

# Right to Know Hazardous Substance Fact Sheet

Common Name: EPICHLOROHYDRIN

Synonyms: 3-Chloropropylene Oxide; 1-Chloro-2,3-Epoxypropane

Chemical Name: Oxirane, (Chloromethyl)-

Date: January 2011 Revision: July 2016

# **Description and Use**

**Epichlorohydrin** is a clear, colorless liquid with an irritating odor. It is used in making epoxy resins and as an insect fumigant, solvent, and stabilizer.

#### ▶ ODOR THRESHOLD = 0.08 to 12 ppm

► Odor thresholds vary greatly. Do not rely on odor alone to determine potentially hazardous exposures.

# **Reasons for Citation**

- ► Epichlorohydrin is on the Right to Know Hazardous Substance List because it is cited by OSHA, ACGIH, DOT, NIOSH, NTP, DEP, IARC, NFPA and EPA.
- ► This chemical is on the Special Health Hazard Substance List.

SEE GLOSSARY ON PAGE 5.

# **FIRST AID**

# **Eye Contact**

▶ Immediately flush with large amounts of water for at least 30 minutes, lifting upper and lower lids. Remove contact lenses, if worn, while flushing. Seek medical attention.

#### **Skin Contact**

► Quickly remove contaminated clothing. Immediately wash contaminated skin with large amounts of water.

#### Inhalation

- ► Remove the person from exposure.
- ▶ Begin rescue breathing (using universal precautions) if breathing has stopped and CPR if heart action has stopped.
- ► Transfer promptly to a medical facility.
- ► Medical observation is recommended for 24 to 48 hours after overexposure, as pulmonary edema may be delayed.

# **EMERGENCY NUMBERS**

Poison Control: 1-800-222-1222 CHEMTREC: 1-800-424-9300 NJDEP Hotline: 1-877-927-6337

National Response Center: 1-800-424-8802

CAS Number: 106-89-8

RTK Substance Number: 0828

DOT Number: UN 2023

#### **EMERGENCY RESPONDERS >>>> SEE LAST PAGE**

Hazard Summary		
Hazard Rating	NJDHSS	NFPA
HEALTH	-	4
FLAMMABILITY	-	3
REACTIVITY	-	2

**CARCINOGEN** 

FLAMMABLE AND REACTIVE POISONOUS GASES ARE PRODUCED IN FIRE CONTAINERS MAY EXPLODE IN FIRE

Hazard Rating Key: 0=minimal; 1=slight; 2=moderate; 3=serious; 4=severe

- ► Epichlorohydrin can affect you when inhaled and by passing through the skin.
- ► Epichlorohydrin is a CARCINOGEN and MUTAGEN. HANDLE WITH EXTREME CAUTION.
- ► Contact can irritate and burn the skin and eyes.
- ▶ Exposure can irritate the eyes, nose and throat.
- ▶ Inhaling **Epichlorohydrin** can irritate the lungs. Higher exposures may cause a build-up of fluid in the lungs (pulmonary edema), a medical emergency.
- ► Epichlorohydrin may cause a skin allergy and may cause an asthma-like allergy.
- ▶ Epichlorohydrin may damage the liver and kidneys.
- ▶ High exposure can affect the nervous system.
- ► Epichlorohydrin is FLAMMABLE and REACTIVE and a DANGEROUS FIRE and EXPLOSION HAZARD.
- ► Epichlorohydrin can polymerize violently when exposed to HEAT.

#### **Workplace Exposure Limits**

OSHA: The legal airborne permissible exposure limit (PEL) is **5 ppm** averaged over an 8-hour workshift.

NIOSH: Recommends that exposure to occupational carcinogens be limited to the lowest feasible concentration.

ACGIH: The threshold limit value (TLV) is **0.5 ppm** averaged over an 8-hour workshift.

- ▶ Epichlorohydrin is a PROBABLE CARCINOGEN in humans. There may be no safe level of exposure to a carcinogen, so all contact should be reduced to the lowest possible level.
- ► The above exposure limits are for air levels only. When skin contact also occurs, you may be overexposed, even though air levels are less than the limits listed above.

