

Right to Know Hazardous Substance Fact Sheet

REACTIVITY

Common Name: CHLORODIFLUOROETHANE

Synonyms: Difluoro-1-Chloroethane; Freon 142 Chemical Name: Ethane, 1-Chloro-1,1-Difluoro-

Date: December 1998 Revision: February 2008

Description and Use

Chlorodifluoroethane is a colorless gas with a slightly *Ether*-like odor. It is used as a refrigerant and solvent, and as a propellant in aerosol sprays. It is an alternative for the refrigerants CFC-11 and CFC-114.

Reasons for Citation

- Chlorodifluoroethane is on the Right to Know Hazardous Substance List because it is cited by DOT, DEP, IRIS, NFPA and EPA.
- ► This chemical is on the Special Health Hazard Substance List.

SEE GLOSSARY ON PAGE 5.

FIRST AID

Eye Contact

▶ Immediately flush with large amounts of water for at least 15 minutes, lifting upper and lower lids. Remove contact lenses, if worn, while rinsing.

Skin Contact

Immerse affected part in warm water. Seek medical attention.

Inhalation

- ▶ Remove the person from exposure.
- ▶ Begin rescue breathing (using universal precautions) if breathing has stopped and CPR if heart action has stopped.
- ▶ Transfer promptly to a medical facility.

EMERGENCY NUMBERS

Poison Control: 1-800-222-1222 CHEMTREC: 1-800-424-9300 NJDEP Hotline: 1-877-927-6337

National Response Center: 1-800-424-8802

CAS Number: 75-68-3
RTK Substance Number: 0385
DOT Number: UN 2517

EMERGENCY RESPONDERS >>>> SEE BACK PAGE

Hazard Summary Hazard Rating NJDOH NFPA HEALTH - 2 FLAMMABILITY - 4

0

FLAMMABLE
ASPHYXIANT
POISONOUS GASES ARE PRO

POISONOUS GASES ARE PRODUCED IN FIRE CONTAINERS MAY EXPLODE IN FIRE

Hazard Rating Key: 0=minimal; 1=slight; 2=moderate; 3=serious; 4=severe

- ▶ Chlorodifluoroethane can affect you when inhaled.
- ► Exposure can cause headache, dizziness, lightheadedness, and passing out. Very high levels may cause death.
- ► Chlorodifluoroethane may affect the heartbeat causing irregular rhythms.
- ▶ Skin contact may cause frostbite.
- ► Chlorodifluoroethane is a FLAMMABLE GAS and a DANGEROUS FIRE HAZARD.

Workplace Exposure Limits

No occupational exposure limits have been established for **Chlorodifluoroethane**. However, it may pose a health risk. Always follow safe work practices.

Determining Your Exposure

- ▶ Read the product manufacturer's Material Safety Data Sheet (MSDS) and the label to determine product ingredients and important safety and health information about the product mixture.
- ► For each individual hazardous ingredient, read the New Jersey Department of Health Hazardous Substance Fact Sheet, available on the RTK website (www.nj.gov/health/eoh/rtkweb) or in your facility's RTK Central File or Hazard Communication Standard file.
- ➤ You have a right to this information under the New Jersey Worker and Community Right to Know Act, the Public Employees Occupational Safety and Health (PEOSH) Act if you are a public worker in New Jersey, and under the federal Occupational Safety and Health Act (OSHA) if you are a private worker.
- ► The New Jersey Right to Know Act requires most employers to label chemicals in the workplace and requires public employers to provide their employees with information concerning chemical hazards and controls. The federal OSHA Hazard Communication Standard (29 CFR 1910.1200) and the PEOSH Hazard Communication Standard (N.J.A.C. 12:100-7) require employers to provide similar information and training to their employees.

This Fact Sheet is a summary of available information regarding the health hazards that may result from exposure. Duration of exposure, concentration of the substance and other factors will affect your susceptibility to any of the potential effects described below.

