Right to Know
Hazardous Substance Fact Sheet

Common Name: ALUMINUM FLUORIDE

Synonyms: Aluminum Trifluoride
Chemical Name: Aluminum Fluoride (AlF₃)
Date: December 1998  Revision: October 2008

CAS Number: 7784-18-1
RTK Substance Number: 0059
DOT Number: UN 1759

Description and Use
Aluminum Fluoride is an odorless, white or colorless, crystalline (sand-like) powder. It is used in the production of Aluminum, as a flux in welding applications, and in ceramic glazes and enamels.

Reasons for Citation
- Aluminum Fluoride is on the Right to Know Hazardous Substance List because it is cited by OSHA, ACGIH, DOT and NIOSH.
- This chemical is on the Special Health Hazard Substance List.

First Aid

Eye Contact
- Immediately flush with large amounts of water for at least 15 minutes, lifting upper and lower lids. Remove contact lenses, if worn, while rinsing.

Skin Contact
- Quickly remove contaminated clothing. Immediately wash contaminated skin with large amounts of water.

Inhalation
- Remove the person from exposure.
- Begin rescue breathing (using universal precautions) if breathing has stopped and CPR if heart action has stopped.
- Transfer promptly to a medical facility.

Emergency Numbers
Poison Control: 1-800-222-1222
CHEMTREC: 1-800-424-9300
NJDEP Hotline: 1-877-927-6337
National Response Center: 1-800-424-8802

Hazard Summary

<table>
<thead>
<tr>
<th>Hazard Rating</th>
<th>NJDOH</th>
<th>NFPA</th>
</tr>
</thead>
<tbody>
<tr>
<td>HEALTH</td>
<td>2</td>
<td>-</td>
</tr>
<tr>
<td>FLAMMABILITY</td>
<td>0</td>
<td>-</td>
</tr>
<tr>
<td>REACTIVITY</td>
<td>0</td>
<td>-</td>
</tr>
<tr>
<td>CORROSIVE</td>
<td></td>
<td></td>
</tr>
<tr>
<td>POISONOUS GASES ARE PRODUCED IN FIRE</td>
<td></td>
<td></td>
</tr>
<tr>
<td>DOES NOT BURN</td>
<td></td>
<td></td>
</tr>
</tbody>
</table>

Hazard Rating Key: 0=minimal; 1=slight; 2=moderate; 3=serious; 4=severe

- Aluminum Fluoride can affect you when inhaled and may be absorbed through the skin.
- Contact can irritate the skin and eyes.
- Inhaling Aluminum Fluoride can irritate the nose, throat and lungs.
- Aluminum Fluoride may cause an asthma-like allergy.
- Exposure can cause headache, nausea and vomiting.
- Long term exposure may affect joints and ligaments causing stiffness and disability.
- Very high exposure can cause Fluoride poisoning with stomach pain, weakness, convulsions, collapse and death.
- Repeated, very high exposures can cause the deposit of Fluoride in the bones and teeth, a condition called Fluorosis. This effect does NOT occur at the level of Fluoride in water to prevent cavities in teeth.
- Aluminum Fluoride is a DOT CORROSIVE material.

Workplace Exposure Limits

The following exposure limit is for Aluminum soluble salts (measured as Aluminum):

- NIOSH: The recommended airborne exposure limit (REL) is 2 mg/m³ averaged over a 10-hour workshift.

The exposure limits for Fluorides are:

- OSHA: The legal airborne permissible exposure limit (PEL) is 2.5 mg/m³ averaged over an 8-hour workshift.

- NIOSH: The recommended airborne exposure limit (REL) is 2.5 mg/m³ averaged over a 10-hour workshift.

- ACGIH: The threshold limit value (TLV) is 2.5 mg/m³ averaged over an 8-hour workshift.

- The above exposure limits are for air levels only. When skin contact also occurs, you may be overexposed, even though air levels are less than the limits listed above.
Determine Your Exposure

- Read the product manufacturer's Material Safety Data Sheet (MSDS) and the label to determine product ingredients and important safety and health information about the product mixture.
- For each individual hazardous ingredient, read the New Jersey Department of Health Hazardous Substance Fact Sheet, available on the RTK website (www.nj.gov/health/eoh/rtkweb) or in your facility's RTK Central File or Hazard Communication Standard file.
- You have a right to this information under the New Jersey Worker and Community Right to Know Act, the Public Employees Occupational Safety and Health (PEOSH) Act if you are a public worker in New Jersey, and under the federal Occupational Safety and Health Act (OSHA) if you are a private worker.
- The New Jersey Right to Know Act requires most employers to label chemicals in the workplace and requires public employers to provide their employees with information concerning chemical hazards and controls. The federal OSHA Hazard Communication Standard (29 CFR 1910.1200) and the PEOSH Hazard Communication Standard (N.J.A.C. 12:100-7) require employers to provide similar information and training to their employees.