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# **Determining Your Exposure**

- ▶ Read the product manufacturer's Material Safety Data Sheet (MSDS) and the label to determine product ingredients and important safety and health information about the product mixture.
- ► For each individual hazardous ingredient, read the New Jersey Department of Health and Senior Services Hazardous Substance Fact Sheet, available on the RTK Program website
  - (http://www.state.nj.us/health/workplacehealthandsafety/right-to-know/) or in your facility's RTK Central File or Hazard Communication Standard file.
- ▶ You have a right to this information under the New Jersey Worker and Community Right to Know Act and the Public Employees Occupational Safety and Health (PEOSH) Act if you are a public worker in New Jersey, and under the federal Occupational Safety and Health Act (OSHA) if you are a private worker.
- ▶ The New Jersey Right to Know Act requires most employers to label chemicals in the workplace and requires public employers to provide their employees with information concerning chemical hazards and controls. The federal OSHA Hazard Communication Standard (29 CFR 1910.1200) and the PEOSH Hazard Communication Standard (N.J.A.C. 12:100-7) require employers to provide similar information and training to their employees.

This Fact Sheet is a summary of available information regarding the health hazards that may result from exposure. Duration of exposure, concentration of the substance and other factors will affect your susceptibility to any of the potential effects described below.

# **Health Hazard Information**

# **Acute Health Effects**

The following acute (short-term) health effects may occur immediately or shortly after exposure to **Epichlorohydrin**:

- ▶ Contact can irritate and burn the skin and eyes.
- ▶ Exposure can irritate the eyes, nose and throat.
- ▶ Inhaling **Epichlorohydrin** can irritate the lungs causing coughing and/or shortness of breath. Higher exposures may cause a build-up of fluid in the lungs (pulmonary edema), a medical emergency, with severe shortness of breath.

#### **Chronic Health Effects**

The following chronic (long-term) health effects can occur at some time after exposure to **Epichlorohydrin** and can last for months or years:

#### Cancer Hazard

- ▶ Epichlorohydrin is a PROBABLE CARCINOGEN in humans. There is evidence that it causes lung cancer in humans and it has been shown to cause nasal cavity and skin cancer in animals.
- Many scientists believe there is no safe level of exposure to a carcinogen.

#### Reproductive Hazard

▶ Epichlorohydrin may decrease fertility in males.

#### Other Effects

- ► Epichlorohydrin may cause a skin allergy. If allergy develops, very low future exposure can cause itching and a skin rash.
- ► Repeated high exposure may cause an asthma-like allergy. Future exposure can cause asthma attacks with shortness of breath, wheezing, coughing, and/or chest tightness.
- ▶ Epichlorohydrin may damage the liver and kidneys.
- ▶ High exposure can affect the nervous system.

#### **Medical**

# **Medical Testing**

For frequent or potentially high exposure (half the TLV or greater), the following are recommended before beginning work and at regular times after that:

- ▶ Liver function tests
- ▶ Lung function tests. The results may be normal if the person is not having an attack at the time of the test.

If symptoms develop or overexposure is suspected, the following are recommended:

- ▶ Consider chest x-ray after acute overexposure
- ► Kidney function tests
- ▶ Exam of the nervous system
- ► Evaluation by a qualified allergist can help diagnose skin allergy.

Any evaluation should include a careful history of past and present symptoms with an exam. Medical tests that look for damage already done are <u>not</u> a substitute for controlling exposure.

Request copies of your medical testing. You have a legal right to this information under the OSHA Access to Employee Exposure and Medical Records Standard (29 CFR 1910.1020).

#### **Mixed Exposures**

- Smoking can cause heart disease, lung cancer, emphysema, and other respiratory problems. It may worsen respiratory conditions caused by chemical exposure. Even if you have smoked for a long time, stopping now will reduce your risk of developing health problems.
- ► More than light alcohol consumption can cause liver damage. Drinking alcohol may increase the liver damage caused by **Epichlorohydrin**.

