

Hazardous Substance Fact Sheet

CAS Number:

DOT Number:

RTK Substance Number:

Common Name: ANTIMONY POTASSIUM TARTRATE

Synonyms: Potassium Antimony Tartrate; Tartar Emetic

Chemical Name: Antimonate (2-), Bis[.mu.-[2,3-Di(Hydroxy-.kappa.O) Butanedioato(4-)-. kappa.O1:.kappa.O4]] Di-, Dipotassium, Trihydrate,

stereoisomer

Date: May 2010 Revision: September 2010

Description and Use

Antimony Potassium Tartrate is an odorless, colorless to white crystalline (sand-like) powder. It is used in textile and leather processing, and as an insecticide.

Reasons for Citation

► Antimony Potassium Tartrate is on the Right to Know Hazardous Substance List because it is cited by OSHA, ACGIH, DOT, NIOSH, DEP and EPA.

SEE GLOSSARY ON PAGE 5.

FIRST AID

Eve Contact

▶ Immediately flush with large amounts of water for at least 15 minutes, lifting upper and lower lids. Remove contact lenses, if worn, while rinsing.

Skin Contact

▶ Remove contaminated clothing and wash contaminated skin with water.

Inhalation

- ▶ Remove the person from exposure.
- ▶ Begin rescue breathing (using universal precautions) if breathing has stopped and CPR if heart action has stopped.
- ▶ Transfer promptly to a medical facility.

EMERGENCY NUMBERS

Poison Control: 1-800-222-1222 CHEMTREC: 1-800-424-9300 NJDEP Hotline: 1-877-927-6337

National Response Center: 1-800-424-8802

EMERGENCY RESPONDERS >>>> SEE LAST PAGE

28300-74-5

0145

UN 1551

Hazard Summary Hazard Rating NJDOH NFPA HEALTH 3 FLAMMABILITY 0 REACTIVITY 0 -

POISONOUS GASES ARE PRODUCED IN FIRE DOES NOT BURN

Hazard Rating Key: 0=minimal; 1=slight; 2=moderate; 3=serious; 4=severe

- ► Antimony Potassium Tartrate can affect you when inhaled and by passing through the skin.
- ► Contact can irritate and burn the skin and eyes, and can cause a skin rash.
- ► Inhaling Antimony Potassium Tartrate can irritate the nose, throat and lungs.
- ► Exposure to **Antimony Potassium Tartrate** can cause headache, dizziness, nausea and vomiting.
- ► Repeated exposure can affect the lungs and cause an abnormal chest x-ray to develop.
- ▶ Prolonged or repeated contact can cause ulcers or sores in
- ► Antimony Potassium Tartrate may damage the liver and kidneys, and affect the heart.

Workplace Exposure Limits

The following exposure limits are for *Antimony compounds* (measured as *Antimony*):

OSHA: The legal airborne permissible exposure limit (PEL) is **0.5 mg/m³** averaged over an 8-hour workshift.

NIOSH: The recommended airborne exposure limit (REL) is **0.5 mg/m³** averaged over a 10-hour workshift.

ACGIH: The threshold limit value (TLV) is **0.5 mg/m**³ averaged over an 8-hour workshift.

▶ The above exposure limits are for air levels only. When skin contact also occurs, you may be overexposed, even though air levels are less than the limits listed above.

Determining Your Exposure

- ▶ Read the product manufacturer's Material Safety Data Sheet (MSDS) and the label to determine product ingredients and important safety and health information about the product mixture.
- ► For each individual hazardous ingredient, read the New Jersey Department of Health Hazardous Substance Fact Sheet, available on the RTK website (www.nj.gov/health/eoh/rtkweb) or in your facility's RTK Central File or Hazard Communication Standard file.
- ➤ You have a right to this information under the New Jersey Worker and Community Right to Know Act and the Public Employees Occupational Safety and Health (PEOSH) Act if you are a public worker in New Jersey, and under the federal Occupational Safety and Health Act (OSHA) if you are a private worker.
- ▶ The New Jersey Right to Know Act requires most employers to label chemicals in the workplace and requires public employers to provide their employees with information concerning chemical hazards and controls. The federal OSHA Hazard Communication Standard (29 CFR 1910.1200) and the PEOSH Hazard Communication Standard (N.J.A.C. 12:100-7) require employers to provide similar information and training to their employees.