This Fact Sheet is a summary of available information regarding the health hazards that may result from exposure. Duration of exposure, concentration of the substance and other factors will affect your susceptibility to any of the potential effects described below.

Health Hazard Information

Acute Health Effects
The following acute (short-term) health effects may occur immediately or shortly after exposure to Aluminum Fluoride:

- Contact can irritate the skin and eyes.
- Inhaling Aluminum Fluoride can irritate the nose and throat, causing coughing and wheezing.
- Exposure can cause headache, nausea, vomiting, loss of appetite and sometimes nosebleeds.
- Very high exposure can cause Fluoride poisoning with stomach pain, weakness, convulsions, collapse and death.

Chronic Health Effects
The following chronic (long-term) health effects can occur at some time after exposure to Aluminum Fluoride and can last for months or years:

Cancer Hazard
- According to the information presently available to the New Jersey Department of Health, Aluminum Fluoride has not been tested for its ability to cause cancer in animals.

Reproductive Hazard
- According to the information presently available to the New Jersey Department of Health, Aluminum Fluoride has not been tested for its ability to affect reproduction.

Other Effects
- Aluminum Fluoride can irritate the lungs. Repeated exposure may cause bronchitis to develop with coughing, phlegm, and/or shortness of breath.
- Aluminum Fluoride may cause an asthma-like allergy. Future exposure can cause asthma attacks with shortness of breath, wheezing, coughing, and/or chest tightness.
- Long-term exposure may affect joints and ligaments causing stiffness and disability.
- Repeated, very high exposures can cause the deposit of Fluoride in the bones and teeth, a condition called Fluorosis, causing pain, mottling of the teeth and disability.

Medical

Medical Testing
For frequent or potentially high exposure (half the PEL or greater for Fluorides), the following are recommended before beginning work and at regular times after that:

- Lung function tests. The results may be normal if the person is not having an attack at the time of the test.
- Fluoride level in urine. Levels higher than 4 mg/liter indicate overexposure.

If symptoms develop or overexposure is suspected, the following are recommended:

- Chest x-ray after acute overexposure
- Bone Density (DEXA) Scan

Any evaluation should include a careful history of past and present symptoms with an exam. Medical tests that look for damage already done are not a substitute for controlling exposure.

Request copies of your medical testing. You have a legal right to this information under the OSHA Access to Employee Exposure and Medical Records Standard (29 CFR 1910.1020).

Mixed Exposures
- Smoking can cause heart disease, lung cancer, emphysema, and other respiratory problems. It may worsen respiratory conditions caused by chemical exposure. Even if you have smoked for a long time, stopping now will reduce your risk of developing health problems.
Workplace Controls and Practices

Very toxic chemicals, or those that are reproductive hazards or sensitizers, require expert advice on control measures if a less toxic chemical cannot be substituted. Control measures include: (1) enclosing chemical processes for severely irritating and corrosive chemicals, (2) using local exhaust ventilation for chemicals that may be harmful with a single exposure, and (3) using general ventilation to control exposures to skin and eye irritants. For further information on workplace controls, consult the NIOSH document on Control Banding at www.cdc.gov/niosh/topics/ctrlbanding/.

The following work practices are also recommended:

- Label process containers.
- Provide employees with hazard information and training.
- Monitor airborne chemical concentrations.
- Use engineering controls if concentrations exceed recommended exposure levels.
- Provide eye wash fountains and emergency showers.
- Wash or shower if skin comes in contact with a hazardous material.
- Always wash at the end of the workshift.
- Change into clean clothing if clothing becomes contaminated.
- Do not take contaminated clothing home.
- Get special training to wash contaminated clothing.
- Do not eat, smoke, or drink in areas where chemicals are being handled, processed or stored.
- Wash hands carefully before eating, smoking, drinking, applying cosmetics or using the toilet.

In addition, the following may be useful or required:

- Use a vacuum or a wet method to reduce dust during clean-up. DO NOT DRY SWEEP.
- Use a high efficiency particulate air (HEPA) filter when vacuuming. Do not use a standard shop vacuum.

Personal Protective Equipment

The OSHA Personal Protective Equipment Standard (29 CFR 1910.132) requires employers to determine the appropriate personal protective equipment for each hazard and to train employees on how and when to use protective equipment.

The following recommendations are only guidelines and may not apply to every situation.

