Nullealth Hazardous Substance Fact Sheet

Common Name: DIMETHYL MERCURY

Synonyms: None

Chemical Name: Mercury, Dimethyl-

Date: August 1998 Revision: November 2007

Description and Use

Dimethyl Mercury is an *Alkyl Mercury compound* that is a colorless liquid. It is used as a reagent and in making other chemicals.

Reasons for Citation

- Dimethyl Mercury is on the Right to Know Hazardous Substance List because it is cited by OSHA, ACGIH, DOT, NIOSH, DEP, IARC and EPA.
- ► This chemical is on the Special Health Hazard Substance List.

SEE GLOSSARY ON PAGE 5.

FIRST AID

Eye Contact

Immediately flush with large amounts of water for at least 15 minutes, lifting upper and lower lids. Remove contact lenses, if worn, while rinsing.

Skin Contact

Quickly remove contaminated clothing. Immediately wash contaminated skin with large amounts of soap and water.

Inhalation

- ► Remove the person from exposure.
- Begin rescue breathing (using universal precautions) if breathing has stopped and CPR if heart action has stopped.
- Transfer promptly to a medical facility.

EMERGENCY NUMBERS

Poison Control: 1-800-222-1222 CHEMTREC: 1-800-424-9300 NJDEP Hotline: 1-877-927-6337 National Response Center: 1-800-424-8802

CAS Number:	593-74-8
RTK Substance Number:	0763
DOT Number:	UN 2024

EMERGENCY RESPONDERS >>>> SEE BACK PAGE

Hazard Summary		
Hazard Rating	NJDOH	NFPA
HEALTH	3	-
FLAMMABILITY	3	-
REACTIVITY	0	-

CARCINOGEN

FLAMMABLE

POISONOUS GASES ARE PRODUCED IN FIRE CONTAINERS MAY EXPLODE IN FIRE

Hazard Rating Key: 0=minimal; 1=slight; 2=moderate; 3=serious; 4=severe

- Dimethyl Mercury can affect you when inhaled and by passing through the skin.
- Dimethyl Mercury should be handled as a CARCINOGEN--WITH EXTREME CAUTION.
- ► Contact can severely irritate and burn the skin and eyes.
- ► Inhaling **Dimethyl Mercury** can irritate the nose, throat and lungs.
- High or repeated exposure can cause Mercury poisoning with tingling or "pins and needles" feeling in fingers, irritability and weakness, slurred speech and metallic taste.
- Long term exposure can cause delayed, permanent brain damage and death.
- Dimethyl Mercury may damage the kidneys.
- Dimethyl Mercury is FLAMMABLE LIQUID and a DANGEROUS FIRE HAZARD.

Workplace Exposure Limits

The following exposure limits are for *Mercury*, *Alkyl* compounds (measured as *Mercury*):

- OSHA: The legal airborne permissible exposure limit (PEL) is **0.01 mg/m**³ averaged over an 8-hour workshift <u>and</u> **0.04 mg/m**³, not to be exceeded during any 15-minute work period.
- NIOSH: The recommended airborne exposure limit (REL) is **0.01 mg/m³** averaged over a 10-hour workshift <u>and</u> **0.03 mg/m³**, not to be exceeded during any 15-minute work period.
- ACGIH: The threshold limit value (TLV) is **0.01 mg/m³** averaged over an 8-hour workshift <u>and</u> **0.03 mg/m³** as a STEL (short-term exposure limit).
- Dimethyl Mercury may be a CARCINOGEN in humans. There may be <u>no</u> safe level of exposure to a carcinogen, so all contact should be reduced to the lowest possible level.

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► The above exposure limits are for air levels only. When skin contact also occurs, you may be overexposed, even though air levels are less than the limits listed above.

