# **Null Health** Hazardous Substance Fact Sheet

### Common Name: CALCIUM CHROMATE

- Synonyms: C.I. Pigment Yellow 33; Calcium Chrome Yellow
- Chemical Name: Chromic Acid, Calcium Salt
- Date: September 1998 Revision: September 2008

### **Description and Use**

**Calcium Chromate** is an odorless yellow, crystalline (sand-like) powder. It is used to inhibit corrosion and depolarize batteries, and as a pigment.

### **Reasons for Citation**

- Calcium Chromate is on the Right to Know Hazardous Substance List because it is cited by OSHA, ACGIH, DOT, NIOSH, NTP, DEP, IARC, IRIS and EPA.
- ► This chemical is on the Special Health Hazard Substance List.

#### SEE GLOSSARY ON PAGE 5.

#### Eye Contact

- FIRST AID
- Immediately flush with large amounts of water for at least 30 minutes, lifting upper and lower lids. Remove contact lenses, if worn, while flushing. Seek medical attention.

#### **Skin Contact**

Quickly remove contaminated clothing. Immediately wash contaminated skin with large amounts of soap and water.

#### Inhalation

- ▶ Remove the person from exposure.
- Begin rescue breathing (using universal precautions) if breathing has stopped and CPR if heart action has stopped.
- Transfer promptly to a medical facility.

### **EMERGENCY NUMBERS**

Poison Control: 1-800-222-1222 CHEMTREC: 1-800-424-9300

NJDEP Hotline: 1-877-927-6337

National Response Center: 1-800-424-8802

CAS Number:	13765-19-0
RTK Substance Number:	0315
DOT Number:	UN 3077

#### **EMERGENCY RESPONDERS >>>> SEE PAGE 6**

### Hazard Summary

Hazard Rating	NJDOH	NFPA
HEALTH	4	-
FLAMMABILITY	0	-
REACTIVITY	0	-

CARCINOGEN OXIDIZER

POISONOUS GASES ARE PRODUCED IN FIRE

Hazard Rating Key: 0=minimal; 1=slight; 2=moderate; 3=serious; 4=severe

- Calcium Chromate can affect you when inhaled and may be absorbed through the skin.
- ► Calcium Chromate is a CARCINOGEN and MUTAGEN. HANDLE WITH EXTREME CAUTION.
- Contact can severely irritate and burn the skin and eyes.
- ► Inhaling Calcium Chromate can irritate the nose and throat.
- Calcium Chromate may cause a skin allergy and an asthma-like allergy.
- Inhaling Calcium Chromate can cause a sore and/or a hole in the "bone" (septum) dividing the inner nose.
- ► Calcium Chromate may damage the liver and kidneys.
- Calcium Chromate is not combustible but is a STRONG OXIDIZER which enhances the combustion of other substances.

#### Workplace Exposure Limits

- OSHA: The legal airborne permissible exposure limit (PEL) is **0.1 mg/m<sup>3</sup>** (as *Chromates*), not to be exceeded at any time.
- NIOSH: The recommended airborne exposure limit (REL) is **0.001 mg/m<sup>3</sup>** (as *Chromates*) averaged over a 10-hour workshift.
- ACGIH: The threshold limit value (TLV) is **0.001 mg/m<sup>3</sup>** averaged over an 8-hour workshift.
- Calcium Chromate is a CARCINOGEN in humans. There may be <u>no</u> safe level of exposure to a carcinogen, so all contact should be reduced to the lowest possible level.
- ► The above exposure limits are for air levels only. When skin contact also occurs, you may be overexposed, even though air levels are less than the limits listed above.

### **Determining Your Exposure**

- Read the product manufacturer's Material Safety Data Sheet (MSDS) and the label to determine product ingredients and important safety and health information about the product mixture.
- ► For each individual hazardous ingredient, read the New Jersey Department of Health Hazardous Substance Fact Sheet, available on the RTK website (www.nj.gov/health/eoh/rtkweb) or in your facility's RTK Central File or Hazard Communication Standard file.
- ➤ You have a right to this information under the New Jersey Worker and Community Right to Know Act, the Public Employees Occupational Safety and Health (PEOSH) Act if you are a public worker in New Jersey, and under the federal Occupational Safety and Health Act (OSHA) if you are a private worker.
- The New Jersey Right to Know Act requires most employers to label chemicals in the workplace and requires public employers to provide their employees with information concerning chemical hazards and controls. The federal OSHA Hazard Communication Standard (29 CFR 1910.1200) and the PEOSH Hazard Communication Standard (N.J.A.C. 12:100-7) require employers to provide similar information and training to their employees.

