

Right to Know lealth Hazardous Substance Fact Sheet

ANTIMONY Common Name:

Synonyms: Antimony Metal; Antimony Powder

Chemical Name: Antimony

Date: June 2004 Revision: February 2012

Description and Use

Antimony is a naturally occurring, silvery-white, hard, brittle metal. It is also formed as a by-product of smelting Lead and other metals. It is used in alloys with Lead and other metals, electric storage batteries, solder, sheet and pipe metal, castings and pewter.

Reasons for Citation

Antimony is on the Right to Know Hazardous Substance List because it is cited by OSHA, ACGIH, DOT, NIOSH, DEP, IRIS, and EPA.

SEE GLOSSARY ON PAGE 5.

FIRST AID

Eye Contact

Immediately flush with large amounts of water for at least 15 minutes, lifting upper and lower lids. Remove contact lenses, if worn, while rinsing.

Skin Contact

► Quickly remove contaminated clothing. Immediately wash contaminated skin with large amounts of soap and water.

Inhalation

- ▶ Remove the person from exposure.
- Begin rescue breathing (using universal precautions) if breathing has stopped and CPR if heart action has stopped.
- Transfer promptly to a medical facility.

EMERGENCY NUMBERS

Poison Control: 1-800-222-1222 CHEMTREC: 1-800-424-9300 NJDEP Hotline: 1-877-927-6337 National Response Center: 1-800-424-8802

CAS Number:	7440-36-0
RTK Substance Number:	0141
DOT Number:	UN 2871

EMERGENCY RESPONDERS >>>> SEE LAST PAGE

Hazard Summary

Hazard Rating	NJDOH	NFPA		
HEALTH	2	-		
FLAMMABILITY	2	-		
REACTIVITY	0	-		

COMBUSTIBLE POWDER

POISONOUS GASES ARE PRODUCED IN FIRE

Hazard Rating Key: 0=minimal; 1=slight; 2=moderate; 3=serious; 4=severe

- Antimony can affect you when inhaled and by passing through the skin.
- Contact can irritate the skin and eyes. Prolonged or repeated contact may cause redness and itchy skin rash (dermatitis).
- Inhaling Antimony can irritate the nose, throat and lungs causing coughing, wheezing and/or shortness of breath.
- ► Exposure to Antimony can cause headache, dizziness, nausea and vomiting, abdominal pain, and loss of sleep.
- ▶ Inhaling Antimony can cause an ulcer or hole in the "bone" (septum) dividing the inner nose.
- Repeated exposure can affect the lungs and cause an abnormal chest x-ray to develop.
- Antimony may damage the liver and kidneys and may affect the heart.

Workplace Exposure Limits

- OSHA: The legal airborne permissible exposure limit (PEL) is 0.5 mg/m³ averaged over an 8-hour workshift.
- NIOSH: The recommended airborne exposure limit (REL) is 0.5 mg/m³ averaged over a 10-hour workshift.
- ACGIH: The threshold limit value (TLV) is 0.5 mg/m^3 averaged over an 8-hour workshift.

Determining Your Exposure

- Read the product manufacturer's Material Safety Data Sheet (MSDS) and the label to determine product ingredients and important safety and health information about the product mixture.
- ► For each individual hazardous ingredient, read the New Jersey Department of Health Hazardous Substance Fact Sheet, available on the RTK website (www.nj.gov/health/eoh/rtkweb) or in your facility's RTK Central File or Hazard Communication Standard file.
- ➤ You have a right to this information under the New Jersey Worker and Community Right to Know Act and the Public Employees Occupational Safety and Health (PEOSH) Act if you are a public worker in New Jersey, and under the federal Occupational Safety and Health Act (OSHA) if you are a private worker.
- The New Jersey Right to Know Act requires most employers to label chemicals in the workplace and requires public employers to provide their employees with information concerning chemical hazards and controls. The federal OSHA Hazard Communication Standard (29 CFR 1910.1200) and the PEOSH Hazard Communication Standard (N.J.A.C. 12:100-7) require employers to provide similar information and training to their employees.

This Fact Sheet is a summary of available information regarding the health hazards that may result from exposure. Duration of exposure, concentration of the substance and other factors will affect your susceptibility to any of the potential effects described below.