Determining Your Exposure

- Read the product manufacturer's Material Safety Data Sheet (MSDS) and the label to determine product ingredients and important safety and health information about the product mixture.
- For each individual hazardous ingredient, read the New Jersey Department of Health Hazardous Substance Fact Sheet, available on the RTK website (www.nj.gov/health/eoh/rtkweb) or in your facility's RTK Central File or Hazard Communication Standard file.
- ➤ You have a right to this information under the New Jersey Worker and Community Right to Know Act, the Public Employees Occupational Safety and Health (PEOSH) Act if you are a public worker in New Jersey, and under the federal Occupational Safety and Health Act (OSHA) if you are a private worker.
- The New Jersey Right to Know Act requires most employers to label chemicals in the workplace and requires public employers to provide their employees with information concerning chemical hazards and controls. The federal OSHA Hazard Communication Standard (29 CFR 1910.1200) and the PEOSH Hazard Communication Standard (N.J.A.C. 12:100-7) require employers to provide similar information and training to their employees.

This Fact Sheet is a summary of available information regarding the health hazards that may result from exposure. Duration of exposure, concentration of the substance and other factors will affect your susceptibility to any of the potential effects described below.

Health Hazard Information

Acute Health Effects

The following acute (short-term) health effects may occur immediately or shortly after exposure to **Dimethyl Mercury**:

- ► Contact can severely irritate and burn the skin and eyes.
- Inhaling Dimethyl Mercury can irritate the nose, throat and lungs causing coughing, wheezing and/or shortness of breath.

Chronic Health Effects

The following chronic (long-term) health effects can occur at some time after exposure to **Dimethyl Mercury** and can last for months or years:

Cancer Hazard

- Dimethyl Mercury may be a CARCINOGEN in humans since it has been shown to cause kidney cancer in animals.
- Many scientists believe there is no safe level of exposure to a carcinogen.

Reproductive Hazard

While Dimethyl Mercury has not been tested for its ability to affect reproduction, it should be HANDLED WITH CAUTION since several related Mercury compounds are known teratogens in humans.

Other Effects

- High or repeated exposure can cause Mercury poisoning. Symptoms include sore gums, tingling or "pins and needles" feeling in fingers, lips and tongue, clumsiness, irritability and weakness, slurred speech and metallic taste.
- Long term exposure can cause delayed, permanent brain damage and death with little or no warning.
- Dimethyl Mercury may damage the kidneys.
- Mercury accumulates in the body with repeated exposures. It can take months or years for the body to get rid of excess Mercury.

Medical

Medical Testing

For frequent or potentially high exposure (half the PEL or greater, or significant skin contact) the following are recommended before beginning work and at regular times after that:

- ► Exam of the nervous system, including handwriting
- ► Urine *Mercury* level (usually less than 0.02 mg/liter)
- Kidney function test

If symptoms develop or overexposure is suspected, the following are recommended:

 Consider neurobehavioral, nerve conduction and urinary enzyme testing

Any evaluation should include a careful history of past and present symptoms with an exam. Medical tests that look for damage already done are <u>not</u> a substitute for controlling exposure.

Request copies of your medical testing. You have a legal right to this information under the OSHA Access to Employee Exposure and Medical Records Standard (29 CFR 1910.1020).

Mixed Exposures

Creams to whiten or bleach skin may contain *Mercury*. If you use them, you may be at increased risk of *Mercury* poisoning. A high fish diet, especially of marine predatory fish (fish-eating fish), also may increase your blood *Mercury* levels.

Workplace Controls and Practices

Very toxic chemicals, or those that are reproductive hazards or sensitizers, require expert advice on control measures if a less toxic chemical cannot be substituted. Control measures include: (1) enclosing chemical processes for severely irritating and corrosive chemicals, (2) using local exhaust ventilation for chemicals that may be harmful with a single exposure, and (3) using general ventilation to control exposures to skin and eye irritants. For further information on workplace controls, consult the NIOSH document on Control Banding at <u>www.cdc.gov/niosh/topics/ctrlbanding/</u>.

The following work practices are also recommended:

- ► Label process containers.
- Provide employees with hazard information and training.
- Monitor airborne chemical concentrations.
- Use engineering controls if concentrations exceed recommended exposure levels.
- ▶ Provide eye wash fountains and emergency showers.
- Wash or shower if skin comes in contact with a hazardous material.
- Always wash at the end of the workshift.
- Change into clean clothing if clothing becomes contaminated.
- ► Do not take contaminated clothing home.
- Get special training to wash contaminated clothing.
- Do not eat, smoke, or drink in areas where chemicals are being handled, processed or stored.
- ➤ Wash hands carefully before eating, smoking, drinking, applying cosmetics or using the toilet.
- ➤ For clean-up, use a specialized charcoal-filtered vacuum or suction pump to avoid generating *Mercury vapor*. Do not disturb spilled material.