Health Hazard Information

Acute Health Effects

The following acute (short-term) health effects may occur immediately or shortly after exposure to

Chlorodifluoroethane:

- Exposure can cause headache, dizziness, lightheadedness, and passing out. Very high levels may cause death.
- ► Chlorodifluoroethane may affect the heartbeat causing irregular rhythms.
- ▶ Skin contact may cause frostbite.

Chronic Health Effects

The following chronic (long-term) health effects can occur at some time after exposure to **Chlorodifluoroethane** and can last for months or years:

Cancer Hazard

According to the information presently available to the New Jersey Department of Health, Chlorodifluoroethane has not been tested for its ability to cause cancer in animals.

Reproductive Hazard

According to the information presently available to the New Jersey Department of Health, Chlorodifluoroethane has been tested and has not been shown to affect reproduction.

Other Effects

▶ No chronic (long-term) health effects are known at this time.

Medical

Medical Testing

If symptoms develop or overexposure is suspected, the following is recommended:

 Special 24 to 48 hour EKG (Halter monitor) to observe and record abnormal heart rhythms

Any evaluation should include a careful history of past and present symptoms with an exam. Medical tests that look for damage already done are <u>not</u> a substitute for controlling exposure.

Request copies of your medical testing. You have a legal right to this information under the OSHA Access to Employee Exposure and Medical Records Standard (29 CFR 1910.1020).

Workplace Controls and Practices

Very toxic chemicals, or those that are reproductive hazards or sensitizers, require expert advice on control measures if a less toxic chemical cannot be substituted. Control measures include: (1) enclosing chemical processes for severely irritating and corrosive chemicals, (2) using local exhaust ventilation for chemicals that may be harmful with a single exposure, and (3) using general ventilation to control exposures to skin and eye irritants. For further information on workplace controls, consult the NIOSH document on Control Banding at www.cdc.gov/niosh/topics/ctrlbanding/.

The following work practices are also recommended:

- ▶ Label process containers.
- ▶ Provide employees with hazard information and training.
- ▶ Monitor airborne chemical concentrations.
- Use engineering controls if concentrations exceed recommended exposure levels.
- ▶ Provide eye wash fountains and emergency showers.
- Wash or shower if skin comes in contact with a hazardous material
- ▶ Always wash at the end of the workshift.
- Change into clean clothing if clothing becomes contaminated.
- ▶ Do not take contaminated clothing home.
- ▶ Get special training to wash contaminated clothing.
- ▶ Do not eat, smoke, or drink in areas where chemicals are being handled, processed or stored.
- Wash hands carefully before eating, smoking, drinking, applying cosmetics or using the toilet.

In addition, the following may be useful or required:

- ▶ Before entering a confined space where Chlorodifluoroethane is present, check to make sure sufficient Oxygen (19.5%) exists.
- ▶ Before entering a confined space where Chlorodifluoroethane may be present, check to make sure that an explosive concentration does not exist.
- Specific engineering controls are required for this chemical by OSHA. Refer to the OSHA Compressed gases Standard (29 CFR 1910.101).

Personal Protective Equipment

The OSHA Personal Protective Equipment Standard (29 CFR 1910.132) requires employers to determine the appropriate personal protective equipment for each hazard and to train employees on how and when to use protective equipment.

The following recommendations are only guidelines and may not apply to every situation.

Gloves and Clothing

Avoid skin contact with Chlorodifluoroethane. Wear personal protective equipment made from material which can not be permeated or degraded by this substance. Safety equipment suppliers and manufacturers can provide recommendations on the most protective glove and clothing material for your operation.

- ➤ Safety equipment manufacturers recommend *insulated Neoprene* for gloves and where exposure to cold equipment, vapors, or liquid may occur, employees should be provided with special clothing designed to prevent the freezing of body tissues.
- ▶ All protective clothing (suits, gloves, footwear, headgear) should be clean, available each day, and put on before work.