Health Hazard Information

Acute Health Effects

The following acute (short-term) health effects may occur immediately or shortly after exposure to **Antimony**:

- Contact can irritate the skin and eyes. Prolonged or repeated contact may cause redness and itchy skin rash (dermatitis).
- Inhaling Antimony can irritate the nose, throat and lungs causing coughing, wheezing and/or shortness of breath.
- Exposure to Antimony can cause headache, dizziness, nausea and vomiting, abdominal pain, and loss of sleep.

Chronic Health Effects

The following chronic (long-term) health effects can occur at some time after exposure to **Antimony** and can last for months or years:

Cancer Hazard

According to the information presently available to the New Jersey Department of Health, Antimony has been tested and has not been shown to cause cancer in animals.

Reproductive Hazard

There is limited evidence that Antimony may affect female fertility.

Other Effects

- Inhaling Antimony can cause an ulcer or a hole in the "bone" (septum) dividing the inner nose, sometimes with bleeding, discharge, and/or formation of a crust.
- Repeated exposure can affect the lungs, cause an abnormal chest x-ray to develop, and lead to permanent lung damage.
- ► Antimony may damage the liver and kidneys and may affect the heart.

Medical

Medical Testing

For frequent or potentially high exposure (half the PEL or greater), the following is recommended before beginning work and at regular times after that:

► Urine test for Antimony

If symptoms develop or overexposure is suspected, the following are recommended:

- Chest x-ray and lung function tests
- Liver and kidney function tests
- ► EKG

Any evaluation should include a careful history of past and present symptoms with an exam. Medical tests that look for damage already done are <u>not</u> a substitute for controlling exposure.

You have a legal right to request copies of your medical testing under the OSHA Access to Employee Exposure and Medical Records Standard (29 CFR 1910.1020).

Mixed Exposures

- Smoking can cause heart disease, lung cancer, emphysema, and other respiratory problems. It may worsen respiratory conditions caused by chemical exposure. Even if you have smoked for a long time, stopping now will reduce your risk of developing health problems.
- More than light alcohol consumption can cause liver damage. Drinking alcohol can increase the liver damage caused by Antimony.

Workplace Controls and Practices

Very toxic chemicals, or those that are reproductive hazards or sensitizers, require expert advice on control measures if a less toxic chemical cannot be substituted. Control measures include: (1) enclosing chemical processes for severely irritating and corrosive chemicals, (2) using local exhaust ventilation for chemicals that may be harmful with a single exposure, and (3) using general ventilation to control exposures to skin and eye irritants. For further information on workplace controls, consult the NIOSH document on Control Banding at <u>www.cdc.gov/niosh/topics/ctrlbanding/</u>.

The following work practices are also recommended:

- ► Label process containers.
- Provide employees with hazard information and training.
- Monitor airborne chemical concentrations.
- Use engineering controls if concentrations exceed recommended exposure levels.
- ▶ Provide eye wash fountains and emergency showers.
- Wash or shower if skin comes in contact with a hazardous material.
- ► Always wash at the end of the workshift.
- Change into clean clothing if clothing becomes contaminated.
- ► Do not take contaminated clothing home.
- Get special training to wash contaminated clothing.
- Do not eat, smoke, or drink in areas where chemicals are being handled, processed or stored.
- ► Wash hands carefully before eating, smoking, drinking, applying cosmetics or using the toilet.

In addition, the following may be useful or required:

- Before entering a confined space where Antimony powder and dust may be present, check to make sure that an explosive concentration does not exist.
- ► Use a vacuum or a wet method to reduce dust during cleanup. DO NOT DRY SWEEP.
- ► Use a high efficiency particulate air (HEPA) filter when vacuuming. Do <u>not</u> use a standard shop vacuum.

Personal Protective Equipment

The OSHA Personal Protective Equipment Standard (29 CFR 1910.132) requires employers to determine the appropriate personal protective equipment for each hazard and to train employees on how and when to use protective equipment.

The following recommendations are only guidelines and may not apply to every situation.

Gloves and Clothing

- Avoid skin contact with Antimony. Wear personal protective equipment made from material that can not be permeated or degraded by this substance. Safety equipment suppliers and manufacturers can provide recommendations on the most protective glove and clothing material for your operation.
- ► The recommended glove materials for **Antimony** are Nitrile, Neoprene and Natural Rubber.

