Null Health Hazardous Substance Fact Sheet

Common Name: CARBONYL SULFIDE

Synonyms: Carbon Oxysulfide; Oxycarbon Sulfide Chemical Name: Carbon Oxide Sulfide (COS)

Date: September 1998 Revision: May 2009

Description and Use

Carbonyl Sulfide is a colorless gas with a distinct *Sulfur* (rotten egg) odor. It is used as a grain fumigant and to make other *Sulfur* compounds. It is also found in tobacco smoke and in emissions from coal gasification and diesel engines.

Reasons for Citation

- Carbonyl Sulfide is on the Right to Know Hazardous Substance List because it is cited by DOT, DEP, IRIS, NFPA and EPA.
- This chemical is on the Special Health Hazard Substance List.

SEE GLOSSARY ON PAGE 5.

FIRST AID

Eye Contact

Immediately flush with large amounts of water for at least 15 minutes, lifting upper and lower lids. Remove contact lenses, if worn, while flushing. Seek medical attention.

Skin Contact

- Quickly remove contaminated clothing. Immediately wash contaminated skin with large amounts of water. Seek medical attention.
- ► In case of exposure to liquefied gas, immerse affected part in warm water. Seek medical attention.

Inhalation

- ► Remove the person from exposure.
- Begin rescue breathing (using universal precautions) if breathing has stopped and CPR if heart action has stopped.
- ► Transfer promptly to a medical facility.
- Medical observation is recommended for 24 to 48 hours after overexposure, as pulmonary edema may be delayed.

EMERGENCY NUMBERS

Poison Control: 1-800-222-1222 CHEMTREC: 1-800-424-9300 NJDEP Hotline: 1-877-927-6337 National Response Center: 1-800-424-8802

CAS Number:	463-58-1
RTK Substance Number:	0349
DOT Number:	UN 2204

EMERGENCY RESPONDERS >>>> SEE LAST PAGE

Hazard Summary

Hazard Rating	NJDOH	NFPA
HEALTH	-	3
FLAMMABILITY	-	4
REACTIVITY	-	1

FLAMMABLE

POISONOUS GASES ARE PRODUCED IN FIRE CONTAINERS MAY EXPLODE IN FIRE

Hazard Rating Key: 0=minimal; 1=slight; 2=moderate; 3=serious; 4=severe

- Carbonyl Sulfide can affect you when inhaled and by passing through the skin.
- Carbonyl Sulfide can irritate the skin and may cause pain and redness. Contact with the *liquefied gas* may cause frostbite.
- ► Eye contact can cause irritation with possible eye damage.
- Contact can irritate the nose and throat.
- Inhaling Carbonyl Sulfide can irritate the lungs. Higher exposures may cause a build-up of fluid in the lungs (pulmonary edema), a medical emergency.
- Exposure can cause nausea, vomiting, weakness and muscle cramps, and may cause an irregular heartbeat (arrhythmia)
- High or repeated exposure may affect the nervous system. Higher concentrations can cause convulsions, sudden collapse and even death.
- ► Carbonyl Sulfide may affect the brain.
- Carbonyl Sulfide is a FLAMMABLE GAS and a DANGEROUS FIRE HAZARD.

Workplace Exposure Limits

No occupational exposure limits have been established for **Carbonyl Sulfide**. However, it may pose a health risk. Always follow safe work practices.

It should be recognized that Carbonyl Sulfide can be absorbed through your skin, thereby increasing your exposure.

Determining Your Exposure

- Read the product manufacturer's Material Safety Data Sheet (MSDS) and the label to determine product ingredients and important safety and health information about the product mixture.
- For each individual hazardous ingredient, read the New Jersey Department of Health Hazardous Substance Fact Sheet, available on the RTK website (www.nj.gov/health/eoh/rtkweb) or in your facility's RTK Central File or Hazard Communication Standard file.
- ➤ You have a right to this information under the New Jersey Worker and Community Right to Know Act, the Public Employees Occupational Safety and Health (PEOSH) Act if you are a public worker in New Jersey, and under the federal Occupational Safety and Health Act (OSHA) if you are a private worker.
- The New Jersey Right to Know Act requires most employers to label chemicals in the workplace and requires public employers to provide their employees with information concerning chemical hazards and controls. The federal OSHA Hazard Communication Standard (29 CFR 1910.1200) and the PEOSH Hazard Communication Standard (N.J.A.C. 12:100-7) require employers to provide similar information and training to their employees.

