

# Right to Know Hazardous Substance Fact Sheet

Common Name: BARIUM

Synonyms: None

Chemical Name: Barium

Date: May 2009 Revision: June 2016

# **Description and Use**

**Barium** is a silver to white, metallic powder or solid. It is found in nature as *Barite* ore and is used in oil and gas drilling, in electronic tubes, and as a stabilizer for plastics.

# **Reasons for Citation**

- ▶ Barium is on the Right to Know Hazardous Substance List because it is cited by OSHA, ACGIH, DOT, NIOSH, DEP and EPA.
- ► This chemical is on the Special Health Hazard Substance List.

SEE GLOSSARY ON PAGE 5.

# **FIRST AID**

# **Eye Contact**

▶ Immediately flush with large amounts of water for at least 15 minutes, lifting upper and lower lids. Remove contact lenses, if worn, while rinsing.

#### **Skin Contact**

Quickly remove contaminated clothing. Immediately wash contaminated skin with large amounts of soap and water.

# Inhalation

- ▶ Remove the person from exposure.
- ▶ Begin rescue breathing (using universal precautions) if breathing has stopped and CPR if heart action has stopped.
- ▶ Transfer promptly to a medical facility.

# **EMERGENCY NUMBERS**

Poison Control: 1-800-222-1222 CHEMTREC: 1-800-424-9300 NJDEP Hotline: 1-877-927-6337

National Response Center: 1-800-424-8802

CAS Number: 7440-39-3

RTK Substance Number: 0180
DOT Number: UN 1400

# **EMERGENCY RESPONDERS >>> SEE LAST PAGE**

# **Hazard Summary**

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Hazard Rating	NJDHSS	NFPA
HEALTH	2	-
FLAMMABILITY	3	-
REACTIVITY	2 <del>-W</del>	-

FLAMMABLE AND REACTIVE POISONOUS GASES ARE PRODUCED IN FIRE CONTAINERS MAY EXPLODE IN FIRE DO NOT USE WATER, CO<sub>2</sub> OR FOAM

Hazard Rating Key: 0=minimal; 1=slight; 2=moderate; 3=serious; 4=severe

- ▶ Barium can affect you when inhaled.
- ▶ Contact can irritate and burn the skin and eyes.
- ▶ Inhaling **Barium** can irritate the nose, throat and lungs.
- ▶ Very high exposure can cause **Barium** poisoning with symptoms of nausea, vomiting, diarrhea, irregular heartbeat, muscle weakness, tremors, paralysis and even death.
- ▶ Barium may damage the kidneys.
- Barium is FLAMMABLE and REACTIVE and a DANGEROUS FIRE and EXPLOSION HAZARD.
- ► Barium may ignite spontaneously in AIR or on contact with WATER.

# **Workplace Exposure Limits**

OSHA: The legal airborne permissible exposure limit (PEL) is **0.5 mg/m³** averaged over an 8-hour workshift.

NIOSH: The recommended airborne exposure limit (REL) is **0.5 mg/m³** averaged over a 10-hour workshift.

ACGIH: The threshold limit value (TLV) is **0.5 mg/m³** averaged over an 8-hour workshift.

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# **Determining Your Exposure**

▶ Read the product manufacturer's Material Safety Data Sheet (MSDS) and the label to determine product ingredients and important safety and health information about the product mixture.

For each individual hazardous ingredient, read the New Jersey Department of Health Hazardous Substance Fact Sheet, available on the RTK Program website (<a href="http://nj.gov/health/workplacehealthandsafety/right-to-know">http://nj.gov/health/workplacehealthandsafety/right-to-know</a>) or in your facility's RTK Central File or Hazard Communication Standard file.

- ➤ You have a right to this information under the New Jersey Worker and Community Right to Know Act, the Public Employees Occupational Safety and Health (PEOSH) Act if you are a public worker in New Jersey, and under the federal Occupational Safety and Health Act (OSHA) if you are a private worker.
- ▶ The New Jersey Right to Know Act requires most employers to label chemicals in the workplace and requires public employers to provide their employees with information concerning chemical hazards and controls. The federal OSHA Hazard Communication Standard (29 CFR 1910.1200) and the PEOSH Hazard Communication Standard (N.J.A.C. 12:100-7) require employers to provide similar information and training to their employees.