This Fact Sheet is a summary of available information regarding the health hazards that may result from exposure. Duration of exposure, concentration of the substance and other factors will affect your susceptibility to any of the potential effects described below.

Health Hazard Information

Acute Health Effects

The following acute (short-term) health effects may occur immediately or shortly after exposure to **Antimony Potassium Tartrate**:

- ► Contact can irritate and burn the skin and eyes, and can cause a skin rash.
- ▶ Inhaling Antimony Potassium Tartrate can irritate the nose, throat and lungs causing coughing, wheezing and/or shortness of breath.
- ► Exposure to **Antimony Potassium Tartrate** can cause headache, dizziness, nausea and vomiting, abdominal pain and loss of sleep.

Chronic Health Effects

The following chronic (long-term) health effects can occur at some time after exposure to **Antimony Potassium Tartrate** and can last for months or years:

Cancer Hazard

According to the information presently available to the New Jersey Department of Health, Antimony Potassium Tartrate has not been tested for its ability to cause cancer in animals.

Reproductive Hazard

► There is no evidence that **Antimony Potassium Tartrate** affects reproduction. This is based on test results presently available to the NJDHSS from published studies.

Other Effects

- ► Repeated exposure can affect the lungs and cause an abnormal chest x-ray to develop.
- ► Prolonged or repeated contact can cause ulcers or sores in the nose.
- ► Antimony Potassium Tartrate may damage the liver and kidneys, and affect the heart.

Medical

Medical Testing

For frequent or potentially high exposure (half the PEL or greater), the following is recommended before beginning work and at regular times after that:

▶ Urine test for *Antimony*

If symptoms develop or overexposure is suspected, the following are recommended:

- ► Chest x-ray and lung function tests
- ▶ EKG
- ▶ Liver and kidney function tests

Any evaluation should include a careful history of past and present symptoms with an exam. Medical tests that look for damage already done are <u>not</u> a substitute for controlling exposure.

Request copies of your medical testing. You have a legal right to this information under the OSHA Access to Employee Exposure and Medical Records Standard (29 CFR 1910.1020).

Mixed Exposures

- ► Smoking can cause heart disease, lung cancer, emphysema, and other respiratory problems. It may worsen respiratory conditions caused by chemical exposure. Even if you have smoked for a long time, stopping now will reduce your risk of developing health problems.
- ► More than light alcohol consumption can cause liver damage. Drinking alcohol may increase the liver damage caused by **Antimony Potassium Tartrate**.

Workplace Controls and Practices

Very toxic chemicals, or those that are reproductive hazards or sensitizers, require expert advice on control measures if a less toxic chemical cannot be substituted. Control measures include: (1) enclosing chemical processes for severely irritating and corrosive chemicals, (2) using local exhaust ventilation for chemicals that may be harmful with a single exposure, and (3) using general ventilation to control exposures to skin and eye irritants. For further information on workplace controls, consult the NIOSH document on Control Banding at www.cdc.gov/niosh/topics/ctrlbanding/.

The following work practices are also recommended:

- ▶ Label process containers.
- ▶ Provide employees with hazard information and training.
- ▶ Monitor airborne chemical concentrations.
- Use engineering controls if concentrations exceed recommended exposure levels.
- ▶ Provide eye wash fountains and emergency showers.
- Wash or shower if skin comes in contact with a hazardous material.
- ▶ Always wash at the end of the workshift.
- Change into clean clothing if clothing becomes contaminated.
- ▶ Do not take contaminated clothing home.
- ▶ Get special training to wash contaminated clothing.
- ▶ Do not eat, smoke, or drink in areas where chemicals are being handled, processed or stored.
- Wash hands carefully before eating, smoking, drinking, applying cosmetics or using the toilet.

In addition, the following may be useful or required:

Use a vacuum or a wet method to reduce dust during cleanup. DO NOT DRY SWEEP.

Personal Protective Equipment

The OSHA Personal Protective Equipment Standard (29 CFR 1910.132) requires employers to determine the appropriate personal protective equipment for each hazard and to train employees on how and when to use protective equipment.

The following recommendations are only guidelines and may not apply to every situation.