This Fact Sheet is a summary of available information regarding the health hazards that may result from exposure. Duration of exposure, concentration of the substance and other factors will affect your susceptibility to any of the potential effects described below.

Health Hazard Information

Acute Health Effects

The following acute (short-term) health effects may occur immediately or shortly after exposure to **Carbonyl Sulfide**:

- Carbonyl Sulfide can irritate the skin and may cause pain and redness. Contact with the *liquefied gas* may cause frostbite.
- ► Eye contact can cause irritation with possible eye damage.
- Contact can irritate the nose and throat causing coughing and wheezing.
- Inhaling Carbonyl Sulfide can irritate the lungs causing coughing and/or shortness of breath. Higher exposures may cause a build-up of fluid in the lungs (pulmonary edema), a medical emergency, with severe shortness of breath.
- Exposure can cause nausea, vomiting, weakness and muscle cramps, and may cause an irregular heartbeat (arrhythmia)

Chronic Health Effects

The following chronic (long-term) health effects can occur at some time after exposure to **Carbonyl Sulfide** and can last for months or years:

Cancer Hazard

According to the information presently available to the New Jersey Department of Health, Carbonyl Sulfide has not been tested for its ability to cause cancer in animals.

Reproductive Hazard

According to the information presently available to the New Jersey Department of Health, Carbonyl Sulfide has not been tested for its ability to affect reproduction.

Other Effects

- ➤ High or repeated exposure may affect the nervous system causing headache, dizziness, lightheadedness and passing out. Higher concentrations can cause convulsions, sudden collapse and even death.
- Carbonyl Sulfide may affect the brain causing reduced memory, inability to concentrate and personality changes.
- Carbonyl Sulfide can irritate the lungs. Repeated exposure may cause bronchitis to develop with coughing, phlegm, and/or shortness of breath.

Medical

Medical Testing

If symptoms develop or overexposure is suspected, the following are recommended:

- Chest x-ray and lung function tests
- Special 24-48 hour EKG (Holter monitor) to observe and record abnormal heart rhythms
- Exam of the nervous system
- ► EEG

Any evaluation should include a careful history of past and present symptoms with an exam. Medical tests that look for damage already done are <u>not</u> a substitute for controlling exposure.

Request copies of your medical testing. You have a legal right to this information under the OSHA Access to Employee Exposure and Medical Records Standard (29 CFR 1910.1020).

Mixed Exposures

Smoking can cause heart disease, lung cancer, emphysema, and other respiratory problems. It may worsen respiratory conditions caused by chemical exposure. Even if you have smoked for a long time, stopping now will reduce your risk of developing health problems.

Workplace Controls and Practices

Very toxic chemicals, or those that are reproductive hazards or sensitizers, require expert advice on control measures if a less toxic chemical cannot be substituted. Control measures include: (1) enclosing chemical processes for severely irritating and corrosive chemicals, (2) using local exhaust ventilation for chemicals that may be harmful with a single exposure, and (3) using general ventilation to control exposures to skin and eye irritants. For further information on workplace controls, consult the NIOSH document on Control Banding at www.cdc.gov/niosh/topics/ctrlbanding/.

The following work practices are also recommended:

- ► Label process containers.
- ▶ Provide employees with hazard information and training.
- Monitor airborne chemical concentrations.
- Use engineering controls if concentrations exceed recommended exposure levels.
- ▶ Provide eye wash fountains and emergency showers.
- Wash or shower if skin comes in contact with a hazardous material.
- Always wash at the end of the workshift.
- Change into clean clothing if clothing becomes contaminated.
- ► Do not take contaminated clothing home.
- Get special training to wash contaminated clothing.
- Do not eat, smoke, or drink in areas where chemicals are being handled, processed or stored.
- Wash hands carefully before eating, smoking, drinking, applying cosmetics or using the toilet.

In addition, the following may be useful or required:

- Specific actions are required for this chemical by OSHA. Refer to the OSHA Compressed gases Standard (29 CFR 1910.101).
- Before entering a confined space where Carbonyl Sulfide may be present, check to make sure that an explosive concentration does not exist.
- Where possible, transfer Carbonyl Sulfide from cylinders or other containers to process containers in an enclosed system.