This Fact Sheet is a summary of available information regarding the health hazards that may result from exposure. Duration of exposure, concentration of the substance and other factors will affect your susceptibility to any of the potential effects described below.

# **Health Hazard Information**

# **Acute Health Effects**

The following acute (short-term) health effects may occur immediately or shortly after exposure to **Barium**:

- ▶ Contact can irritate and burn the skin and eyes.
- ▶ Inhaling Barium can irritate the nose, throat and lungs causing coughing, wheezing and/or shortness of breath.
- ▶ Very high exposure can cause **Barium** poisoning with symptoms of nausea, vomiting, diarrhea, irregular heartbeat, muscle weakness, tremors, paralysis and even death.

#### **Chronic Health Effects**

The following chronic (long-term) health effects can occur at some time after exposure to **Barium** and can last for months or years:

#### Cancer Hazard

► While **Barium** has been tested, it is not classifiable as to its potential to cause cancer.

# Reproductive Hazard

While Barium has been tested, further testing is required to assess its potential to cause reproductive harm.

#### Other Effects

- Barium can irritate the lungs. Repeated exposure may cause bronchitis to develop with coughing, phlegm, and/or shortness of breath.
- ▶ Barium may damage the kidneys.
- ▶ Repeated exposure may cause an abnormal chest x-ray. This usually takes years to develop.

#### Medical

#### **Medical Testing**

For frequent or potentially high exposure (half the PEL or greater), the following are recommended before beginning work and at regular times after that:

- ▶ Lung function tests
- ► Kidney function tests

If symptoms develop or overexposure is suspected, the following is recommended:

#### ▶ EKG

Any evaluation should include a careful history of past and present symptoms with an exam. Medical tests that look for damage already done are <u>not</u> a substitute for controlling exposure.

Request copies of your medical testing. You have a legal right to this information under the OSHA Access to Employee Exposure and Medical Records Standard (29 CFR 1910.1020).

# **Mixed Exposures**

Smoking can cause heart disease, lung cancer, emphysema, and other respiratory problems. It may worsen respiratory conditions caused by chemical exposure. Even if you have smoked for a long time, stopping now will reduce your risk of developing health problems. BARIUM Page 3 of 6

# **Workplace Controls and Practices**

Very toxic chemicals, or those that are reproductive hazards or sensitizers, require expert advice on control measures if a less toxic chemical cannot be substituted. Control measures include: (1) enclosing chemical processes for severely irritating and corrosive chemicals, (2) using local exhaust ventilation for chemicals that may be harmful with a single exposure, and (3) using general ventilation to control exposures to skin and eye irritants. For further information on workplace controls, consult the NIOSH document on Control Banding at www.cdc.gov/niosh/topics/ctrlbanding/.

The following work practices are also recommended:

- ▶ Label process containers.
- ▶ Provide employees with hazard information and training.
- ▶ Monitor airborne chemical concentrations.
- ► Use engineering controls if concentrations exceed recommended exposure levels.
- ▶ Provide eye wash fountains and emergency showers.
- Wash or shower if skin comes in contact with a hazardous material
- ▶ Always wash at the end of the workshift.
- Change into clean clothing if clothing becomes contaminated.
- ▶ Do not take contaminated clothing home.
- ▶ Get special training to wash contaminated clothing.
- ▶ Do not eat, smoke, or drink in areas where chemicals are being handled, processed or stored.
- Wash hands carefully before eating, smoking, drinking, applying cosmetics or using the toilet.

In addition, the following may be useful or required:

▶ Before entering a confined space where Barium may be present, check to make sure that an explosive concentration does not exist.

# **Personal Protective Equipment**

The OSHA Personal Protective Equipment Standard (29 CFR 1910.132) requires employers to determine the appropriate personal protective equipment for each hazard and to train employees on how and when to use protective equipment.

The following recommendations are only guidelines and may not apply to every situation.

# **Gloves and Clothing**

- ▶ Avoid skin contact with **Barium**. Wear personal protective equipment made from material which can not be permeated or degraded by this substance. Safety equipment suppliers and manufacturers can provide recommendations on the most protective glove and clothing material for your operation.
- Safety equipment manufacturers recommend Nitrile and Natural Rubber for gloves, and Tyvek®, or the equivalent, as a protective clothing material.
- ▶ All protective clothing (suits, gloves, footwear, headgear) should be clean, available each day, and put on before work.