This Fact Sheet is a summary of available information regarding the health hazards that may result from exposure. Duration of exposure, concentration of the substance and other factors will affect your susceptibility to any of the potential effects described below.

### **Health Hazard Information**

#### **Acute Health Effects**

The following acute (short-term) health effects may occur immediately or shortly after exposure to **Calcium Chromate**:

- Contact can severely irritate and burn the skin and eyes with possible eye damage.
- Inhaling Calcium Chromate can irritate the nose and throat causing coughing and wheezing.

#### **Chronic Health Effects**

The following chronic (long-term) health effects can occur at some time after exposure to **Calcium Chromate** and can last for months or years:

#### Cancer Hazard

- Calcium Chromate is a CARCINOGEN in humans. There is evidence that *hexavalent Chromium* or *Chromium VI* compounds cause lung cancer in humans and Calcium Chromate has been shown to cause lung cancer in animals.
- Many scientists believe there is no safe level of exposure to a carcinogen. Such substance may also have the potential for causing reproductive damage in humans.

#### **Reproductive Hazard**

According to the information presently available to the New Jersey Department of Health, Calcium Chromate has not been tested for its ability to affect reproduction.

#### **Other Effects**

- ► Inhaling **Calcium Chromate** can cause a sore and/or a hole in the "bone" (septum) dividing the inner nose, sometimes with bleeding, discharge, and/or formation of a crust.
- ► Calcium Chromate may cause a skin allergy. If allergy develops, very low future exposure can cause itching and a skin rash.
- Calcium Chromate may cause an asthma-like allergy. Future exposure can cause asthma attacks with shortness of breath, wheezing, coughing, and/or chest tightness.
- Prolonged skin contact can cause burns, blisters and deep ulcers.
- ► Calcium Chromate may damage the liver and kidneys.

#### Medical

#### **Medical Testing**

Before first exposure, and every twelve (12) months thereafter, OSHA requires your employer to provide (for persons exposed to levels greater than **2.5 micrograms** of *Chromium VI* **per cubic meter** of air) and a work and medical history and exam which shall include:

- Thorough physical examination
- Lung function tests

If symptoms develop or overexposure is suspected, the following are recommended:

- Examine your skin periodically for little bumps or blisters, the first sign of "Chrome ulcers." If not treated early, these can last for years after exposure.
- Evaluation by a qualified allergist can help diagnose skin allergy.
- Liver and kidney function tests

Periodic medical surveillance, consisting of medical examinations and biological monitoring, must be provided by your employer.

OSHA requires your employer to provide you and your doctor with a copy of the OSHA *Chromium (VI)* Standard (29 CFR 1910.1026).

Any evaluation should include a careful history of past and present symptoms with an exam. Medical tests that look for damage already done are <u>not</u> a substitute for controlling exposure.

Request copies of your medical testing. You have a legal right to this information under the OSHA Access to Employee Exposure and Medical Records Standard (29 CFR 1910.1020).

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### **Mixed Exposures**

- Smoking can cause heart disease, lung cancer, emphysema, and other respiratory problems. It may worsen respiratory conditions caused by chemical exposure. Even if you have smoked for a long time, stopping now will reduce your risk of developing health problems.
- ➤ More than light alcohol consumption can cause liver damage. Drinking alcohol may increase the liver damage caused by Calcium Chromate.

### **Workplace Controls and Practices**

Very toxic chemicals, or those that are reproductive hazards or sensitizers, require expert advice on control measures if a less toxic chemical cannot be substituted. Control measures include: (1) enclosing chemical processes for severely irritating and corrosive chemicals, (2) using local exhaust ventilation for chemicals that may be harmful with a single exposure, and (3) using general ventilation to control exposures to skin and eye irritants. For further information on workplace controls, consult the NIOSH document on Control Banding at www.cdc.gov/niosh/topics/ctrlbanding/.

The following work practices are also recommended:

- ► Label process containers.
- ► Provide employees with hazard information and training.
- ► Monitor airborne chemical concentrations.
- Use engineering controls if concentrations exceed recommended exposure levels.
- ▶ Provide eye wash fountains and emergency showers.
- Wash or shower if skin comes in contact with a hazardous material.
- Always wash at the end of the workshift.
- Change into clean clothing if clothing becomes contaminated.
- ► Do not take contaminated clothing home.
- Get special training to wash contaminated clothing.
- Do not eat, smoke, or drink in areas where chemicals are being handled, processed or stored.
- Wash hands carefully before eating, smoking, drinking, applying cosmetics or using the toilet.