Personal Protective Equipment

The OSHA Personal Protective Equipment Standard (29 CFR 1910.132) requires employers to determine the appropriate personal protective equipment for each hazard and to train employees on how and when to use protective equipment.

The following recommendations are only guidelines and may not apply to every situation.

Gloves and Clothing

Avoid skin contact with Carbonyl Sulfide. Wear personal protective equipment made from material which can not be permeated or degraded by this substance. Safety equipment suppliers and manufacturers can provide recommendations on the most protective glove and clothing material for your operation.

- Safety equipment manufacturers recommend Polyvinyl Alcohol, Silver Shield®/4H®, Viton and Barrier® as glove materials for Sulfur compounds, Sulfides and Disulfides, and Tychem® BR, LV, Responder® and TK; and Trellchem® HPS and VPS, or the equivalent, as protective clothing materials for Sulfur compounds, Sulfides and Disulfides.
- ► All protective clothing (suits, gloves, footwear, headgear) should be clean, available each day, and put on before work.
- Where exposure to cold equipment, vapors, or liquid may occur, employees should be provided with *insulated* gloves and special clothing designed to prevent the freezing of body tissues.

Eye Protection

- Wear non-vented, impact resistant goggles when working with fumes, gases, or vapors.
- ► Wear a face shield along with goggles when working with corrosive, highly irritating or toxic substances.

Respiratory Protection

Improper use of respirators is dangerous. Respirators should only be used if the employer has implemented a written program that takes into account workplace conditions, requirements for worker training, respirator fit testing, and medical exams, as described in the OSHA Respiratory Protection Standard (29 CFR 1910.134).

Where the potential for high exposure exists, use a NIOSH approved supplied-air respirator with a full facepiece operated in a pressure-demand or other positive-pressure mode. For increased protection use in combination with an auxiliary self-contained breathing apparatus operated in a pressure-demand or other positive-pressure mode.

Fire Hazards

If employees are expected to fight fires, they must be trained and equipped as stated in the OSHA Fire Brigades Standard (29 CFR 1910.156).

- Carbonyl Sulfide is a FLAMMABLE GAS.
- ► Stop flow of gas or let burn if leak cannot be stopped.
- POISONOUS GASES ARE PRODUCED IN FIRE, including Hydrogen Sulfide.
- ► CONTAINERS MAY EXPLODE IN FIRE.
- ► Use water spray to keep fire-exposed containers cool.
- Vapors may travel to a source of ignition and flash back.
- Vapor is heavier than air and may travel a distance to cause a fire or explosion far from the source.

CARBONYL SULFIDE

Spills and Emergencies

If employees are required to clean-up spills, they must be properly trained and equipped. The OSHA Hazardous Waste Operations and Emergency Response Standard (29 CFR 1910.120) may apply.

If Carbonyl Sulfide is leaked, take the following steps:

- Evacuate personnel and secure and control entrance to the area.
- Eliminate ignition sources.
- Ventilate area of leak to disperse the gas.
- Stop flow of gas. If source of leak is a cylinder and the leak cannot be stopped in place, remove the leaking cylinder to a safe place in the open air, and repair leak or allow cylinder to empty.
- ► Keep Carbonyl Sulfide out of confined spaces, such as sewers, because of the possibility of an explosion.
- It may be necessary to contain and dispose of Carbonyl Sulfide as a HAZARDOUS WASTE. Contact your state Department of Environmental Protection (DEP) or your regional office of the federal Environmental Protection Agency (EPA) for specific recommendations.

Handling and Storage

Prior to working with **Carbonyl Sulfide** you should be trained on its proper handling and storage.

- Carbonyl Sulfide may react with WATER or MOIST AIR to form flammable and toxic gases.
- Carbonyl Sulfide is not compatible with OXIDIZING AGENTS (such as PERCHLORATES, PEROXIDES, PERMANGANATES, CHLORATES, NITRATES, CHLORINE, BROMINE and FLUORINE).
- Store in tightly closed containers in a cool, well-ventilated area.
- Sources of ignition, such as smoking and open flames, are prohibited where Carbonyl Sulfide is used, handled, or stored.
- Metal containers involving the transfer of Carbonyl Sulfide should be grounded and bonded.
- Use explosion-proof electrical equipment and fittings wherever Carbonyl Sulfide is used, handled, manufactured, or stored.
- ► Use only non-sparking tools and equipment, especially when opening and closing containers of **Carbonyl Sulfide**.