In addition, the following may be useful or required:

Before entering a confined space where Dimethyl Mercury may be present, check to make sure that an explosive concentration does not exist.

Personal Protective Equipment

The OSHA Personal Protective Equipment Standard (29 CFR 1910.132) requires employers to determine the appropriate personal protective equipment for each hazard and to train employees on how and when to use protective equipment.

The following recommendations are only guidelines and may not apply to every situation.

Gloves and Clothing

- ➤ Avoid skin contact with Dimethyl Mercury. Wear personal protective equipment made from material which can not be permeated or degraded by this substance. Safety equipment suppliers and manufacturers can provide recommendations on the most protective glove and clothing material for your operation.
- ► Safety equipment manufacturers recommend 4-H/Silver Shield® for gloves and DuPont Tychem® Responder®, CSM and TK for toxic and corrosive chemical vapors and heavy liquids.

► All protective clothing (suits, gloves, footwear, headgear) should be clean, available each day, and put on before work.

Eye Protection

- ► Wear non-vented, impact resistant goggles.
- Wear a face shield along with goggles when working with corrosive, highly irritating or toxic substances.
- Do not wear contact lenses when working with this substance.

Respiratory Protection

Improper use of respirators is dangerous. Respirators should only be used if the employer has implemented a written program that takes into account workplace conditions, requirements for worker training, respirator fit testing, and medical exams, as described in the OSHA Respiratory Protection Standard (29 CFR 1910.134).

- ► Where the potential exists for exposure over 0.01 mg/m³ (as Mercury), use a NIOSH approved supplied-air respirator with a full facepiece operated in a pressure-demand or other positive-pressure mode. For increased protection use in combination with an auxiliary self-contained breathing apparatus operated in a pressure-demand or other positive-pressure mode.
- Exposure to 2 mg/m³ (as Mercury) is immediately dangerous to life and health. If the possibility of exposure above 2 mg/m³ exists, use a NIOSH approved self-contained breathing apparatus with a full facepiece operated in a pressure-demand or other positive-pressure mode equipped with an emergency escape air cylinder.

Fire Hazards

If employees are expected to fight fires, they must be trained and equipped as stated in the OSHA Fire Brigades Standard (29 CFR 1910.156).

- ► Dimethyl Mercury is a FLAMMABLE LIQUID.
- ► Use dry chemical, CO₂, water spray or foam as extinguishing agents.
- POISONOUS GASES ARE PRODUCED IN FIRE, including Mercury vapors.
- ► CONTÁINERS MAY EXPLODE IN FIRE.
- ► Use water spray to keep fire-exposed containers cool.
- ► Vapor is heavier than air and may travel a distance to cause a fire or explosion far from the source.

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Spills and Emergencies

If employees are required to clean-up spills, they must be properly trained and equipped. The OSHA Hazardous Waste Operations and Emergency Response Standard (29 CFR 1910.120) may apply.

If **Dimethyl Mercury** is spilled or leaked, take the following steps:

- Evacuate personnel and secure and control entrance to the area.
- Eliminate all ignition sources.
- ► For clean up use a special charcoal-filtered vacuum or suction pump to avoid generating *Mercury Vapor*. Do not disturb spilled material.
- Absorb liquids in vermiculite, dry sand, earth, or similar inert material and deposit in sealed containers.
- ► Ventilate area after clean-up is complete.
- ► Keep **Dimethyl Mercury** out of confined spaces, such as sewers, because of the possibility of an explosion.
- It may be necessary to contain and dispose of Dimethyl Mercury as a HAZARDOUS WASTE. Contact your state Department of Environmental Protection (DEP) or your regional office of the federal Environmental Protection Agency (EPA) for specific recommendations.

Handling and Storage

Prior to working with **Dimethyl Mercury** you should be trained on its proper handling and storage.