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# **Workplace Controls and Practices**

Very toxic chemicals, or those that are reproductive hazards or sensitizers, require expert advice on control measures if a less toxic chemical cannot be substituted. Control measures include: (1) enclosing chemical processes for severely irritating and corrosive chemicals, (2) using local exhaust ventilation for chemicals that may be harmful with a single exposure, and (3) using general ventilation to control exposures to skin and eye irritants. For further information on workplace controls, consult the NIOSH document on Control Banding at www.cdc.gov/niosh/topics/ctrlbanding/.

The following work practices are also recommended:

- ▶ Label process containers.
- ▶ Provide employees with hazard information and training.
- ▶ Monitor airborne chemical concentrations.
- ► Use engineering controls if concentrations exceed recommended exposure levels.
- ▶ Provide eye wash fountains and emergency showers.
- Wash or shower if skin comes in contact with a hazardous material
- ▶ Always wash at the end of the workshift.
- Change into clean clothing if clothing becomes contaminated.
- ▶ Do not take contaminated clothing home.
- ▶ Get special training to wash contaminated clothing.
- ▶ Do not eat, smoke, or drink in areas where chemicals are being handled, processed or stored.
- Wash hands carefully before eating, smoking, drinking, applying cosmetics or using the toilet.

In addition, the following may be useful or required:

- Before entering a confined space where Epichlorohydrin may be present, check to make sure that an explosive concentration does not exist.
- Where possible, transfer Epichlorohydrin from drums or other containers to process containers in an enclosed system.

# **Personal Protective Equipment**

The OSHA Personal Protective Equipment Standard (29 CFR 1910.132) requires employers to determine the appropriate personal protective equipment for each hazard and to train employees on how and when to use protective equipment.

The following recommendations are only guidelines and may not apply to every situation.

#### **Gloves and Clothing**

- Avoid skin contact with Epichlorohydrin. Wear personal protective equipment made from material which can not be permeated or degraded by this substance. Safety equipment suppliers and manufacturers can provide recommendations on the most protective glove and clothing material for your operation.
- ► The recommended glove materials for **Epichlorohydrin** are Butyl, Viton and Barrier®.
- ► The recommended protective clothing materials for **Epichlorohydrin** are Tychem® BR, CSM and TK; and Trellchem® HPS and VPS, or the equivalent.

► All protective clothing (suits, gloves, footwear, headgear) should be clean, available each day, and put on before work.

#### **Eye Protection**

- ► Wear indirect-vent, impact and splash resistant goggles when working with liquids.
- ► Wear a face shield along with goggles when working with corrosive, highly irritating or toxic substances.
- Do not wear contact lenses when working with this substance.

#### **Respiratory Protection**

*Improper use of respirators is dangerous.* Respirators should only be used if the employer has implemented a written program that takes into account workplace conditions, requirements for worker training, respirator fit testing, and medical exams, as described in the OSHA Respiratory Protection Standard (29 CFR 1910.134).

- ▶ Where the potential exists for exposure over **0.5 ppm**, use a NIOSH approved supplied-air respirator with a full facepiece operated in a pressure-demand or other positive-pressure mode. For increased protection use in combination with an auxiliary self-contained breathing apparatus or an emergency escape air cylinder.
- ▶ Exposure to **75 ppm** is immediately dangerous to life and health. If the possibility of exposure above **75 ppm** exists, use a NIOSH approved self-contained breathing apparatus with a full facepiece operated in a pressure-demand or other positive-pressure mode equipped with an emergency escape air cylinder.

#### Fire Hazards

If employees are expected to fight fires, they must be trained and equipped as stated in the OSHA Fire Brigades Standard (29 CFR 1910.156).

- ▶ Epichlorohydrin is a FLAMMABLE and REACTIVE LIQUID that can polymerize violently when exposed to HEAT.
- ▶ Use dry chemical, CO₂, water spray or alcohol-resistant foam as extinguishing agents.
- ▶ POISONOUS GASES ARE PRODUCED IN FIRE, including *Hydrogen Chloride* and *Phosgene*.
- ► CONTAINERS MAY EXPLODE IN FIRE.
- ▶ Use water spray to keep fire-exposed containers cool.
- ► Vapor is heavier than air and may travel a distance to cause a fire or explosion far from the source or flash back.
- ► Epichlorohydrin may form an ignitable vapor/air mixture in closed tanks or containers.