Occupational Health Information Resources

The New Jersey Department of Health offers multiple services in occupational health. These services include providing informational resources, educational materials, public presentations, and industrial hygiene and medical investigations and evaluations.

For more information, please contact:

New Jersey Department of Health Right to Know PO Box 368 Trenton, NJ 08625-0368 Phone: 609-984-2202 Fax: 609-984-7407 E-mail: rtk@doh.state.nj.us Web address: http://www.nj.gov/health/eoh/rtkweb

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CARBONYL SULFIDE

GLOSSARY

ACGIH is the American Conference of Governmental Industrial Hygienists. They publish guidelines called Threshold Limit Values (TLVs) for exposure to workplace chemicals.

Acute Exposure Guideline Levels (AEGLs) are established by the EPA. They describe the risk to humans resulting from once-in-a lifetime, or rare, exposure to airborne chemicals.

Boiling point is the temperature at which a substance can change its physical state from a liquid to a gas.

A carcinogen is a substance that causes cancer.

The **CAS number** is unique, identifying number, assigned by the Chemical Abstracts Service, to a specific chemical.

CFR is the Code of Federal Regulations, which are the regulations of the United States government.

A combustible substance is a solid, liquid or gas that will burn.

A **corrosive** substance is a gas, liquid or solid that causes destruction of human skin or severe corrosion of containers.

DEP is the New Jersey Department of Environmental Protection.

DOT is the Department of Transportation, the federal agency that regulates the transportation of chemicals.

EPA is the Environmental Protection Agency, the federal agency responsible for regulating environmental hazards.

ERG is the Emergency Response Guidebook. It is a guide for emergency responders for transportation emergencies involving hazardous substances.

Emergency Response Planning Guideline (ERPG) values provide estimates of concentration ranges where one reasonably might anticipate observing adverse effects.

A fetus is an unborn human or animal.

A **flammable** substance is a solid, liquid, vapor or gas that will ignite easily and burn rapidly.

The **flash point** is the temperature at which a liquid or solid gives off vapor that can form a flammable mixture with air.

IARC is the International Agency for Research on Cancer, a scientific group.

Ionization Potential is the amount of energy needed to remove an electron from an atom or molecule. It is measured in electron volts.

IRIS is the Integrated Risk Information System database on human health effects that may result from exposure to various chemicals, maintained by federal EPA.

LEL or **Lower Explosive Limit**, is the lowest concentration of a combustible substance (gas or vapor) in the air capable of continuing an explosion.

mg/m³ means milligrams of a chemical in a cubic meter of air. It is a measure of concentration (weight/volume).

A **mutagen** is a substance that causes mutations. A **mutation** is a change in the genetic material in a body cell. Mutations can lead to birth defects, miscarriages, or cancer.

NFPA is the National Fire Protection Association. It classifies substances according to their fire and explosion hazard.

NIOSH is the National Institute for Occupational Safety and Health. It tests equipment, evaluates and approves respirators, conducts studies of workplace hazards, and proposes standards to OSHA.

NTP is the National Toxicology Program which tests chemicals and reviews evidence for cancer.

OSHA is the federal Occupational Safety and Health Administration, which adopts and enforces health and safety standards.

PEOSHA is the New Jersey Public Employees Occupational Safety and Health Act, which adopts and enforces health and safety standards in public workplaces.

Permeated is the movement of chemicals through protective materials.

ppm means parts of a substance per million parts of air. It is a measure of concentration by volume in air.

Protective Action Criteria (PAC) are values established by the Department of Energy and are based on AEGLs and ERPGs. They are used for emergency planning of chemical release events.

A **reactive** substance is a solid, liquid or gas that releases energy under certain conditions.

STEL is a Short Term Exposure Limit which is usually a 15minute exposure that should not be exceeded at any time during a work day.

A **teratogen** is a substance that causes birth defects by damaging the fetus.

UEL or **Upper Explosive Limit** is the highest concentration in air above which there is too much fuel (gas or vapor) to begin a reaction or explosion.

