

ealth Hazardous Substance Fact Sheet

Common Name: LITHIUM CHROMATE

Synonyms: Dilithium Chromate; Chromium Lithium Oxide Chemical Name: Chromic Acid (H₂CrO₄), Dilithium Salt Date: August 1998 Revision: August 2008

Description and Use

Lithium Chromate is a yellow, odorless powder used as a corrosion inhibitor, heat transfer agent, and oxidizing agent in leather and metal finishing. It is also used in photography, wood preservatives, batteries, safety matches and cement.

Reasons for Citation

- Lithium Chromate is on the Right to Know Hazardous Substance List because it is cited by OSHA, ACGIH, DOT, NIOSH, NTP, DEP, IRIS and EPA.
- ► This chemical is on the Special Health Hazard Substance List.

SEE GLOSSARY ON PAGE 5.

Eye Contact

FIRST AID

Immediately flush with large amounts of water for at least 30 minutes, lifting upper and lower lids. Remove contact lenses, if worn, while flushing. Seek medical attention immediately.

Skin Contact

 Quickly remove contaminated clothing. Immediately wash contaminated skin with large amounts of water. Seek medical attention immediately.

Inhalation

- Remove the person from exposure.
- Begin rescue breathing (using universal precautions) if breathing has stopped and CPR if heart action has stopped.
- Transfer promptly to a medical facility.

EMERGENCY NUMBERS

Poison Control: 1-800-222-1222

CHEMTREC: 1-800-424-9300

NJDEP Hotline: 1-877-927-6337

National Response Center: 1-800-424-8802

CAS Number:	14307-35-8
RTK Substance Number:	1125
DOT Number:	UN 3077

EMERGENCY RESPONDERS >>>> SEE BACK PAGE

Hazard Summary			
Hazard Rating	NJDOH	NFPA	
HEALTH	4	-	
FLAMMABILITY	0	-	
REACTIVITY	0	-	

CARCINOGEN OXIDIZER

POISONOUS GASES ARE PRODUCED IN FIRE

Hazard Rating Key: 0=minimal; 1=slight; 2=moderate; 3=serious; 4=severe

- ► Lithium Chromate can affect you when inhaled.
- ► Lithium Chromate is a CARCINOGEN. HANDLE WITH EXTREME CAUTION.
- ► Contact can severely irritate and burn the skin and eyes.
- ► Inhaling Lithium Chromate can irritate the nose, throat and lungs.
- Repeated exposure can cause loss of appetite, nausea, convulsions and personality changes.
- ▶ Lithium Chromate may cause a skin allergy.
- Long-term exposure can cause deep, slow-heeling ulcers on the skin and an ulcer or hole in the "bone" (septum) dividing the inner nose.
- ► Lithium Chromate may damage the liver and kidneys.
- Lithium Chromate is not combustible but is a STRONG OXIDIZER which enhances the combustion of other substances.

Workplace Exposure Limits

The following exposure limits are for *hexavalent Chromium* (or *Cr VI*) *compounds* (measured as *Cr*):

- OSHA: The legal airborne permissible exposure limit (PEL) is **0.005 mg/m³** averaged over an 8-hour workshift.
- NIOSH: The recommended airborne exposure limit (REL) is **0.001 mg/m³** averaged over a 10-hour workshift.
- ACGIH: The threshold limit value (TLV) is **0.05 mg/m³** averaged over an 8-hour workshift.
- Lithium Chromate is a CARCINOGEN in humans. There may be <u>no</u> safe level of exposure to a carcinogen, so all contact should be reduced to the lowest possible level.

Determining Your Exposure

- Read the product manufacturer's Material Safety Data Sheet (MSDS) and the label to determine product ingredients and important safety and health information about the product mixture.
- ► For each individual hazardous ingredient, read the New Jersey Department of Health Hazardous Substance Fact Sheet, available on the RTK website (www.nj.gov/health/eoh/rtkweb) or in your facility's RTK Central File or Hazard Communication Standard file.
- ➤ You have a right to this information under the New Jersey Worker and Community Right to Know Act, the Public Employees Occupational Safety and Health (PEOSH) Act if you are a public worker in New Jersey, and under the federal Occupational Safety and Health Act (OSHA) if you are a private worker.
- The New Jersey Right to Know Act requires most employers to label chemicals in the workplace and requires public employers to provide their employees with information concerning chemical hazards and controls. The federal OSHA Hazard Communication Standard (29 CFR 1910.1200) and the PEOSH Hazard Communication Standard (N.J.A.C. 12:100-7) require employers to provide similar information and training to their employees.