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# **Spills and Emergencies**

If employees are required to clean-up spills, they must be properly trained and equipped. The OSHA Hazardous Waste Operations and Emergency Response Standard (29 CFR 1910.120) may apply.

If **Epichlorohydrin** is spilled or leaked, take the following steps:

- ► Evacuate personnel and secure and control entrance to the area
- ▶ Eliminate all ignition sources.
- ► Absorb liquids in dry sand, earth, or a similar material and place into sealed containers for disposal.
- ▶ Ventilate area of spill or leak.
- ► Keep **Epichlorohydrin** out of confined spaces, such as sewers, because of the possibility of an explosion.
- ▶ DO NOT wash into sewer.
- ▶ It may be necessary to contain and dispose of **Epichlorohydrin** as a HAZARDOUS WASTE. Contact your state Department of Environmental Protection (DEP) or your regional office of the federal Environmental Protection Agency (EPA) for specific recommendations.

# **Handling and Storage**

Prior to working with **Epichlorohydrin** you should be trained on its proper handling and storage.

- ▶ Epichlorohydrin can react with HEAT; STRONG ACIDS (such as HYDROCHLORIC, SULFURIC and NITRIC); and STRONG BASES (such as SODIUM HYDROXIDE and POTASSIUM HYDROXIDE) to cause violent and uncontrollable polymerization.
- ▶ Epichlorohydrin may react violently or explosively with OXIDIZING AGENTS (such as PERCHLORATES, PEROXIDES, PERMANGANATES, CHLORATES, NITRATES, CHLORINE, BROMINE and FLUORINE); ALCOHOLS; AMINES (especially ANILINE and ETHYLENE DIAMINE); ALUMINUM; ZINC; METAL SALTS (such as IRON and ALUMINUM CHLORIDE); PHENOLS; POTASSIUM TERT-BUTOXIDE; and WATER.
- ► Epichlorohydrin will react with TRICHLOROETHYLENE to form explosive *Dichloroacetylene*.
- Store in tightly closed containers in a cool, well-ventilated area away from MOISTURE and HEAT.
- Sources of ignition, such as smoking and open flames, are prohibited where Epichlorohydrin is used, handled, or stored.
- ► Metal containers involving the transfer of **Epichlorohydrin** should be grounded and bonded.
- Use explosion-proof electrical equipment and fittings wherever Epichlorohydrin is used, handled, manufactured, or stored.
- ► Use only non-sparking tools and equipment, especially when opening and closing containers of **Epichlorohydrin**.

# Occupational Health Information Resources

The New Jersey Department of Health and Senior Services, Occupational Health Service, offers multiple services in occupational health. These services include providing informational resources, educational materials, public presentations, and industrial hygiene and medical investigations and evaluations.

# For more information, please contact:

New Jersey Department of Health & Senior Services

Right to Know Program

PO Box 368

Trenton, NJ 08625-0368 Phone: 609-984-2202 Fax: 609-984-7407 E-mail: rtk@doh.nj.gov

Web address:

http://www.state.nj.us/health/workplacehealthandsafety/

right-to-know/

The Right to Know Hazardous Substance Fact Sheets are not intended to be copied and sold for commercial purposes.

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#### **GLOSSARY**

**ACGIH** is the American Conference of Governmental Industrial Hygienists. They publish guidelines called Threshold Limit Values (TLVs) for exposure to workplace chemicals.

**Acute Exposure Guideline Levels** (AEGLs) are established by the EPA. They describe the risk to humans resulting from once-in-a lifetime, or rare, exposure to airborne chemicals.

**Boiling point** is the temperature at which a substance can change its physical state from a liquid to a gas.

A **carcinogen** is a substance that causes cancer.

The **CAS number** is unique, identifying number, assigned by the Chemical Abstracts Service, to a specific chemical.

**CFR** is the Code of Federal Regulations, which are the regulations of the United States government.

A combustible substance is a solid, liquid or gas that will burn.