### **Eye Protection**

- ▶ Wear eye protection with side shields or goggles.
- ► Wear a face shield along with goggles when working with corrosive, highly irritating or toxic substances.

#### **Respiratory Protection**

*Improper use of respirators is dangerous.* Respirators should only be used if the employer has implemented a written program that takes into account workplace conditions, requirements for worker training, respirator fit testing, and medical exams, as described in the OSHA Respiratory Protection Standard (29 CFR 1910.134).

- ► Where the potential exists for exposure over 0.5 mg/m³, use a NIOSH approved negative pressure, air-purifying, particulate filter respirator with an N, R or P95 filter. More protection is provided by a full facepiece respirator than by a half-mask respirator, and even greater protection is provided by a powered-air purifying respirator.
- ▶ Leave the area immediately if (1) while wearing a filter or cartridge respirator you can smell, taste, or otherwise detect **Barium**, (2) while wearing particulate filters abnormal resistance to breathing is experienced, or (3) eye irritation occurs while wearing a full facepiece respirator. Check to make sure the respirator-to-face seal is still good. If it is, replace the filter or cartridge. If the seal is no longer good, you may need a new respirator.
- ► Consider all potential sources of exposure in your workplace. You may need a combination of filters, prefilters or cartridges to protect against different forms of a chemical (such as vapor and mist) or against a mixture of chemicals.
- ▶ Where the potential exists for exposure over 5 mg/m³, use a NIOSH approved supplied-air respirator with a full facepiece operated in a pressure-demand or other positive-pressure mode. For increased protection use in combination with an auxiliary self-contained breathing apparatus or an emergency escape air cylinder.
- ▶ Exposure to **50 mg/m³** is immediately dangerous to life and health. If the possibility of exposure above **50 mg/m³** exists, use a NIOSH approved self-contained breathing apparatus with a full facepiece operated in a pressure-demand or other positive-pressure mode equipped with an emergency escape air cylinder.

#### Fire Hazards

If employees are expected to fight fires, they must be trained and equipped as stated in the OSHA Fire Brigades Standard (29 CFR 1910.156).

- ▶ Barium is a FLAMMABLE SOLID that may ignite spontaneously in AIR or on contact with WATER.
- ▶ Use dry chemicals appropriate for extinguishing metal fires.
- ▶ DO NOT USE WATER, CO2 or FOAM.
- ▶ POISONOUS GASES ARE PRODUCED IN FIRE.
- ► CONTAINERS MAY EXPLODE IN FIRE.

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# **Spills and Emergencies**

If employees are required to clean-up spills, they must be properly trained and equipped. The OSHA Hazardous Waste Operations and Emergency Response Standard (29 CFR 1910.120) may apply.

If Barium is spilled, take the following steps:

- Evacuate personnel and secure and control entrance to the area.
- ▶ Eliminate all ignition sources.
- ► Cover with dry sand or earth to prevent ignition and place into sealed containers for disposal.
- ▶ Ventilate area after clean-up is complete.
- ▶ DO NOT wash into sewer.
- ▶ It may be necessary to contain and dispose of **Barium** as a HAZARDOUS WASTE. Contact your state Department of Environmental Protection (DEP) or your regional office of the federal Environmental Protection Agency (EPA) for specific recommendations.

# **Handling and Storage**

Prior to working with **Barium** you should be trained on its proper handling and storage.

- ▶ Barium reacts violently with WATER and MOIST AIR to generate flammable and explosive *Hydrogen gas*.
- ► Mixtures of finely divided **Barium** and HALOGENATED HYDROCARBONS (such as TRICHLOROETHYLENE and CARBON TETRACHLORIDE) are explosive.
- ▶ Barium is not compatible with OXIDIZING AGENTS (such as PERCHLORATES, PEROXIDES, PERMANGANATES, CHLORATES, NITRATES, CHLORINE, BROMINE and FLUORINE); STRONG ACIDS (such as HYDROCHLORIC, SULFURIC and NITRIC); STRONG BASES (such as SODIUM HYDROXIDE and POTASSIUM HYDROXIDE); AMMONIA: and OXYGEN.
- Store in tightly closed containers in a cool, well-ventilated area away from AIR, HEAT and LIGHT.
- ► Sources of ignition, such as smoking and open flames, are prohibited where **Barium** is used, handled, or stored.
- ► Use explosion-proof electrical equipment and fittings wherever **Barium** is used, handled, manufactured, or stored.
- ▶ Store under inert gas, oil or Oxygen-free liquid.