This Fact Sheet is a summary of available information regarding the health hazards that may result from exposure. Duration of exposure, concentration of the substance and other factors will affect your susceptibility to any of the potential effects described below.

Health Hazard Information

Acute Health Effects

The following acute (short-term) health effects may occur immediately or shortly after exposure to **Lithium Chromate**:

- ► Contact can severely irritate and burn the skin and eyes.
- Inhaling Lithium Chromate can irritate the nose and throat causing coughing and wheezing.
- Repeated exposure can cause loss of appetite, nausea, muscle cramps, tremors and convulsions.

Chronic Health Effects

The following chronic (long-term) health effects can occur at some time after exposure to **Lithium Chromate** and can last for months or years:

Cancer Hazard

- Lithium Chromate is a CARCINOGEN in humans. There is evidence that hexavalent Chromium or Chromium VI compounds cause lung cancer in humans.
- Many scientists believe there is no safe level of exposure to a carcinogen.

Reproductive Hazard

While Lithium Chromate has not been identified as a teratogen or a reproductive hazard, certain Chromium compounds are teratogens and may also cause reproductive damage. Lithium Chromate should be handled WITH EXTREME CAUTION.

Other Effects

- Lithium Chromate can irritate the lungs. Repeated exposure may cause bronchitis to develop with coughing, phlegm, and/or shortness of breath.
- ► Lithium Chromate may cause a skin allergy. If allergy develops, very low future exposure can cause itching and a skin rash.
- Repeated exposure may cause personality changes such as depression, anxiety or irritability.
- Long-term exposure can cause deep, slow-heeling ulcers on the skin.
- Inhaling Lithium Chromate can cause a sore and/or a hole in the "bone" (septum) dividing the inner nose, sometimes with bleeding, discharge, and/or formation of a crust.
- ► Lithium Chromate may damage the liver and kidneys.

Medical

Medical Testing

Before first exposure, and every twelve (12) months thereafter, OSHA requires your employer to provide (for persons exposed to levels greater than **2.5 micrograms** of *Chromium VI* **per cubic meter** of air):

- Complete work and medical history
- Thorough physical examination
- Lung function tests

If symptoms develop or overexposure is suspected, the following are recommended:

- Examine your skin periodically for little bumps or blisters, the first sign of "Chrome ulcers." If not treated early, these can last for years after exposure.
- Evaluation by a qualified allergist can help diagnose skin allergy.
- Liver and kidney function tests

Periodic medical surveillance, consisting of medical examinations and biological monitoring, must be provided by your employer.

OSHA requires your employer to provide you and your doctor with a copy of the OSHA *Chromium (VI)* Standard (29 CFR 1910.1026).

Any evaluation should include a careful history of past and present symptoms with an exam. Medical tests that look for damage already done are <u>not</u> a substitute for controlling exposure.

Request copies of your medical testing. You have a legal right to this information under the OSHA Access to Employee Exposure and Medical Records Standard (29 CFR 1910.1020).

LITHIUM CHROMATE

Mixed Exposures

More than light alcohol consumption can cause liver damage. Drinking alcohol can increase the liver damage caused by Lithium Chromate.

Workplace Controls and Practices

Very toxic chemicals, or those that are reproductive hazards or sensitizers, require expert advice on control measures if a less toxic chemical cannot be substituted. Control measures include: (1) enclosing chemical processes for severely irritating and corrosive chemicals, (2) using local exhaust ventilation for chemicals that may be harmful with a single exposure, and (3) using general ventilation to control exposures to skin and eye irritants. For further information on workplace controls, consult the NIOSH document on Control Banding at www.cdc.gov/niosh/topics/ctrlbanding/.