In addition, the following may be useful or required:

- Specific engineering controls are required for this chemical by OSHA. Refer to the OSHA Chromium VI Standard (29 CFR 1910.1026).
- Use a vacuum or a wet method to reduce dust during cleanup. DO NOT DRY SWEEP.
- ► Use a high efficiency particulate air (HEPA) filter when vacuuming. Do <u>not</u> use a standard shop vacuum.

### **Personal Protective Equipment**

The OSHA Personal Protective Equipment Standard (29 CFR 1910.132) requires employers to determine the appropriate personal protective equipment for each hazard and to train employees on how and when to use protective equipment.

The following recommendations are only guidelines and may not apply to every situation.

### **Gloves and Clothing**

- Avoid skin contact with Calcium Chromate. Wear personal protective equipment made from material that can not be permeated or degraded by this substance. Safety equipment suppliers and manufacturers can provide recommendations on the most protective glove and clothing material for your operation.
- ► Safety equipment manufacturers recommend Butyl, Nitrile, Silver Shield®/4H® and Viton for gloves and Tyvek®, or the equivalent, as a protective material for clothing.
- All protective clothing (suits, gloves, footwear, headgear) should be clean, available each day, and put on before work.

### Eye Protection

- For impact hazards (such as flying fragments, chips or particles), wear safety glasses with side shields or safety goggles.
- Wear indirect-vent, impact and splash resistant goggles when working with liquids.
- ► Wear a face shield along with goggles when working with corrosive, highly irritating or toxic substances.
- Do not wear contact lenses when working with this substance.

### **Respiratory Protection**

*Improper use of respirators is dangerous.* Respirators should only be used if the employer has implemented a written program that takes into account workplace conditions, requirements for worker training, respirator fit testing, and medical exams, as described in the OSHA Respiratory Protection Standard (29 CFR 1910.134).

- Where the potential exists for exposure over 0.001 mg/m<sup>3</sup>, use a NIOSH approved air-purifying, particulate filter respirator with an N95 filter. More protection is provided by a full facepiece respirator than by a half-mask respirator, and even greater protection is provided by a powered-air purifying respirator.
- Leave the area immediately if (1) while wearing a filter or cartridge respirator you can smell, taste, or otherwise detect Calcium Chromate, (2) while wearing particulate filters abnormal resistance to breathing is experienced, or (3) eye irritation occurs while wearing a full facepiece respirator. Check to make sure the respirator-to-face seal is still good. If it is, replace the filter or cartridge. If the seal is no longer good, you may need a new respirator.
- Consider all potential sources of exposure in your workplace. You may need a combination of filters, prefilters or cartridges to protect against different forms of a chemical (such as vapor and mist) or against a mixture of chemicals.
- ► Where the potential exists for exposure over 0.01 mg/m<sup>3</sup>, use a NIOSH approved supplied-air respirator with a full facepiece operated in a pressure-demand or other positive-pressure mode. For increased protection use in combination with an auxiliary self-contained breathing apparatus operated in a pressure-demand or other positive-pressure mode.
- ► Exposure to 15 mg/m<sup>3</sup> (as Chromates) is immediately dangerous to life and health. If the possibility of exposure above 15 mg/m<sup>3</sup> exists, use a NIOSH approved selfcontained breathing apparatus with a full facepiece operated in a pressure-demand or other positive-pressure mode equipped with an emergency escape air cylinder.

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### **Fire Hazards**

If employees are expected to fight fires, they must be trained and equipped as stated in the OSHA Fire Brigades Standard (29 CFR 1910.156).

- Extinguish fire using an agent suitable for type of surrounding fire. Calcium Chromate itself does not burn.
- Calcium Chromate is not combustible but it is a STRONG OXIDIZER which enhances the combustion of other substances.
- POISONOUS GASES ARE PRODUCED IN FIRE, including Chromium fumes.
- ► Use water spray to keep fire-exposed containers cool.
- ► Calcium Chromate may ignite combustibles (wood, paper and oil).

### **Spills and Emergencies**

If employees are required to clean-up spills, they must be properly trained and equipped. The OSHA Hazardous Waste Operations and Emergency Response Standard (29 CFR 1910.120) may apply.

