKS

Isolation Distance: 50 to 100 meters

(160 to 330 feet)

Absorb liquids in vermiculite, dry sand, earth, or a similar material and deposit in sealed containers.

Use water spray to disperse vapors.

Soda Ash (Sodium Carbonate) can be used to

neutralize spills.

This substance is harmful to aquatic organisms.

## **PHYSICAL PROPERTIES**

Odor Threshold: 0.48 to 1.0 ppm Flash Point: 103°F (39°C)

LEL: 4%

UEL: 19.99% Vapor Density: 2.1 (air = 1)

Vapor Pressure: 15 mm Hg at 77°F (25°C)

Water Solubility: Soluble

**Boiling Point:** 244°F (118°C) **Ionization Potential:** 10.66 eV

## **EXPOSURE LIMITS**

OSHA: 10 ppm 8-hr TWA

NIOSH: 10 ppm 10-hr TWA, 15 ppm STEL ACGIH: 10 ppm 10-hr TWA, 15 ppm STEL

IDLH: 50 ppm ERPG-1: 5 ppm ERPG-2: 35 ppm ERPG-3: 250 ppm

## PROTECTIVE EQUIPMENT

Gloves: Neoprene, Butyl Rubber

Coverall: DuPont Tychem® CPF4, Responder®, TK, Reflector®;

CHEMFAB Challenger® 4000.

**Boot:** Neoprene or Butyl

**Respirator:** >10 ppm - air purifying respirator with organic vapor

cartridges, >100 ppm - supplied air

## **HEALTH EFFECTS**

Eyes: Irritation, burns, possible eye damage

Skin: Irritation, burns

Acute: Nose, throat and lung irritation,

pulmonary edema, coughing, shortness

of breath

Chronic: Bronchitis, thickening and cracking of

the skin

## FIRST AID AND DECONTAMINATION

**Remove** the person from exposure.

**Flush** eyes with large amounts of water for at least 30 minutes. Remove contact lenses if worn. Immediate medical attention is necessary. **Remove** contaminated clothing and wash contaminated skin with soap

and water.

Begin artificial respiration if breathing has stopped and CPR if

necessary.

**Medical** observation is recommended for 24 to 48 hours as symptoms may be delayed.