Vapor Density is the ratio of the weight of a given volume of one gas to the weight of another (usually *Hydrogen*), at the same temperature and pressure.

The **vapor pressure** is a force exerted by the vapor in equilibrium with the solid or liquid phase of the same substance. The higher the vapor pressure the higher concentration of the substance in air.



Common Name: CARBONYL SULFIDE

Synonyms: Carbon Oxysulfide; Oxycarbon Sulfide CAS No: 463-58-1 Molecular Formula: COS RTK Substance No: 0349 Description: Colorless gas with a *Sulfide* (rotten egg) odor

HAZARD DATA

Hazard Rating	Firefighting	Reactivity
3 - Health	Carbonyl Sulfide is a FLAMMABLE GAS. Stop flow of gas or let burn if leak cannot be stopped.	Carbonyl Sulfide may react with WATER or MOIST AIR to form flammable and toxic gases.
4 - Fire	POISONOUS GASES ARE PRODUCED IN FIRE,	Carbonyl Sulfide is not compatible with
1 - Reactivity	including Hydrogen Sulfide.	OXIDIZING AGENTS (such as PERCHLORATES,
DOT#: UN 2204	CONTAINERS MAY EXPLODE IN FIRE. Use water spray to keep fire-exposed containers cool.	PEROXIDES, PERMANGANATES, CHLORATES, NITRATES, CHLORINE, BROMINE and
ERG Guide #: 119	Vapors may travel to a source of ignition and flash back.	FLUORINE).
Hazard Class: 2.3	Vapor is heavier than air and may travel a distance to	
(Toxic gas)	cause a fire or explosion far from the source.	

SPILL/LEAKS

Isolation Distance:

Spill: 100 meters (330 feet)

Fire: 1,600 meters (1 mile)

If **Carbonyl Sulfide** is leaked, take the following steps: Stop flow of gas. If source of leak is a cylinder and the leak cannot be stopped in place, remove the leaking cylinder to a safe place in the open air, and repair leak or allow cylinder to empty.

Keep **Carbonyl Sulfide** out of confined spaces, such as sewers, because of the possibility of an explosion. **Carbonyl Sulfide** may bioaccumulate.

EXPOSURE LIMITS

The Protective Action Criteria values are: PAC-1 = 30 ppm PAC-2 = 55 ppm PAC-3 = 150 ppm

	HEALTH EFFECTS
Eyes:	Irritation with possible eye damage
Skin:	Irritation and redness
	Contact with the <i>liquefied gas</i> may cause frostbite
Inhalation:	Nose, throat and lung irritation with coughing, wheezing and severe shortness of breath (pulmonary edema) Nausea, vomiting, weakness and muscle cramps

PHYSICAL PROPERTIES

Odor Threshold:	Sulfide odor
Flash Point:	Flammable gas
LEL:	12%
UEL:	29%
Vapor Density:	2.1 (air = 1)
Vapor Pressure:	9,034 mm Hg at 69.8°F (21°C)
Water Solubility:	Soluble
Boiling Point:	-58°F (-50°C)
Freezing Point:	-218°F (-139°C)
Ionization Potential:	11.19 eV
Molecular Weight:	60.08

	PROTECTIVE EQUIPMENT
Gloves:	Polyvinyl Alcohol, Silver Shield®/4H®, Viton and Barrier® (>8-hr breakthrough for <i>Sulfur compounds</i> , <i>Sulfides</i> and <i>Disulfides</i>)
Coveralls:	Tychem® BR, LV, Responder® and TK; and Trellchem® HPS and VPS (>8-hr breakthrough for <i>Sulfur compounds</i> , <i>Sulfides</i> and <i>Disulfides</i>)
Respirator:	>30 ppm - SCBA

FIRST AID AND DECONTAMINATION

Remove the person from exposure.

Flush eyes with large amounts of water for at least 15 minutes. Remove contact lenses if worn. Seek medical attention.

Quickly remove contaminated clothing and wash contaminated skin with large amounts of water. Seek medical attention.

If exposed to *liquefied gas*, immerse affected part in warm water.

Begin artificial respiration if breathing has stopped and CPR if necessary. **Transfer** promptly to a medical facility.

Medical observation is recommended as symptoms may be delayed.