# Occupational Health Information Resources

The New Jersey Department of Health, Occupational Health Service, offers multiple services in occupational health. These services include providing informational resources, educational materials, public presentations, and industrial hygiene and medical investigations and evaluations.

# For more information, please contact:

New Jersey Department of Health Right to Know Program

PO Box 368

Trenton, NJ 08625-0368 Phone: 609-984-2202 Fax: 609-984-7407 E-mail: rtk@doh.nj.gov

Web address:

http://nj.gov/health/workplacehealthandsafety/right-to-

know

The Right to Know Hazardous Substance Fact Sheets are not intended to be copied and sold for commercial purposes.

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#### **GLOSSARY**

**ACGIH** is the American Conference of Governmental Industrial Hygienists. They publish guidelines called Threshold Limit Values (TLVs) for exposure to workplace chemicals.

**Acute Exposure Guideline Levels** (AEGLs) are established by the EPA. They describe the risk to humans resulting from once-in-a lifetime, or rare, exposure to airborne chemicals.

**Boiling point** is the temperature at which a substance can change its physical state from a liquid to a gas.

A **carcinogen** is a substance that causes cancer.

The **CAS number** is unique, identifying number, assigned by the Chemical Abstracts Service, to a specific chemical.

**CFR** is the Code of Federal Regulations, which are the regulations of the United States government.

A combustible substance is a solid, liquid or gas that will burn.

A **corrosive** substance is a gas, liquid or solid that causes destruction of human skin or severe corrosion of containers.

**DEP** is the New Jersey Department of Environmental Protection.

**DOT** is the Department of Transportation, the federal agency that regulates the transportation of chemicals.

**EPA** is the Environmental Protection Agency, the federal agency responsible for regulating environmental hazards.

**ERG** is the Emergency Response Guidebook. It is a guide for emergency responders for transportation emergencies involving hazardous substances.

**Emergency Response Planning Guideline** (ERPG) values provide estimates of concentration ranges where one reasonably might anticipate observing adverse effects.

A fetus is an unborn human or animal.

A **flammable** substance is a solid, liquid, vapor or gas that will ignite easily and burn rapidly.

The **flash point** is the temperature at which a liquid or solid gives off vapor that can form a flammable mixture with air.

**IARC** is the International Agency for Research on Cancer, a scientific group.

**Ionization Potential** is the amount of energy needed to remove an electron from an atom or molecule. It is measured in electron volts.

**IRIS** is the Integrated Risk Information System database on human health effects that may result from exposure to various chemicals, maintained by federal EPA.

**LEL** or **Lower Explosive Limit**, is the lowest concentration of a combustible substance (gas or vapor) in the air capable of continuing an explosion.

mg/m<sup>3</sup> means milligrams of a chemical in a cubic meter of air. It is a measure of concentration (weight/volume).

A **mutagen** is a substance that causes mutations. A **mutation** is a change in the genetic material in a body cell. Mutations can lead to birth defects, miscarriages, or cancer.

**NFPA** is the National Fire Protection Association. It classifies substances according to their fire and explosion hazard.

**NIOSH** is the National Institute for Occupational Safety and Health. It tests equipment, evaluates and approves respirators, conducts studies of workplace hazards, and proposes standards to OSHA.

**NTP** is the National Toxicology Program which tests chemicals and reviews evidence for cancer.

**OSHA** is the federal Occupational Safety and Health Administration, which adopts and enforces health and safety standards.

**PEOSHA** is the New Jersey Public Employees Occupational Safety and Health Act, which adopts and enforces health and safety standards in public workplaces.

**Permeated** is the movement of chemicals through protective materials.

**ppm** means parts of a substance per million parts of air. It is a measure of concentration by volume in air.

**Protective Action Criteria** (PAC) are values established by the Department of Energy and are based on AEGLs and ERPGs. They are used for emergency planning of chemical release events.

A **reactive** substance is a solid, liquid or gas that releases energy under certain conditions.