Eye Protection

- ► Wear non-vented, impact resistant goggles when working with fumes, gases, or vapors.
- ▶ Wear a face shield along with goggles when working with corrosive, highly irritating or toxic substances.
- ► Do not wear contact lenses when working with this substance.

Respiratory Protection

Improper use of respirators is dangerous. Respirators should only be used if the employer has implemented a written program that takes into account workplace conditions, requirements for worker training, respirator fit testing, and medical exams, as described in the OSHA Respiratory Protection Standard (29 CFR 1910.134).

► Exposure to **Chlorodifluoroethane** is dangerous because it can replace *Oxygen* and lead to suffocation. Only NIOSH approved self-contained breathing apparatus with a full facepiece operated in the positive pressure mode should be used in *Oxygen* deficient environments.

Fire Hazards

If employees are expected to fight fires, they must be trained and equipped as stated in the OSHA Fire Brigades Standard (29 CFR 1910.156).

- ▶ Chlorodifluoroethane is a FLAMMABLE GAS.
- ▶ Stop flow of gas.
- ▶ Shut off all ignition sources.
- ▶ POISONOUS GASES ARE PRODUCED IN FIRE, including Phosgene, Hydrogen Fluoride, Hydrogen Chloride, Fluorine and Carbonyl Fluoride.
- CONTAINERS MAY EXPLODE IN FIRE. Use water spray to keep fire-exposed containers cool, but DO NOT direct water jet on liquid.
- ► FIRE MAY RESTART AFTER IT HAS BEEN EXTINGUISHED.
- ▶ Vapors may travel to a source of ignition and flash back.
- Vapor is heavier than air and may travel a distance to cause a fire or explosion far from the source.

CHLORODIFLUOROETHANE

Spills and Emergencies

If employees are required to clean-up spills, they must be properly trained and equipped. The OSHA Hazardous Waste Operations and Emergency Response Standard (29 CFR 1910.120) may apply.

If Chlorodifluoroethane is leaked, take the following steps:

- Evacuate personnel and secure and control entrance to the area.
- ▶ Ventilate area of leak to disperse the gas.
- ▶ Stop flow of gas. If source of leak is a cylinder and the leak cannot be stopped in place, remove the leaking cylinder to a safe place in the open air, and repair leak or allow cylinder to empty.
- ► Turn leaking cylinder with the leak up to prevent escape of the gas in liquid form.
- ► Keep Chlorodifluoroethane out of confined spaces, such as sewers, because of the possibility of an explosion.
- ▶ It may be necessary to contain and dispose of **Chlorodifluoroethane** as a HAZARDOUS WASTE. Contact your state Department of Environmental Protection (DEP) or your regional office of the federal Environmental Protection Agency (EPA) for specific recommendations.

Handling and Storage

Prior to working with **Chlorodifluoroethane** you should be trained on its proper handling and storage.

- ➤ Chlorodifluoroethane reacts violently with OXIDIZING AGENTS (such as PERCHLORATES, PEROXIDES, PERMANGANATES, CHLORATES, NITRATES, CHLORINE, BROMINE and FLUORINE); ALKALI METALS (such as LITHIUM, SODIUM and POTASSIUM); and ALKALINE EARTH METALS (such as BERYLLIUM, MAGNESIUM and CALCIUM).
- Sources of ignition, such as smoking and open flames, are prohibited where Chlorodifluoroethane is used, handled, or stored.
- ► Metal containers involving the transfer of **Chlorodifluoroethane** should be grounded and bonded.
- Use explosion-proof electrical equipment and fittings wherever Chlorodifluoroethane is used, handled, manufactured, or stored.
- ► Use only non-sparking tools and equipment, especially when opening and closing containers of **Chlorodifluoroethane**.
- ▶ DO NOT expose **Chlorodifluoroethane** to HEAT, FLAMES or RED HOT METAL as it will decompose to form *Hydrogen Fluoride* and *Phosgene gases*.