Gloves and Clothing

- Avoid skin contact with Aluminum Fluoride. Wear personal protective equipment made from material that can not be permeated or degraded by this substance. Safety equipment suppliers and manufacturers can provide recommendations on the most protective glove and clothing material for your operation.
- Safety equipment manufacturers recommend Neoprene, Natural Rubber, Polyvinyl Chloride and Viton for gloves and DuPont Tyvek®, or the equivalent, as a protective material for clothing.
- All protective clothing (suits, gloves, footwear, headgear) should be clean, available each day, and put on before work.

Eye Protection

- Wear eye protection with side shields or goggles.
- Wear a face shield along with goggles when working with corrosive, highly irritating or toxic substances.

Respiratory Protection

Improper use of respirators is dangerous. Respirators should only be used if the employer has implemented a written program that takes into account workplace conditions, requirements for worker training, respirator fit testing, and medical exams, as described in the OSHA Respiratory Protection Standard (29 CFR 1910.134).

- Where the potential exists for exposure over 2.5 mg/m³ (as Fluorides), use a NIOSH approved negative pressure, air-purifying, particulate filter respirator with an N100 filter. More protection is provided by a full facepiece respirator than by a half-mask respirator, and even greater protection is provided by a powered-air purifying respirator.
- Leave the area immediately if (1) while wearing a filter or cartridge respirator you can smell, taste, or otherwise detect Aluminum Fluoride, (2) while wearing particulate filters abnormal resistance to breathing is experienced, or (3) eye irritation occurs while wearing a full facepiece respirator. Check to make sure the respirator-to-face seal is still good. If it is, replace the filter or cartridge. If the seal is no longer good, you may need a new respirator.
- Consider all potential sources of exposure in your workplace. You may need a combination of filters, prefilters or cartridges to protect against different forms of a chemical (such as vapor and mist) or against a mixture of chemicals.
- Where the potential exists for exposure over 25 mg/m³ (as Fluorides), use a NIOSH approved supplied-air respirator with a full facepiece operated in a pressure-demand or other positive-pressure mode. For increased protection use in combination with an auxiliary self-contained breathing apparatus operated in a pressure-demand or other positive-pressure mode.
- Exposure to 250 mg/m³ (as Fluorides) is immediately dangerous to life and health. If the possibility of exposure above 250 mg/m³ exists, use a NIOSH approved self-contained breathing apparatus with a full facepiece operated in a pressure-demand or other positive-pressure mode equipped with an emergency escape air cylinder.

Fire Hazards

If employees are expected to fight fires, they must be trained and equipped as stated in the OSHA Fire Brigades Standard (29 CFR 1910.156).

- Extinguish fire using an agent suitable for type of surrounding fire. Aluminum Fluoride itself does not burn.
- POISONOUS GASES ARE PRODUCED IN FIRE, including Aluminum Oxides, Fluorine and Hydrogen Fluoride (in the presence of water).
- Use water spray to keep fire-exposed containers cool.
**Spills and Emergencies**

If employees are required to clean-up spills, they must be properly trained and equipped. The OSHA Hazardous Waste Operations and Emergency Response Standard (29 CFR 1910.120) may apply.

If Aluminum Fluoride is spilled, take the following steps:

- Evacuate personnel and secure and control entrance to the area.
- Eliminate all ignition sources.
- Moisten spilled material first or use a HEPA-filter vacuum for clean-up and place into covered containers for disposal.
- Ventilate and wash area after clean-up is complete.
- DO NOT wash into sewer.
- It may be necessary to contain and dispose of Aluminum Fluoride as a HAZARDOUS WASTE. Contact your state Department of Environmental Protection (DEP) or your regional office of the federal Environmental Protection Agency (EPA) for specific recommendations.

**Handling and Storage**

Prior to working with Aluminum Fluoride you should be trained on its proper handling and storage.

- Aluminum Fluoride, in contact with SODIUM and POTASSIUM, is sensitive to impact and a violent reaction may occur.
- Aluminum Fluoride will explode when heated with REDUCING AGENTS (such as LITHIUM, SODIUM, ALUMINUM and their HYDRIDES).
- Aluminum Fluoride is not compatible with STRONG ACIDS (such as HYDROCHLORIC, SULFURIC and NITRIC) and ACID FUMES.
- Store in tightly closed containers in a cool, well-ventilated area away from AIR, MOISTURE and HEAT.

**Occupational Health Information Resources**

The New Jersey Department of Health offers multiple services in occupational health. These services include providing informational resources, educational materials, public presentations, and industrial hygiene and medical investigations and evaluations.