Gloves and Clothing

- ► Avoid skin contact with **Antimony Potassium Tartrate**. Wear personal protective equipment made from material which can not be permeated or degraded by this substance. Safety equipment suppliers and manufacturers can provide recommendations on the most protective glove and clothing material for your operation.
- Safety equipment manufacturers recommend Nitrile and Natural Rubber for gloves, and Tyvek®, or the equivalent, as a protective clothing material.
- ▶ All protective clothing (suits, gloves, footwear, headgear) should be clean, available each day, and put on before work.

Eye Protection

- ▶ Wear eye protection with side shields or goggles.
- Wear a face shield along with goggles when working with corrosive, highly irritating or toxic substances.

Respiratory Protection

Improper use of respirators is dangerous. Respirators should only be used if the employer has implemented a written program that takes into account workplace conditions, requirements for worker training, respirator fit testing, and medical exams, as described in the OSHA Respiratory Protection Standard (29 CFR 1910.134).

- For field applications check with your supervisor and your safety equipment supplier regarding the appropriate respiratory equipment.
- ▶ Where the potential exists for exposure over **0.5 mg/m³**, use a NIOSH approved negative pressure, air-purifying, particulate filter respirator with an N, R or P95 filter. More protection is provided by a full facepiece respirator than by a half-mask respirator, and even greater protection is provided by a powered-air purifying respirator.
- ▶ Leave the area immediately if (1) while wearing a filter or cartridge respirator you can smell, taste, or otherwise detect **Antimony Potassium Tartrate**, (2) while wearing particulate filters abnormal resistance to breathing is experienced, or (3) eye irritation occurs while wearing a full facepiece respirator. Check to make sure the respirator-to-face seal is still good. If it is, replace the filter or cartridge. If the seal is no longer good, you may need a new respirator.
- ▶ Consider all potential sources of exposure in your workplace. You may need a combination of filters, prefilters or cartridges to protect against different forms of a chemical (such as vapor and mist) or against a mixture of chemicals.
- Where the potential exists for exposure over 5 mg/m³, use a NIOSH approved supplied-air respirator with a full facepiece operated in a pressure-demand or other positive-pressure mode. For increased protection use in combination with an auxiliary self-contained breathing apparatus or an emergency escape air cylinder.
- ▶ Exposure to **50 mg/m³** is immediately dangerous to life and health. If the possibility of exposure above **50 mg/m³** exists, use a NIOSH approved self-contained breathing apparatus with a full facepiece operated in a pressure-demand or other positive-pressure mode equipped with an emergency escape air cylinder.

Fire Hazards

If employees are expected to fight fires, they must be trained and equipped as stated in the OSHA Fire Brigades Standard (29 CFR 1910.156).

- ► Extinguish fire using an agent suitable for type of surrounding fire. **Antimony Potassium Tartrate** itself does not burn.
- ▶ POISONOUS GASES ARE PRODUCED IN FIRE.
- ▶ Use water spray to keep fire-exposed containers cool.

ANTIMONY POTASSIUM TARTRATE

Spills and Emergencies

If employees are required to clean-up spills, they must be properly trained and equipped. The OSHA Hazardous Waste Operations and Emergency Response Standard (29 CFR 1910.120) may apply.

If **Antimony Potassium Tartrate** is spilled, take the following steps:

- ► Evacuate personnel and secure and control entrance to the area
- ▶ Eliminate all ignition sources.
- ▶ Moisten spilled material first or use a HEPA-filter vacuum for clean-up and place into sealed containers for disposal.
- ▶ Ventilate and wash area after clean-up is complete.
- ▶ DO NOT wash into sewer.
- ▶ It may be necessary to contain and dispose of **Antimony**Potassium Tartrate as a HAZARDOUS WASTE. Contact
 your state Department of Environmental Protection (DEP) or
 your regional office of the federal Environmental Protection
 Agency (EPA) for specific recommendations.

Handling and Storage

Prior to working with **Antimony Potassium Tartrate** you should be trained on its proper handling and storage.