- The recommended protective clothing material for Antimony is Tyvek® or the equivalent.
- ► All protective clothing (suits, gloves, footwear, headgear) should be clean, available each day, and put on before work.

Eye Protection

Wear direct vent goggles when airborne particles or dust are present.

Respiratory Protection

Improper use of respirators is dangerous. Respirators should only be used if the employer has implemented a written program that takes into account workplace conditions, requirements for worker training, respirator fit testing, and medical exams, as described in the OSHA Respiratory Protection Standard (29 CFR 1910.134). **Only NIOSH approved respirators should be used.**

- Where the potential exists for exposure over 0.5 mg/m³, use a negative pressure, air-purifying, particulate filter respirator with an N, R or P95 filter. More protection is provided by a full facepiece respirator than by a half-mask respirator, and even greater protection is provided by a powered-air purifying respirator.
- ► Leave the area immediately if (1) while wearing a filter or cartridge respirator you can smell, taste, or otherwise detect **Antimony**, (2) while wearing particulate filters abnormal resistance to breathing is experienced, or (3) eye irritation occurs while wearing a full facepiece respirator. Check to make sure the respirator-to-face seal is still good. If it is, replace the filter or cartridge. If the seal is no longer good, you may need a new respirator.
- Consider all potential sources of exposure in your workplace. You may need a combination of filters, prefilters or cartridges to protect against different forms of a chemical (such as vapor and mist) or against a mixture of chemicals.
- Where the potential exists for exposure over 5 mg/m³, use a supplied-air respirator with a full facepiece operated in a pressure-demand or other positive-pressure mode. For increased protection use in combination with an auxiliary self-contained breathing apparatus or an emergency escape air cylinder.
- Exposure to 50 mg/m³ is immediately dangerous to life and health. If the possibility of exposure above 50 mg/m³ exists, use a self-contained breathing apparatus with a full facepiece operated in a pressure-demand or other positivepressure mode equipped with an emergency escape air cylinder.

Fire Hazards

If employees are expected to fight fires, they must be trained and equipped as stated in the OSHA Fire Brigades Standard (29 CFR 1910.156).

- ► Antimony is not combustible in bulk form. However, Antimony powder and dust may be COMBUSTIBLE.
- Use sand, dry chemical, CO₂, water spray or foam as extinguishing agents.
- ► DO NOT USE WATER on *molten* Antimony.
- POISONOUS GASES ARE PRODUCED IN FIRE, including Antimony Oxide and Antimony Hydride (Stibine).
- Antimony may form an ignitable dust/air mixture in closed tanks or containers.
- ► Finely dispersed Antimony powder and dust may form explosive mixtures in air.

Spills and Emergencies

If employees are required to clean-up spills, they must be properly trained and equipped. The OSHA Hazardous Waste Operations and Emergency Response Standard (29 CFR 1910.120) may apply.

If Antimony is spilled, take the following steps:

- Evacuate personnel and secure and control entrance to the area.
- Eliminate all ignition sources.
- Moisten solid spilled material first or use a HEPA-filter vacuum for clean-up and place into sealed containers for disposal.
- ► Ventilate and wash area after clean-up is complete.
- ► DO NOT wash into sewer.
- It may be necessary to contain and dispose of Antimony as a HAZARDOUS WASTE. Contact your state Department of Environmental Protection (DEP) or your regional office of the federal Environmental Protection Agency (EPA) for specific recommendations.

Handling and Storage

Prior to working with **Antimony** you should be trained on its proper handling and storage.

- Antimony reacts violently with HALOGENS (such as FLUORINE, CHLORINE and BROMINE) and STRONG ACIDS (such as HYDROCHLORIC, SULFURIC and NITRIC) to cause fires and explosions.
- Contact with STRONG ACIDS (such as HYDROCHLORIC, SULFURIC and NITRIC) and freshly formed (nascent) HYDROGEN can also form toxic Antimony Hydride (Stibine) gas.
- Antimony is not compatible with OXIDIZING AGENTS (such as PERCHLORATES, PEROXIDES, PERMANGANATES, CHLORATES, and NITRATES); STRONG BASES (such as SODIUM HYDROXIDE and POTASSIUM HYDROXIDE); IODINE; and POWDERED METALS.
- Store in tightly closed containers in a cool, well-ventilated area.
- Sources of ignition, such as smoking and open flames, are prohibited where Antimony powder is used, handled, or stored.
- ► Ground and bond containers when transferring Antimony powder.
- ► Use only non-sparking tools and equipment, especially when opening and closing containers of **Antimony** *powder*.