**For more information, please contact:**

New Jersey Department of Health
Right to Know
PO Box 368
Trenton, NJ 08625-0368
Phone: 609-984-2202
Fax: 609-984-7407
E-mail: rtk@doh.state.nj.us
Web address: http://www.nj.gov/health/eoh/rtkweb

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GLOSSARY

ACGIH is the American Conference of Governmental Industrial Hygienists. They publish guidelines called Threshold Limit Values (TLVs) for exposure to workplace chemicals.

Acute Exposure Guideline Levels (AEGLs) are established by the EPA. They describe the risk to humans resulting from once-in-a lifetime, or rare, exposure to airborne chemicals.

Boiling point is the temperature at which a substance can change its physical state from a liquid to a gas.

A carcinogen is a substance that causes cancer.

The CAS number is unique, identifying number, assigned by the Chemical Abstracts Service, to a specific chemical.

CFR is the Code of Federal Regulations, which are the regulations of the United States government.

A combustible substance is a solid, liquid or gas that will burn.

A corrosive substance is a gas, liquid or solid that causes destruction of human skin or severe corrosion of containers.

DEP is the New Jersey Department of Environmental Protection.

DOT is the Department of Transportation, the federal agency that regulates the transportation of chemicals.

EPA is the Environmental Protection Agency, the federal agency responsible for regulating environmental hazards.

ERG is the Emergency Response Guidebook. It is a guide for emergency responders for transportation emergencies involving hazardous substances.

Emergency Response Planning Guideline (ERPG) values are intended to provide estimates of concentration ranges where one reasonably might anticipate observing adverse effects.

A fetus is an unborn human or animal.

A flammable substance is a solid, liquid, vapor or gas that will ignite easily and burn rapidly.

The flash point is the temperature at which a liquid or solid gives off vapor that can form a flammable mixture with air.

IARC is the International Agency for Research on Cancer, a scientific group.

Ionization Potential is the amount of energy needed to remove an electron from an atom or molecule. It is measured in electron volts.

IRIS is the Integrated Risk Information System database maintained by federal EPA. The database contains information on human health effects that may result from exposure to various chemicals in the environment.

LEL or Lower Explosive Limit, is the lowest concentration of a combustible substance (gas or vapor) in the air capable of continuing an explosion.

mg/m³ means milligrams of a chemical in a cubic meter of air. It is a measure of concentration (weight/volume).

A mutagen is a substance that causes mutations. A mutation is a change in the genetic material in a body cell. Mutations can lead to birth defects, miscarriages, or cancer.

NFPA is the National Fire Protection Association. It classifies substances according to their fire and explosion hazard.

NIOSH is the National Institute for Occupational Safety and Health. It tests equipment, evaluates and approves respirators, conducts studies of workplace hazards, and proposes standards to OSHA.

NTP is the National Toxicology Program which tests chemicals and reviews evidence for cancer.

OSHA is the federal Occupational Safety and Health Administration, which adopts and enforces health and safety standards.

PEOSH is the New Jersey Public Employees Occupational Safety and Health Act, which adopts and enforces health and safety standards in public workplaces.

Permeated is the movement of chemicals through protective materials.

PIH is a DOT designation for chemicals which are Poison Inhalation Hazards.

ppm means parts of a substance per million parts of air. It is a measure of concentration by volume in air.

A reactive substance is a solid, liquid or gas that releases energy under certain conditions.

STEL is a Short Term Exposure Limit which is usually a 15-minute exposure that should not be exceeded at any time during a work day.

A teratogen is a substance that causes birth defects by damaging the fetus.

UEL or Upper Explosive Limit is the highest concentration in air above which there is too much fuel (gas or vapor) to begin a reaction or explosion.

Vapor Density is the ratio of the weight of a given volume of one gas to the weight of another (usually Hydrogen), at the same temperature and pressure.

The vapor pressure is a measure of how readily a liquid or a solid mixes with air at its surface. A higher vapor pressure indicates a higher concentration of the substance in air and therefore increases the likelihood of breathing it in.
Common Name: **ALUMINUM FLUORIDE**

Synonyms: Aluminum Trifluoride  
CAS No: 7784-18-1  
Molecular Formula: AlF₃  
RTK Substance No: 0059  
Description: Odorless, white or colorless, crystalline powder