- ▶ Antimony Potassium Tartrate is not compatible with MINERAL ACIDS (such as HYDROCHLORIC, SULFURIC and NITRIC); TANNIC ACID; PERCHLORIC ACID; ALKALI METALS (such as LITHIUM, SODIUM and POTASSIUM); CARBONATES (such as LIME WATER); LEAD; MERCURY; SILVER SALTS; and OXIDIZING AGENTS (such as PERCHLORATES, PEROXIDES, PERMANGANATES, CHLORATES, NITRATES, CHLORINE, BROMINE and FLUORINE).
- ► Antimony Potassium Tartrate can react with freshly formed HYDROGEN to form extremely flammable and poisonous *Stibine gas*.
- Store in tightly closed containers in a cool, well-ventilated area.

Occupational Health Information Resources

The New Jersey Department of Health offers multiple services in occupational health. These services include providing informational resources, educational materials, public presentations, and industrial hygiene and medical investigations and evaluations.

For more information, please contact:

New Jersey Department of Health Right to Know

PO Box 368

Trenton, NJ 08625-0368 Phone: 609-984-2202

Fax: 609-984-7407

E-mail: rtk@doh.state.nj.us

Web address: http://www.nj.gov/health/eoh/rtkweb

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ANTIMONY POTASSIUM TARTRATE

GLOSSARY

ACGIH is the American Conference of Governmental Industrial Hygienists. They publish guidelines called Threshold Limit Values (TLVs) for exposure to workplace chemicals.

Acute Exposure Guideline Levels (AEGLs) are established by the EPA. They describe the risk to humans resulting from once-in-a lifetime, or rare, exposure to airborne chemicals.

Boiling point is the temperature at which a substance can change its physical state from a liquid to a gas.

A **carcinogen** is a substance that causes cancer.

The **CAS number** is unique, identifying number, assigned by the Chemical Abstracts Service, to a specific chemical.

CFR is the Code of Federal Regulations, which are the regulations of the United States government.

A combustible substance is a solid, liquid or gas that will burn.

A **corrosive** substance is a gas, liquid or solid that causes destruction of human skin or severe corrosion of containers.

The **critical temperature** is the temperature above which a gas cannot be liquefied, regardless of the pressure applied.

DEP is the New Jersey Department of Environmental Protection.

DOT is the Department of Transportation, the federal agency that regulates the transportation of chemicals.

EPA is the Environmental Protection Agency, the federal agency responsible for regulating environmental hazards.

ERG is the Emergency Response Guidebook. It is a guide for emergency responders for transportation emergencies involving hazardous substances.

Emergency Response Planning Guideline (ERPG) values provide estimates of concentration ranges where one reasonably might anticipate observing adverse effects.

A fetus is an unborn human or animal.

A **flammable** substance is a solid, liquid, vapor or gas that will ignite easily and burn rapidly.

The **flash point** is the temperature at which a liquid or solid gives off vapor that can form a flammable mixture with air.

IARC is the International Agency for Research on Cancer, a scientific group.

Ionization Potential is the amount of energy needed to remove an electron from an atom or molecule. It is measured in electron volts.

IRIS is the Integrated Risk Information System database on human health effects that may result from exposure to various chemicals, maintained by federal EPA. **LEL** or **Lower Explosive Limit**, is the lowest concentration of a combustible substance (gas or vapor) in the air capable of continuing an explosion.

mg/m³ means milligrams of a chemical in a cubic meter of air. It is a measure of concentration (weight/volume).

A **mutagen** is a substance that causes mutations. A **mutation** is a change in the genetic material in a body cell. Mutations can lead to birth defects, miscarriages, or cancer.

NFPA is the National Fire Protection Association. It classifies substances according to their fire and explosion hazard.

NIOSH is the National Institute for Occupational Safety and Health. It tests equipment, evaluates and approves respirators, conducts studies of workplace hazards, and proposes standards to OSHA.

NTP is the National Toxicology Program which tests chemicals and reviews evidence for cancer.

OSHA is the federal Occupational Safety and Health Administration, which adopts and enforces health and safety standards.

PEOSHA is the New Jersey Public Employees Occupational Safety and Health Act, which adopts and enforces health and safety standards in public workplaces.

Permeated is the movement of chemicals through protective materials.

ppm means parts of a substance per million parts of air. It is a measure of concentration by volume in air.