Occupational Health Information Resources

The New Jersey Department of Health offers multiple services in occupational health. These services include providing informational resources, educational materials, public presentations, and industrial hygiene and medical investigations and evaluations.

For more information, please contact:

New Jersey Department of Health Right to Know PO Box 368 Trenton, NJ 08625-0368 Phone: 609-984-2202 Fax: 609-984-7407 E-mail: rtk@doh.state.nj.us Web address: http://www.nj.gov/health/eoh/rtkweb

The Right to Know Hazardous Substance Fact Sheets are not intended to be copied and sold for commercial purposes.

GLOSSARY

ACGIH is the American Conference of Governmental Industrial Hygienists. They publish guidelines called Threshold Limit Values (TLVs) for exposure to workplace chemicals.

Acute Exposure Guideline Levels (AEGLs) are established by the EPA. They describe the risk to humans resulting from once-in-a lifetime, or rare, exposure to airborne chemicals.

Boiling point is the temperature at which a substance can change its physical state from a liquid to a gas.

A carcinogen is a substance that causes cancer.

The **CAS number** is unique, identifying number, assigned by the Chemical Abstracts Service, to a specific chemical.

CFR is the Code of Federal Regulations, which are the regulations of the United States government.

A combustible substance is a solid, liquid or gas that will burn.

A **corrosive** substance is a gas, liquid or solid that causes destruction of human skin or severe corrosion of containers.

The **critical temperature** is the temperature above which a gas cannot be liquefied, regardless of the pressure applied.

DEP is the New Jersey Department of Environmental Protection.

DOT is the Department of Transportation, the federal agency that regulates the transportation of chemicals.

EPA is the Environmental Protection Agency, the federal agency responsible for regulating environmental hazards.

ERG is the Emergency Response Guidebook. It is a guide for emergency responders for transportation emergencies involving hazardous substances.

Emergency Response Planning Guideline (ERPG) values provide estimates of concentration ranges where one reasonably might anticipate observing adverse effects.

A fetus is an unborn human or animal.

A **flammable** substance is a solid, liquid, vapor or gas that will ignite easily and burn rapidly.

The **flash point** is the temperature at which a liquid or solid gives off vapor that can form a flammable mixture with air.

IARC is the International Agency for Research on Cancer, a scientific group.

Ionization Potential is the amount of energy needed to remove an electron from an atom or molecule. It is measured in electron volts.

IRIS is the Integrated Risk Information System database on human health effects that may result from exposure to various chemicals, maintained by federal EPA.

LEL or **Lower Explosive Limit**, is the lowest concentration of a combustible substance (gas or vapor) in the air capable of continuing an explosion.

mg/m³ means milligrams of a chemical in a cubic meter of air. It is a measure of concentration (weight/volume).

A **mutagen** is a substance that causes mutations. A **mutation** is a change in the genetic material in a body cell. Mutations can lead to birth defects, miscarriages, or cancer.

NFPA is the National Fire Protection Association. It classifies substances according to their fire and explosion hazard.

NIOSH is the National Institute for Occupational Safety and Health. It tests equipment, evaluates and approves respirators, conducts studies of workplace hazards, and proposes standards to OSHA.

NTP is the National Toxicology Program which tests chemicals and reviews evidence for cancer.

OSHA is the federal Occupational Safety and Health Administration, which adopts and enforces health and safety standards.

PEOSHA is the New Jersey Public Employees Occupational Safety and Health Act, which adopts and enforces health and safety standards in public workplaces.

Permeated is the movement of chemicals through protective materials.

ppm means parts of a substance per million parts of air. It is a measure of concentration by volume in air.

Protective Action Criteria (PAC) are values established by the Department of Energy and are based on AEGLs and ERPGs. They are used for emergency planning of chemical release events.

A **reactive** substance is a solid, liquid or gas that releases energy under certain conditions.