### HAZARD DATA

<table>
<thead>
<tr>
<th>Hazard Rating</th>
<th>Firefighting</th>
<th>Reactivity</th>
</tr>
</thead>
<tbody>
<tr>
<td>2 - Health</td>
<td>Extinguish fire using an agent suitable for type of surrounding fire. <strong>Aluminum Fluoride</strong> itself does not burn. POISONOUS GASES ARE PRODUCED IN FIRE, including <strong>Aluminum Oxides</strong>, <strong>Fluorine</strong> and <strong>Hydrogen Fluoride</strong> (in the presence of water). Use water spray to keep fire-exposed containers cool.</td>
<td><strong>Aluminum Fluoride</strong>, in contact with SODIUM and POTASSIUM, is sensitive to impact and a violent reaction may occur. <strong>Aluminum Fluoride</strong> will explode when heated with REDUCING AGENTS (such as LITHIUM, SODIUM, ALUMINUM and their HYDRIDES). <strong>Aluminum Fluoride</strong> is not compatible with STRONG ACIDS (such as HYDROCHLORIC, SULFURIC and NITRIC) and ACID FUMES.</td>
</tr>
<tr>
<td>0 - Fire</td>
<td></td>
<td></td>
</tr>
<tr>
<td>0 - Reactivity</td>
<td></td>
<td></td>
</tr>
</tbody>
</table>

**DOT#:** UN 1759  
**ERG Guide #:** 154  
**Hazard Class:** 8 (Corrosive)

### SPILL/LEAKS

**Isolation Distance:**  
**Spill:** 25 meters (75 feet)  
**Fire:** 800 meters (1/2 mile)  
Moisten spilled material first or use a HEPA-filter vacuum for clean-up and place into covered containers for disposal.  
Harmful to aquatic life in very low concentrations.

### PHYSICAL PROPERTIES

<table>
<thead>
<tr>
<th>Property</th>
<th>Value</th>
</tr>
</thead>
<tbody>
<tr>
<td>Odor Threshold</td>
<td>Odorless</td>
</tr>
<tr>
<td>Flash Point</td>
<td>Noncombustible</td>
</tr>
<tr>
<td>Vapor Density</td>
<td>2.9 (air = 1)</td>
</tr>
<tr>
<td>Vapor Pressure</td>
<td>1 mm Hg at 2,260°F (1,238°C)</td>
</tr>
<tr>
<td>Specific Gravity</td>
<td>3.1 (water = 1)</td>
</tr>
<tr>
<td>Water Solubility</td>
<td>Soluble</td>
</tr>
<tr>
<td>Boiling Point</td>
<td>2,799°F (1,537°C)</td>
</tr>
<tr>
<td>Melting Point</td>
<td>2,356°F (1,291°C)</td>
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<tr>
<td>Molecular Weight</td>
<td>84</td>
</tr>
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</table>

### EXPOSURE LIMITS

<table>
<thead>
<tr>
<th>Agency</th>
<th>Limit</th>
</tr>
</thead>
<tbody>
<tr>
<td>OSHA</td>
<td>2.5 mg/m³, 8-hr TWA</td>
</tr>
<tr>
<td>NIOSH</td>
<td>2.5 mg/m³, 10-hr TWA</td>
</tr>
<tr>
<td>ACGIH</td>
<td>2.5 mg/m³, 8-hr TWA</td>
</tr>
<tr>
<td>IDLH</td>
<td>250 mg/m³</td>
</tr>
</tbody>
</table>

(All of the above are for Fluorides)

### PROTECTIVE EQUIPMENT

<table>
<thead>
<tr>
<th>Equipment Type</th>
<th>Protection Details</th>
</tr>
</thead>
<tbody>
<tr>
<td>Gloves</td>
<td>Neoprene, Natural Rubber, Polyvinyl Chloride and Viton</td>
</tr>
<tr>
<td>Coveralls</td>
<td>DuPont Tyvek®</td>
</tr>
<tr>
<td>Respirator</td>
<td>&gt;2.5 mg/m³ - Full facepiece APR with High efficiency filter &gt;25 mg/m³ - Supplied air</td>
</tr>
</tbody>
</table>

### HEALTH EFFECTS

<table>
<thead>
<tr>
<th>Route</th>
<th>Effects</th>
</tr>
</thead>
<tbody>
<tr>
<td>Eyes</td>
<td>Irritation</td>
</tr>
<tr>
<td>Skin</td>
<td>Irritation</td>
</tr>
</tbody>
</table>
| Inhalation        | Nose, throat and lung irritation with coughing, wheezing and shortness of breath  
|                   | Headache, nausea and vomiting, weakness, convulsions and collapse       |

### FIRST AID AND DECONTAMINATION

Remove the person from exposure.  
Flush eyes with large amounts of water for at least 15 minutes. Remove contact lenses if worn.  
Quickly remove contaminated clothing. Immediately wash contaminated skin with large amounts of water.  
Begin artificial respiration if breathing has stopped and CPR if necessary.  
Transfer promptly to a medical facility.