Occupational Health Information Resources

The New Jersey Department of Health offers multiple services in occupational health. These services include providing informational resources, educational materials, public presentations, and industrial hygiene and medical investigations and evaluations.

For more information, please contact:

New Jersey Department of Health Right to Know

PO Box 368

Trenton, NJ 08625-0368 Phone: 609-984-2202 Fax: 609-984-7407

E-mail: rtk@doh.state.nj.us

Web address: http://www.nj.gov/health/eoh/rtkweb

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CHLORODIFLUOROETHANE

GLOSSARY

ACGIH is the American Conference of Governmental Industrial Hygienists. They publish guidelines called Threshold Limit Values (TLVs) for exposure to workplace chemicals.

Acute Exposure Guideline Levels (AEGLs) are established by the EPA. They describe the risk to humans resulting from once-in-a lifetime, or rare, exposure to airborne chemicals.

Boiling point is the temperature at which a substance can change its physical state from a liquid to a gas.

A **carcinogen** is a substance that causes cancer.

The **CAS number** is unique, identifying number, assigned by the Chemical Abstracts Service, to a specific chemical.

CFR is the Code of Federal Regulations, which are the regulations of the United States government.

A combustible substance is a solid, liquid or gas that will burn.

A **corrosive** substance is a gas, liquid or solid that causes destruction of human skin or severe corrosion of containers.

DEP is the New Jersey Department of Environmental Protection.

DOT is the Department of Transportation, the federal agency that regulates the transportation of chemicals.

EPA is the Environmental Protection Agency, the federal agency responsible for regulating environmental hazards.

ERG is the Emergency Response Guidebook. It is a guide for emergency responders for transportation emergencies involving hazardous substances.

Emergency Response Planning Guideline (ERPG) values are intended to provide estimates of concentration ranges where one reasonably might anticipate observing adverse effects.

A fetus is an unborn human or animal.

A **flammable** substance is a solid, liquid, vapor or gas that will ignite easily and burn rapidly.

The **flash point** is the temperature at which a liquid or solid gives off vapor that can form a flammable mixture with air.

IARC is the International Agency for Research on Cancer, a scientific group.

Ionization Potential is the amount of energy needed to remove an electron from an atom or molecule. It is measured in electron volts.

IRIS is the Integrated Risk Information System database maintained by federal EPA. The database contains information on human health effects that may result from exposure to various chemicals in the environment.

LEL or **Lower Explosive Limit**, is the lowest concentration of a combustible substance (gas or vapor) in the air capable of continuing an explosion.

mg/m³ means milligrams of a chemical in a cubic meter of air. It is a measure of concentration (weight/volume).

A **mutagen** is a substance that causes mutations. A **mutation** is a change in the genetic material in a body cell. Mutations can lead to birth defects, miscarriages, or cancer.

NFPA is the National Fire Protection Association. It classifies substances according to their fire and explosion hazard.

NIOSH is the National Institute for Occupational Safety and Health. It tests equipment, evaluates and approves respirators, conducts studies of workplace hazards, and proposes standards to OSHA.

NTP is the National Toxicology Program which tests chemicals and reviews evidence for cancer.

OSHA is the federal Occupational Safety and Health Administration, which adopts and enforces health and safety standards.

PEOSHA is the New Jersey Public Employees Occupational Safety and Health Act, which adopts and enforces health and safety standards in public workplaces.

Permeated is the movement of chemicals through protective materials.

PIH is a DOT designation for chemicals which are Poison Inhalation Hazards.

ppm means parts of a substance per million parts of air. It is a measure of concentration by volume in air.

A **reactive** substance is a solid, liquid or gas that releases energy under certain conditions.

STEL is a Short Term Exposure Limit which is usually a 15-minute exposure that should not be exceeded at any time during a work day.

A **teratogen** is a substance that causes birth defects by damaging the fetus.

UEL or **Upper Explosive Limit** is the highest concentration in air above which there is too much fuel (gas or vapor) to begin a reaction or explosion.