A **corrosive** substance is a gas, liquid or solid that causes destruction of human skin or severe corrosion of containers.

The **critical temperature** is the temperature above which a gas cannot be liquefied, regardless of the pressure applied.

**DEP** is the New Jersey Department of Environmental Protection.

**DOT** is the Department of Transportation, the federal agency that regulates the transportation of chemicals.

**EPA** is the Environmental Protection Agency, the federal agency responsible for regulating environmental hazards.

**ERG** is the Emergency Response Guidebook. It is a guide for emergency responders for transportation emergencies involving hazardous substances.

**Emergency Response Planning Guideline** (ERPG) values provide estimates of concentration ranges where one reasonably might anticipate observing adverse effects.

A fetus is an unborn human or animal.

A **flammable** substance is a solid, liquid, vapor or gas that will ignite easily and burn rapidly.

The **flash point** is the temperature at which a liquid or solid gives off vapor that can form a flammable mixture with air.

**IARC** is the International Agency for Research on Cancer, a scientific group.

**Ionization Potential** is the amount of energy needed to remove an electron from an atom or molecule. It is measured in electron volts.

**IRIS** is the Integrated Risk Information System database on human health effects that may result from exposure to various chemicals, maintained by federal EPA.

**LEL** or **Lower Explosive Limit**, is the lowest concentration of a combustible substance (gas or vapor) in the air capable of continuing an explosion.

mg/m<sup>3</sup> means milligrams of a chemical in a cubic meter of air. It is a measure of concentration (weight/volume).

A **mutagen** is a substance that causes mutations. A **mutation** is a change in the genetic material in a body cell. Mutations can lead to birth defects, miscarriages, or cancer.

**NFPA** is the National Fire Protection Association. It classifies substances according to their fire and explosion hazard.

**NIOSH** is the National Institute for Occupational Safety and Health. It tests equipment, evaluates and approves respirators, conducts studies of workplace hazards, and proposes standards to OSHA.

**NTP** is the National Toxicology Program which tests chemicals and reviews evidence for cancer.

**OSHA** is the federal Occupational Safety and Health Administration, which adopts and enforces health and safety standards.

**PEOSHA** is the New Jersey Public Employees Occupational Safety and Health Act, which adopts and enforces health and safety standards in public workplaces.

**Permeated** is the movement of chemicals through protective materials.

**ppm** means parts of a substance per million parts of air. It is a measure of concentration by volume in air.

**Protective Action Criteria** (PAC) are values established by the Department of Energy and are based on AEGLs and ERPGs. They are used for emergency planning of chemical release events.

A **reactive** substance is a solid, liquid or gas that releases energy under certain conditions.

**STEL** is a Short Term Exposure Limit which is usually a 15-minute exposure that should not be exceeded at any time during a work day.

A **teratogen** is a substance that causes birth defects by damaging the fetus.

**UEL** or **Upper Explosive Limit** is the highest concentration in air above which there is too much fuel (gas or vapor) to begin a reaction or explosion.

**Vapor Density** is the ratio of the weight of a given volume of one gas to the weight of another (usually *Air*), at the same temperature and pressure.

The **vapor pressure** is a force exerted by the vapor in equilibrium with the solid or liquid phase of the same substance. The higher the vapor pressure the higher concentration of the substance in air.



# **Right to Know Hazardous Substance Fact Sheet**



Common Name: EPICHLOROHYDRIN

Synonyms: Chloromethyl Oxirane; 3-Chloropropylene Oxide; 1-Chloro-2,3-Epoxypropane