Protective Action Criteria (PAC) are values established by the Department of Energy and are based on AEGLs and ERPGs. They are used for emergency planning of chemical release events.

A **reactive** substance is a solid, liquid or gas that releases energy under certain conditions.

STEL is a Short Term Exposure Limit which is usually a 15-minute exposure that should not be exceeded at any time during a work day.

A **teratogen** is a substance that causes birth defects by damaging the fetus.

UEL or **Upper Explosive Limit** is the highest concentration in air above which there is too much fuel (gas or vapor) to begin a reaction or explosion.

Vapor Density is the ratio of the weight of a given volume of one gas to the weight of another (usually *Air*), at the same temperature and pressure.

The **vapor pressure** is a force exerted by the vapor in equilibrium with the solid or liquid phase of the same substance. The higher the vapor pressure the higher concentration of the substance in air.



Right to Know Hazardous Substance Fact Sheet



Common Name: ANTIMONY POTASSIUM TARTRATE

Synonyms: Potassium Antimony Tartrate; Tartar Emetic

CAS No: 28300-74-5

Molecular Formula: C₄H₄KO₇Sb RTK Substance No: 0145

Description: Odorless, colorless to white, crystalline powder

HAZARD DATA		
Hazard Rating	Firefighting	Reactivity
3 - Health	Extinguish fire using an agent suitable for type of surrounding fire. Antimony Potassium Tartrate	Antimony Potassium Tartrate is not compatible with MINERAL ACIDS (such as HYDROCHLORIC,
0 - Fire	itself does not burn.	SULFURIC and NITRIC); TANNIC ACID, PERCHLORIC
0 - Reactivity	POISONOUS GASES ARE PRODUCED IN FIRE. Use water spray to keep fire-exposed containers cool.	ACID; ALKALI METALS (such as LITHIUM, SODIUM and POTASSIUM); CARBONATES (such as LIME WATER); LEAD; MERCURY; SILVER SALTS; and OXIDIZING AGENTS (such as PERCHLORATES, PEROXIDES, PERMANGANATES, CHLORATES, NITRATES, CHLORINE, BROMINE and FLUORINE).
DOT#: UN 1551 ERG Guide #: 151		
Hazard Class: 6.1		
(Poison)		Antimony Potassium Tartrate can react with freshly formed HYDROGEN to form extremely flammable and poisonous Stibine gas.

SPILL/LEAKS

Isolation Distance:

Spill: 25 meters (75 feet) Fire: 800 meters (1/2 mile)

Moisten spilled material first or use a HEPA-filter vacuum for clean-up and place into sealed containers for disposal.

DO NOT wash into sewer.

Antimony Potassium Tartrate is harmful to aquatic

life at very low concentrations.

PHYSICAL PROPERTIES

Odor Threshold: Odorless

Flash Point: Nonflammable **Specific Gravity:** 2.6 (water = 1)

Water Solubility: Soluble

630° to 635°F (332° to 335°C) **Melting Point:**

Molecular Weight: 324.9

EXPOSURE LIMITS

OSHA: 0.5 mg/m³, 8-hr TWA NIOSH: 0.5 mg/m³, 10-hr TWA ACGIH: 0.5 mg/m³, 8-hr TWA

IDLH: 50 ma/m³

The Protective Action Criteria values are: $PAC-1 = 4.11 \text{ mg/m}^3$ $PAC-2 = 6.86 \text{ mg/m}^3$

 $PAC-3 = 137 \text{ mg/m}^3$

PROTECTIVE EQUIPMENT

Gloves: Nitrile and Natural Rubber

Coveralls: Tyvek®

>0.5 mg/m³ - full facepiece APR with *P100 filters* Respirator:

>50 mg/m³ - SCBA

HEALTH EFFECTS

Eves: Irritation and burns Skin: Irritation, burns and rash

Inhalation: Nose, throat and lung irritation, with

coughing, wheezing and shortness of

breath

Headache, dizziness, nausea and

vomitina

FIRST AID AND DECONTAMINATION

Remove the person from exposure.

Flush eyes with large amounts of water for at least 15 minutes. Remove contact lenses if worn.

Remove contaminated clothing and wash contaminated skin with water.

Begin artificial respiration if breathing has stopped and CPR if necessary.

Transfer promptly to a medical facility.