- Dimethyl Mercury reacts with OXIDIZING AGENTS (such as PERCHLORATES, PEROXIDES, PERMANGANATES, CHLORATES, NITRATES, CHLORINE, BROMINE and FLUORINE) to cause a fire hazard.
- Store in tightly closed containers in a cool, well-ventilated area.
- Sources of ignition, such as smoking and open flames, are prohibited where **Dimethyl Mercury** is used, handled, or stored.
- Metal containers involving the transfer of Dimethyl Mercury should be grounded and bonded.
- Use explosion-proof electrical equipment and fittings wherever **Dimethyl Mercury** is used, handled, manufactured, or stored.
- ► Use only non-sparking tools and equipment, especially when opening and closing containers of **Dimethyl Mercury**.

Occupational Health Information Resources

The New Jersey Department of Health offers multiple services in occupational health. These services include providing informational resources, educational materials, public presentations, and industrial hygiene and medical investigations and evaluations.

For more information, please contact:

New Jersey Department of Health Right to Know PO Box 368 Trenton, NJ 08625-0368 Phone: 609-984-2202 Fax: 609-984-7407 E-mail: rtk@doh.state.nj.us Web address: http://www.nj.gov/health/eoh/rtkweb

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GLOSSARY

ACGIH is the American Conference of Governmental Industrial Hygienists. They publish guidelines called Threshold Limit Values (TLVs) for exposure to workplace chemicals.

Acute Exposure Guideline Levels (AEGLs) are established by the EPA. They describe the risk to humans resulting from once-in-a lifetime, or rare, exposure to airborne chemicals.

Boiling point is the temperature at which a substance can change its physical state from a liquid to a gas.

A carcinogen is a substance that causes cancer.

The **CAS number** is unique, identifying number, assigned by the Chemical Abstracts Service, to a specific chemical.

CFR is the Code of Federal Regulations, which are the regulations of the United States government.

A combustible substance is a solid, liquid or gas that will burn.

A **corrosive** substance is a gas, liquid or solid that causes destruction of human skin or severe corrosion of containers.

DEP is the New Jersey Department of Environmental Protection.

DOT is the Department of Transportation, the federal agency that regulates the transportation of chemicals.

EPA is the Environmental Protection Agency, the federal agency responsible for regulating environmental hazards.

ERG is the Emergency Response Guidebook. It is a guide for emergency responders for transportation emergencies involving hazardous substances.

Emergency Response Planning Guideline (ERPG) values are intended to provide estimates of concentration ranges where one reasonably might anticipate observing adverse effects.

A fetus is an unborn human or animal.

A **flammable** substance is a solid, liquid, vapor or gas that will ignite easily and burn rapidly.

The **flash point** is the temperature at which a liquid or solid gives off vapor that can form a flammable mixture with air.

IARC is the International Agency for Research on Cancer, a scientific group.

Ionization Potential is the amount of energy needed to remove an electron from an atom or molecule. It is measured in electron volts.

IRIS is the Integrated Risk Information System database maintained by federal EPA. The database contains information on human health effects that may result from exposure to various chemicals in the environment.

LEL or **Lower Explosive Limit**, is the lowest concentration of a combustible substance (gas or vapor) in the air capable of continuing an explosion.

mg/m³ means milligrams of a chemical in a cubic meter of air. It is a measure of concentration (weight/volume).

A **mutagen** is a substance that causes mutations. A **mutation** is a change in the genetic material in a body cell. Mutations can lead to birth defects, miscarriages, or cancer.

NFPA is the National Fire Protection Association. It classifies substances according to their fire and explosion hazard.

NIOSH is the National Institute for Occupational Safety and Health. It tests equipment, evaluates and approves respirators, conducts studies of workplace hazards, and proposes standards to OSHA.

NTP is the National Toxicology Program which tests chemicals and reviews evidence for cancer.

OSHA is the federal Occupational Safety and Health Administration, which adopts and enforces health and safety standards.

PEOSHA is the New Jersey Public Employees Occupational Safety and Health Act, which adopts and enforces health and safety standards in public workplaces.

Permeated is the movement of chemicals through protective materials.

PIH is a DOT designation for chemicals which are Poison Inhalation Hazards.

ppm means parts of a substance per million parts of air. It is a measure of concentration by volume in air.