Vapor Density is the ratio of the weight of a given volume of one gas to the weight of another (usually *Hydrogen*), at the same temperature and pressure.

The **vapor pressure** is a measure of how readily a liquid or a solid mixes with air at its surface. A higher vapor pressure indicates a higher concentration of the substance in air and therefore increases the likelihood of breathing it in.



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Emergency Responders Quick Reference

Common Name: CHLORODIFLUOROETHANE

Synonyms: Difluoro-1- Chloroethane; Freon 142

CAS No: 75-68-3

Molecular Formula: C₂H₃CIF₂ RTK Substance No: 0385

Description: Colorless gas with a slight Ether-like odor

HAZARD DATA		
Hazard Rating	Firefighting	Reactivity
2 - Health	FLAMMABLE GAS Shut off all ignition sources.	Chlorodifluoroethane reacts violently with OXIDIZING AGENTS (such as PERCHLORATES,
4 - Fire	POISONOUS GASES ARE PRODUCED IN FIRE,	PEROXIDES, PERMANGANATES, CHLORATES,
0 - Reactivity	including Phosgene, Hydrogen Fluoride, Hydrogen Chloride, Fluorine and Carbonyl Fluoride.	NITRATES, CHLORINE, BROMINE and FLUORINE); ALKALI METALS (such as LITHIUM,
DOT#: UN 2517	CONTAINERS MAY EXPLODE IN FIRE	SODIUM and POTASSIUM); and ALKALINE
ERG Guide #: 115	Use water spray to keep fire-exposed containers cool, but DO NOT direct water jet on liquid.	EARTH METALS (such as BERYLLIUM, MAGNESIUM and CALCIUM).
Hazard Class: 2.1	FIRE MAY RESTART AFTER IT HAS BEEN	DO NOT expose Chlorodifluoroethane to HEAT,
(Flammable gas)	EXTINGUISHED.	FLAMES or RED HOT METAL as it will
	Vapors may travel to a source of ignition and flash back.	decompose to form <i>Hydrogen Fluoride</i> and
	Vapor is heavier than air and may travel a distance to cause a fire or explosion far from the source.	Phosgene gases.

SPILL/LEAKS

Isolation Distance:

Small Spill: 120 meters (400 feet)

Large Spill or Leak: 1,000 meters (3,000 feet)

Fire: 1,600 meters (1 mile)

Turn leaking cylinder with the leak up to prevent

escape of the gas in liquid form. May cause ozone depletion.

PHYSICAL PROPERTIES

Odor Threshold: Slightly ethereal Flash Point: Flammable gas

LEL: 6% UEL: 17.9%

Auto Ignition: $1,170^{\circ}F (632^{\circ}C)$ Vapor Density: 3.5 (air = 1)

Vapor Pressure: 2,540 mm Hg at 77°F (25°C)

 Specific Gravity:
 1.1 (water = 1)

 Water Solubility:
 Insoluble

 Boiling Point:
 14.4°F (-9.8°C)

 Freezing Point:
 -203°F (-130.8°C)

Molecular Weight: 100.47

EXPOSURE LIMITS

ERPG-1: 10,000 ppm ERPG-2: 15,000 ppm ERPG-3: 25,000 ppm

PROTECTIVE EQUIPMENT

Gloves: Insulated Neoprene

Coveralls: Clothing designed to prevent freezing of body tissues

Respirator: >10,000 ppm - Supplied air

HEALTH EFFECTS

Eyes: No information Skin: Frostbite

Inhalation: Headache, dizziness, lightheadedness,

irregular heartbeat

FIRST AID AND DECONTAMINATION

Remove the person from exposure.

Flush eyes with large amounts of water for at least 15 minutes. Remove contact lenses if worn.

Immerse affected part in warm water. Seek medical attention. **Begin** artificial respiration if breathing has stopped and CPR if

necessary.

Transfer to a medical facility.