If Calcium Chromate is spilled, take the following steps:

- Evacuate personnel and secure and control entrance to the area.
- ► Eliminate all ignition sources.
- Moisten spilled material first or use a HEPA-filter vacuum for clean-up and deposit into sealed containers.
- ► Ventilate and wash area after clean-up is complete.
- ▶ DO NOT wash into sewer.
- It may be necessary to contain and dispose of Calcium Chromate as a HAZARDOUS WASTE. Contact your state Department of Environmental Protection (DEP) or your regional office of the federal Environmental Protection Agency (EPA) for specific recommendations.

### Handling and Storage

Prior to working with **Calcium Chromate** you should be trained on its proper handling and storage.

- ➤ A regulated, marked area should be established where Calcium Chromate is handled, used or stored as required by the OSHA Chromium VI Standard (29 CFR 1910.1026).
- ► Calcium Chromate reacts explosively with HYDRAZINE.
- Calcium Chromate is not compatible with STRONG ACIDS (such as HYDROCHLORIC, SULFURIC and NITRIC); ORGANIC MATTER; ALUMINUM; SULFUR; BORON; and ETHANOL.
- Store in tightly closed containers in a cool, well-ventilated area away from PLASTICS and COMBUSTIBLES.

# Occupational Health Information Resources

The New Jersey Department of Health offers multiple services in occupational health. These services include providing informational resources, educational materials, public presentations, and industrial hygiene and medical investigations and evaluations.

#### For more information, please contact:

New Jersey Department of Health Right to Know PO Box 368 Trenton, NJ 08625-0368 Phone: 609-984-2202 Fax: 609-984-7407 E-mail: rtk@doh.state.nj.us Web address: http://www.nj.gov/health/eoh/rtkweb

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## **CALCIUM CHROMATE**

#### GLOSSARY

**ACGIH** is the American Conference of Governmental Industrial Hygienists. They publish guidelines called Threshold Limit Values (TLVs) for exposure to workplace chemicals.

Acute Exposure Guideline Levels (AEGLs) are established by the EPA. They describe the risk to humans resulting from once-in-a lifetime, or rare, exposure to airborne chemicals.

**Boiling point** is the temperature at which a substance can change its physical state from a liquid to a gas.

A carcinogen is a substance that causes cancer.

The **CAS number** is unique, identifying number, assigned by the Chemical Abstracts Service, to a specific chemical.

**CFR** is the Code of Federal Regulations, which are the regulations of the United States government.

A **combustible** substance is a solid, liquid or gas that will burn.

A **corrosive** substance is a gas, liquid or solid that causes destruction of human skin or severe corrosion of containers.

**DEP** is the New Jersey Department of Environmental Protection.

**DOT** is the Department of Transportation, the federal agency that regulates the transportation of chemicals.

**EPA** is the Environmental Protection Agency, the federal agency responsible for regulating environmental hazards.

**ERG** is the Emergency Response Guidebook. It is a guide for emergency responders for transportation emergencies involving hazardous substances.

**Emergency Response Planning Guideline** (ERPG) values provide estimates of concentration ranges where one reasonably might anticipate observing adverse effects.

A fetus is an unborn human or animal.

A **flammable** substance is a solid, liquid, vapor or gas that will ignite easily and burn rapidly.

The **flash point** is the temperature at which a liquid or solid gives off vapor that can form a flammable mixture with air.

**IARC** is the International Agency for Research on Cancer, a scientific group.

**Ionization Potential** is the amount of energy needed to remove an electron from an atom or molecule. It is measured in electron volts.

**IRIS** is the Integrated Risk Information System database on human health effects that may result from exposure to various chemicals, maintained by federal EPA.

**LEL** or **Lower Explosive Limit**, is the lowest concentration of a combustible substance (gas or vapor) in the air capable of continuing an explosion.

**mg/m<sup>3</sup>** means milligrams of a chemical in a cubic meter of air. It is a measure of concentration (weight/volume).

A **mutagen** is a substance that causes mutations. A **mutation** is a change in the genetic material in a body cell. Mutations can lead to birth defects, miscarriages, or cancer.

**NFPA** is the National Fire Protection Association. It classifies substances according to their fire and explosion hazard.

**NIOSH** is the National Institute for Occupational Safety and Health. It tests equipment, evaluates and approves respirators, conducts studies of workplace hazards, and proposes standards to OSHA.

**NTP** is the National Toxicology Program which tests chemicals and reviews evidence for cancer.

**OSHA** is the federal Occupational Safety and Health Administration, which adopts and enforces health and safety standards.

**PEOSHA** is the New Jersey Public Employees Occupational Safety and Health Act, which adopts and enforces health and safety standards in public workplaces.