A **reactive** substance is a solid, liquid or gas that releases energy under certain conditions.

STEL is a Short Term Exposure Limit which is usually a 15minute exposure that should not be exceeded at any time during a work day.

A **teratogen** is a substance that causes birth defects by damaging the fetus.

UEL or **Upper Explosive Limit** is the highest concentration in air above which there is too much fuel (gas or vapor) to begin a reaction or explosion.

Vapor Density is the ratio of the weight of a given volume of one gas to the weight of another (usually *Hydrogen*), at the same temperature and pressure.

The **vapor pressure** is a measure of how readily a liquid or a solid mixes with air at its surface. A higher vapor pressure indicates a higher concentration of the substance in air and therefore increases the likelihood of breathing it in.



Common Name: DIMETHYL MERCURY

Synonyms: None CAS No: 593-74-8 Molecular Formula: C₂H₆Hg RTK Substance No: 0763 Description: Colorless liquid

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Hazard Rating	Firefighting	Reactivity
3 - Health 3 - Fire	Dimethyl Mercury is a FLAMMABLE LIQUID. Use dry chemical, CO ₂ , water spray or foam as extinguishing agents.	Dimethyl Mercury reacts with OXIDIZING AGENTS (such as PERCHLORATES, PEROXIDES, PERMANGANATES, CHLORATES, NITRATES, CHLORINE, BROMINE and FLUORINE) to cause a fire
0 - Reactivity DOT#: UN 2024 ERG Guide #: 151	POISONOUS GASES ARE PRODUCED IN FIRE, including <i>Mercury vapors</i> . CONTAINERS MAY EXPLODE IN FIRE.	hazard.
Hazard Class: 6.1 (Poison)	Use water spray to keep fire-exposed containers cool. Vapor is heavier than air and may travel a distance to cause a fire or explosion far from the source.	
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SPILL/LEAKS

Isolation Distance:

Small Spills: 60 meters (200 feet)

Large Spills: 270 meters (900 feet)

Absorb liquids in vermiculite, dry sand, earth, or a similar material and deposit in sealed containers.

DO NOT wash into sewer.

DO NOT let this substance enter the environment as it bioaccumulates.

EXPOSURE LIMITS

	0.01 mg/m ³ , 8-hr TWA; 0.04 mg/m ³ , STEL
NIOSH:	0.01 mg/m ³ , 10-hr TWA; 0.03 mg/m ³ , STEL
ACGIH:	0.01 mg/m ³ , 8-hr TWA; 0.03 mg/m ³ , STEL
IDLH:	2 mg/m ³
	(All the above are as Maroury)

(All the above are as Mercury)

HEALTH EFFECTS

Eyes:	Irritation and burns
Skin:	Irritation and burns
Acute:	Irritation of the nose, throat and lungs with coughing, wheezing and/or shortness of breath
Chronic:	Carcinogen (kidney) in animals
	Several Methyl Mercury compounds are known teratogens
	Sore gums, tingling or "pins and needles" feeling in fingers, irritability and weakness, slurred speech and metallic taste

PHYSICAL PROPERTIES Odor Threshold: No information

Gloves: Coveralls:		/er Shield® (60-minutes breakthrough) : Tychem® Responder®, CSM and TK for tox
	PRO	
Molecular We	eight:	230.7
Melting Point		-45.4°F (-43°C)
Boiling Point		204°F (96°C)
Water Solubi	lity:	Insoluble
Specific Grav	vity:	3 (water = 1)
Vapor Pressu	ıre:	50 mm Hg at 68°F (20°C)
Vapor Densit	y:	7.9 (air = 1)
Flash Point:		41°F (5°C)
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Coveralls:	DuPont Tychem® Responder®, CSM and TK for toxic and corrosive chemical vapors
Boots:	No information
Respirator:	>0.01 mg/m ³ - Supplied air

FIRST AID AND DECONTAMINATION

Remove the person from exposure.

- **Flush** eyes with large amounts of water for at least 15 minutes. Remove contact lenses if worn.
- **Remove** contaminated clothing and wash contaminated skin with large amounts of soap and water.

Begin artificial respiration if breathing has stopped and CPR if necessary.

Transfer to a medical facility.