STEL is a Short Term Exposure Limit which is usually a 15minute exposure that should not be exceeded at any time during a work day.

A **teratogen** is a substance that causes birth defects by damaging the fetus.

UEL or **Upper Explosive Limit** is the highest concentration in air above which there is too much fuel (gas or vapor) to begin a reaction or explosion.

Vapor Density is the ratio of the weight of a given volume of one gas to the weight of another (usually *Air*), at the same temperature and pressure.

The **vapor pressure** is a force exerted by the vapor in equilibrium with the solid or liquid phase of the same substance. The higher the vapor pressure the higher concentration of the substance in air.



Common Name: ANTIMONY

Synonyms: Antimony Metal; Antimony Powder CAS No: 7440-36-0 Molecular Formula: Sb RTK Substance No: 0141 Description: Naturally occurring, silvery-white, hard, brittle metal that is also formed from smelting *Lead* and other metals

HAZARD DATA					
Hazard Rating	Firefighting		Reactivity		
2 - Health 2 - Fire 0 - Reactivity DOT#: UN 2871 ERG Guide #: 170 Hazard Class: 6.1 (Toxic)	 Antimony is not combustible in bulk form. However, Antimony powder and dust may be COMBUSTIBLE. Use sand, dry chemical, CO₂, water spray or foam as extinguishing agents. DO NOT USE WATER on molten Antimony. POISONOUS GASES ARE PRODUCED IN FIRE, including Antimony Oxide and Antimony Hydride (Stibine). Antimony may form an ignitable dust/air mixture in closed tanks or containers. Finely dispersed Antimony powder and dust may form explosive mixtures in air. 		Antimony reacts violently with HALOGENS (such as FLUORINE, CHLORINE and BROMINE) and STRONG ACIDS (such as HYDROCHLORIC, SULFURIC and NITRIC) to cause fires and explosions. Contact with STRONG ACIDS (such as HYDROCHLORIC, SULFURIC and NITRIC) and freshly formed (nascent) HYDROGEN can also form toxic <i>Antimony Hydride</i> (<i>Stibine</i>) gas Antimony is not compatible with OXIDIZING AGENTS (such as PERCHLORATES, PEROXIDES, PERMANGANATES, CHLORATES, and NITRATES); STRONG BASES (such as SODIUM HYDROXIDE and POTASSIUM HYDROXIDE); IODINE; and POWDERED METALS.		
SP	ILL/LEAKS		PHYSICAL PROPERTIES		
vacuum for clean-up for disposal. Ground and bond con Antimony powder.	2 mile) material first or use a HEPA-filter and place into sealed containers ntainers when transferring g tools and equipment.	Flash Point: Vapor Pressur Specific Gravit Water Solubilit Boiling Point: Melting Point: Molecular Wei	ty: 6.69 (water = 1) ty: Insoluble $2,975^{\circ}\text{F} (1,635^{\circ}\text{C})$ $1,166^{\circ}\text{F} (630^{\circ}\text{C})$		
EXPOSURE LIMITS		PROTECTIVE EQUIPMENT			
OSHA: 0.5 mg/m³, 8- NIOSH: 0.5 mg/m³, 10 ACGIH: 0.5 mg/m³ IDLH: 50 mg/m³ The Protective Action C PAC-1 = 1.5 mg/m³ PAC-3 = 5)-hr TWA hr TWA riteria values are: PAC-2 = 20 mg/m ³	Gloves: Coveralls: Respirator:	Nitrile, Neoprene and Natural Rubber Tyvek Spill or >0.5 mg/m ³ : full facepiece APR with <i>P100 filters</i> Fire or >5 mg/m ³ : SCBA		
HEALTH EFFECTS		FIRS	T AID AND DECONTAMINATION		
Inhalation: Nose, t coughin breath Heada	n n, redness and itchy skin rash hroat and lung irritation, with ng, wheezing and shortness of che, dizziness, nausea, vomiting, dominal pain	Flush eyes with contact lenses Quickly remove large amounts Begin artificial	erson from exposure. In large amounts of water for at least 15 minutes. Remove e contaminated clothing and wash contaminated skin with of soap and water. respiration if breathing has stopped and CPR if necessary. btly to a medical facility.		