**Permeated** is the movement of chemicals through protective materials.

**ppm** means parts of a substance per million parts of air. It is a measure of concentration by volume in air.

**Protective Action Criteria** (PAC) are values established by the Department of Energy and are based on AEGLs and ERPGs. They are used for emergency planning of chemical release events.

A **reactive** substance is a solid, liquid or gas that releases energy under certain conditions.

**STEL** is a Short Term Exposure Limit which is usually a 15minute exposure that should not be exceeded at any time during a work day.

A **teratogen** is a substance that causes birth defects by damaging the fetus.

**UEL** or **Upper Explosive Limit** is the highest concentration in air above which there is too much fuel (gas or vapor) to begin a reaction or explosion.

**Vapor Density** is the ratio of the weight of a given volume of one gas to the weight of another (usually *Hydrogen*), at the same temperature and pressure.

The **vapor pressure** is a force exerted by the vapor in equilibrium with the solid or liquid phase of the same substance. The higher the vapor pressure the higher concentration of the substance in air.

# **INFORMATION FOR EMERGENCY RESPONDERS**

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### Common Name: CALCIUM CHROMATE

Synonyms: C.I. Pigment Yellow 33; Calcium Chrome Yellow CAS No: 13765-19-0 Molecular Formula: CaCrO RTK Substance No: 0315 Description: Odorless yellow, crystalline powder

	HAZARD DATA		
Hazard Rating	Firefighting	Reactivity	
4 - Health 0 - Fire	Extinguish fire using an agent suitable for type of surrounding fire. <b>Calcium Chromate</b> itself does not burn. <b>Calcium Chromate</b> is not combustible but it is a	Calcium Chromate reacts explosively with HYDRAZINE. Calcium Chromate is not compatible with STRONG ACIDS (such as HYDROCHLORIC, SULFURIC and	
0 - Reactivity DOT#: UN 3077 ERG Guide #: 171 Hazard Class: 9 (Environmentally Hazardous Material)	STRONG OXIDIZER which enhances the combustion of other substances. POISONOUS GASES ARE PRODUCED IN FIRE, including <i>Chromium fumes</i> . Use water spray to keep fire-exposed containers cool. <b>Calcium Chromate</b> may ignite combustibles (wood, paper and oil).	NITRIC); ORGANIC MATTER; ALUMINUM; SULFUR; BORON; and ETHANOL. Store in tightly closed containers in a cool, well-ventilated area away from PLASTICS and COMBUSTIBLES.	

### SPILL/LEAKS

#### Isolation Distance:

Spill: 25 meters (75 feet) Fire: 800 meters (1/2 mile)

Moisten spilled material first or use a HEPA-filter vacuum for clean-up and deposit into sealed containers.

DO NOT wash into sewer.

### **EXPOSURE LIMITS**

- **OSHA:** 0.1 mg/m<sup>3</sup>, Ceiling (as *Chromates*)
- **NIOSH:** 0.001 mg/m<sup>3</sup>, 10-hr TWA (as *Chromates*)
- **ACGIH:** 0.001 mg/m<sup>3</sup>, 8-hr TWA
- **IDLH:** 15 mg/m<sup>3</sup> (as *Chromates*)

### **HEALTH EFFECTS**

Eyes:	Irritation and burns
Skin:	Irritation and burns, itching and ulcers
Inhalation:	Nose and throat irritation with coughing and wheezing
Chronic:	Hexavalent Chromium or Chromium VI compounds cause lung cancer in humans

# PHYSICAL PROPERTIES

Odor Threshold:	Odorless
Flash Point:	Nonflammable
Vapor Density:	2.89 (air = 1)
Vapor Pressure:	<1 mm Hg at 68°F (20°C)
Specific Gravity:	2.9 (water = 1)
Water Solubility:	Slightly soluble
Melting Point:	392°F (200°C) (dihydrate)
Molecular Weight:	156.1

	PROTECTIVE EQUIPMENT
Gloves:	Butyl, Nitrile, Silver Shield®/4H® and Viton
Coveralls:	Tyvek®
Respirator:	>0.001 mg/m <sup>3</sup> - APR with High efficiency filter >0.01 mg/m <sup>3</sup> - Supplied air

# FIRST AID AND DECONTAMINATION

**Remove** the person from exposure.

**Flush** eyes with large amounts of water for at least 30 minutes. Remove contact lenses if worn. Seek medical attention.

**Quickly** remove contaminated clothing. Immediately wash contaminated skin with large amounts of soap and water.

**Begin** artificial respiration if breathing has stopped and CPR if necessary. **Transfer** promptly to a medical facility.