**STEL** is a Short Term Exposure Limit which is usually a 15-minute exposure that should not be exceeded at any time during a work day.

A **teratogen** is a substance that causes birth defects by damaging the fetus.

**UEL** or **Upper Explosive Limit** is the highest concentration in air above which there is too much fuel (gas or vapor) to begin a reaction or explosion.

**Vapor Density** is the ratio of the weight of a given volume of one gas to the weight of another (usually *Hydrogen*), at the same temperature and pressure.

The **vapor pressure** is a force exerted by the vapor in equilibrium with the solid or liquid phase of the same substance. The higher the vapor pressure the higher concentration of the substance in air.



# **Right to Know Hazardous Substance Fact Sheet**



Common Name: BARIUM

Synonyms: None CAS No: 7440-39-3 Molecular Formula: Ba RTK Substance No: 0180

Description: Silver to white, metallic, powder or solid

HAZARD DATA			
Hazard Rating	Firefighting	Reactivity	
2 - Health	Barium is a FLAMMABLE SOLID that may ignite spontaneously in AIR or on contact with WATER.  Use dry chemicals appropriate for extinguishing metal fires.  DO NOT USE WATER, CO <sub>2</sub> or FOAM.  POISONOUS GASES ARE PRODUCED IN FIRE.  CONTAINERS MAY EXPLODE IN FIRE.	<b>Barium</b> reacts violently with WATER and MOIST AIR to generate flammable and explosive <i>Hydrogen gas</i> .	
3 - Fire 2 <del>-W</del> - Reactivity		Mixtures of finely divided Barium and HALOGENATED HYDROCARBONS (such as TRICHLOROETHYLENE and CARBON TETRACHLORIDE) are explosive.  Barium is not compatible with OXIDIZING AGENTS (such as PERCHLORATES, PEROXIDES, PERMANGANATES, CHLORATES, NITRATES, CHLORINE, BROMINE and FLUORINE); STRONG ACIDS (such as HYDROCHLORIC, SULFURIC and NITRIC); STRONG BASES (such as SODIUM HYDROXIDE and POTASSIUM HYDROXIDE); AMMONIA; and OXYGEN.	
DOT#: UN 1400 ERG Guide #: 138			
Hazard Class: 4 (Flammable Solid)			

# SPILL/LEAKS

# **Isolation Distance:**

Spill: 25 meters (75 feet) Fire: 800 meters (1/2 mile)

Cover with dry sand or earth to prevent ignition and

place into sealed containers for disposal.

DO NOT wash into sewer.

# **PHYSICAL PROPERTIES**

Flash Point: Flammable solid

**Vapor Pressure:** 10 mm Hg at 1,920°F (1,049°C)

Specific Gravity: 3.5 (water = 1)

Water Solubility: Reactive

**Boiling Point:** 2,084° to 2,979°F (1,140° to 1,637°C)

**Melting Point:** 1,310°F (710°C)

Molecular Weight: 137.34

# **EXPOSURE LIMITS**

ACGIH: 0.5 mg/m<sup>3</sup>, 8-hr TWA

**IDLH:** 50 mg/m<sup>3</sup>

The Protective Action Criteria values are:

PAC-1 = 1.5 mg/m<sup>3</sup> PAC-2 = 180 mg/m<sup>3</sup> PAC-3 = 1,100 mg/m<sup>3</sup>

# PROTECTIVE EQUIPMENT

Gloves: Nitrile and Natural Rubber

Coveralls: Tyvek®

**Respirator:** >0.5 mg/m3 - full facepiece APR with High efficiency filters

>5 mg/m<sup>3</sup> - SCBA

# **HEALTH EFFECTS**

Eyes: Irritation and burns
Skin: Irritation and burns

**Inhalation:** Nose, throat and lung irritation with

coughing, wheezing and shortness of

breath

Nausea, vomiting, irregular heartbeat, muscle weakness, tremors, paralysis and

even death

# FIRST AID AND DECONTAMINATION

**Remove** the person from exposure.

**Flush** eyes with large amounts of water for at least 15 minutes. Remove contact lenses if worn.

**Quickly** remove contaminated clothing and wash contaminated skin with large amounts of soap and water.

Begin artificial respiration if breathing has stopped and CPR if necessary.

Transfer promptly to a medical facility.