CAS No: 106-89-8

Molecular Formula: C<sub>3</sub>H<sub>5</sub>ClO RTK Substance No: 0828

Description: Clear, colorless liquid with an irritating odor

HAZARD DATA			
Hazard Rating	Firefighting	Reactivity	
4 - Health	Epichlorohydrin is a FLAMMABLE and REACTIVE LIQUID that can polymerize violently when exposed to	Epichlorohydrin can react with HEAT; STRONG ACIDS (such as HYDROCHLORIC, SULFURIC and NITRIC) and STRONG	
3 - Fire	HEAT.	BASES (such as SODIUM HYDROXIDE and POTASSIUM	
2 - Reactivity	Use dry chemical, CO <sub>2</sub> , water spray or alcohol-resistant foam as extinguishing agents.	HYDROXIDE) to cause violent and uncontrollable polymerization.	
<b>DOT#</b> : UN 2023	POISONOUS GASES ARE PRODUCED IN FIRE, including <i>Hydrogen Chloride</i> and <i>Phosgene</i> .	Epichlorohydrin may react violently or explosively with OXIDIZING AGENTS (such as PERCHLORATES, PEROXIDES,	
ERG Guide #: 131P	CONTAINERS MAY EXPLODE IN FIRE.	PERMANGANATES, CHLORATES, NITRATES, CHLORINE,	
Hazard Class: 6.1 (Poison)	Use water spray to keep fire-exposed containers cool.  Vapor is heavier than air and may travel a distance to cause a fire or explosion far from the source or flash back.	BROMINE and FLUORINE); ALCOHOLS; AMINES (especially ANILINE and ETHYLENE DIAMINE); ALUMINUM; ZINC; METAL SALTS (such as IRON and ALUMINUM CHLORIDE); PHENOLS; POTASSIUM TERT-BUTOXIDE; and WATER.  Epichlorohydrin will react with TRICHLOROETHYLENE to form	
	<b>Epichlorohydrin</b> may form an ignitable vapor/air mixture in closed tanks or containers.	explosive Dichloroacetylene.	

# SPILL/LEAKS

#### **Isolation Distance:**

**Spill:** 50 meters (150 feet) **Fire:** 800 meters (1/2 mile)

Absorb liquids in dry sand, earth, or a similar material and

place into sealed containers for disposal. Use only non-sparking tools and equipment.

Metal containers involving the transfer of **Epichlorohydrin** 

should be grounded and bonded.

Keep **Epichlorohydrin** out of confined spaces, such as sewers, because of the possibility of an explosion.

Epichlorohydrin is harmful to aquatic life.

# **PHYSICAL PROPERTIES**

 Odor Threshold:
 0.08 to 12 ppm

 Flash Point:
 88°F (31°C)

 LEL:
 3.8%

 UEL:
 21%

Auto Ignition Temp:  $772^{\circ}F$  (411°C) Vapor Density: 3.29 (air = 1)

Vapor Pressure: 13 mm Hg at 68°F (20°C)

Specific Gravity:1.17 (water = 1)Water Solubility:Slightly solubleBoiling Point:242°F (117°C)Freezing Point:-54°F (-47.8°C)Ionization Potential:10.6 eVMolecular Weight:92.53

# **EXPOSURE LIMITS**

OSHA: 5 ppm, 8-hr TWA

NIOSH: Lowest feasible concentration

ACGIH: 0.5 ppm, 8-hr TWA

IDLH: 75 ppm

The Protective Action Criteria values are:

PAC-1 = 1.7 ppm PAC-2 = 24 ppm PAC-3 = 72 ppm

# PROTECTIVE EQUIPMENT

Gloves: Butyl, Viton and Barrier® (>8-hr breakthrough)

Coveralls: Tychem® BR, CSM and TK; Trellchem® HPS and VPS (>8-hr

breakthrough)

Respirator: >0.5 ppm - SCBA

# **HEALTH EFFECTS**

Eyes: Irritation and burns

**Skin:** Irritation and burns (skin absorbable)

Inhalation: Nose, throat and lung irritation with coughing,

and severe shortness of breath (pulmonary

edema)

Chronic: Cancer (nasal cavity and skin) in animals

# FIRST AID AND DECONTAMINATION

Remove the person from exposure.

Flush eyes with large amounts of water for at least 30 minutes. Remove

contact lenses if worn. Seek medical attention.

Quickly remove contaminated clothing and wash contaminated skin with

large amounts of water.

Begin artificial respiration if breathing has stopped and CPR if necessary.

**Transfer** promptly to a medical facility.

Medical observation is recommended as symptoms may be delayed.