The following work practices are also recommended:

- ► Label process containers.
- Provide employees with hazard information and training.
- Monitor airborne chemical concentrations.
- Use engineering controls if concentrations exceed recommended exposure levels.
- ► Provide eye wash fountains and emergency showers.
- Wash or shower if skin comes in contact with a hazardous material.
- Always wash at the end of the workshift.
- Change into clean clothing if clothing becomes contaminated.
- ► Do not take contaminated clothing home.
- ► Get special training to wash contaminated clothing.
- Do not eat, smoke, or drink in areas where chemicals are being handled, processed or stored.
- ► Wash hands carefully before eating, smoking, drinking, applying cosmetics or using the toilet.

In addition, the following may be useful or required:

- Specific engineering controls are required for this chemical by OSHA. Refer to the OSHA Chromium VI Standard (29 CFR 1910.1026).
- ► Use a vacuum or a wet method to reduce dust during cleanup. DO NOT DRY SWEEP.
- ► Use a high efficiency particulate air (HEPA) filter when vacuuming. Do <u>not</u> use a standard shop vacuum.

Personal Protective Equipment

The OSHA Personal Protective Equipment Standard (29 CFR 1910.132) requires employers to determine the appropriate personal protective equipment for each hazard and to train employees on how and when to use protective equipment.

The following recommendations are only guidelines and may not apply to every situation.

Gloves and Clothing

Avoid skin contact with Lithium Chromate. Wear personal protective equipment made from material that can not be permeated or degraded by this substance. Safety equipment suppliers and manufacturers can provide recommendations on the most protective glove and clothing material for your operation.

- Safety equipment manufacturers recommend Neoprene and Natural Rubber for gloves and DuPont Tyvek®, or the equivalent, as a protective material for clothing.
- All protective clothing (suits, gloves, footwear, headgear) should be clean, available each day, and put on before work.

Eye Protection

- ► Wear eye protection with side shields or goggles.
- Wear non-vented, impact resistant goggles when working with fumes, gases, or vapors.
- ► Wear a face shield along with goggles when working with corrosive, highly irritating or toxic substances.

Respiratory Protection

Improper use of respirators is dangerous. Respirators should only be used if the employer has implemented a written program that takes into account workplace conditions, requirements for worker training, respirator fit testing, and medical exams, as described in the OSHA Respiratory Protection Standard (29 CFR 1910.134).

- ➤ Where the potential exists for exposure over 0.001 mg/m³ (as hexavalent Chromium), use a NIOSH approved airpurifying, particulate filter respirator with an N95 filter. More protection is provided by a full facepiece respirator than by a half-mask respirator, and even greater protection is provided by a powered-air purifying respirator.
- Leave the area immediately if (1) while wearing a filter or cartridge respirator you can smell, taste, or otherwise detect Lithium Chromate, (2) while wearing particulate filters abnormal resistance to breathing is experienced, or (3) eye irritation occurs while wearing a full facepiece respirator. Check to make sure the respirator-to-face seal is still good. If it is, replace the filter or cartridge. If the seal is no longer good, you may need a new respirator.
- Consider all potential sources of exposure in your workplace. You may need a combination of filters, prefilters or cartridges to protect against different forms of a chemical (such as vapor and mist) or against a mixture of chemicals.
- Where the potential exists for exposure over 0.01 mg/m³ (as hexavalent Chromium), use a NIOSH approved supplied-air respirator with a full facepiece operated in a pressure-demand or other positive-pressure mode. For increased protection use in combination with an auxiliary self-contained breathing apparatus operated in a pressure-demand or other positive-pressure mode.
- ► Exposure to **15 mg/m³** (as *Chromates*) is immediately dangerous to life and health. If the possibility of exposure above **15 mg/m³** exists, use a NIOSH approved self-contained breathing apparatus with a full facepiece operated in a pressure-demand or other positive-pressure mode equipped with an emergency escape air cylinder.

LITHIUM CHROMATE

Fire Hazards

If employees are expected to fight fires, they must be trained and equipped as stated in the OSHA Fire Brigades Standard (29 CFR 1910.156).

- Extinguish fire using an agent suitable for type of surrounding fire. Lithium Chromate itself does not burn.
- ► POISONOUS GASES ARE PRODUCED IN FIRE, including Lithium Oxides and Chromium Oxides.
- Use water spray to keep fire-exposed containers cool.
- Lithium Chromate may ignite combustibles (wood, paper and oil).

Spills and Emergencies

If employees are required to clean-up spills, they must be properly trained and equipped. The OSHA Hazardous Waste Operations and Emergency Response Standard (29 CFR 1910.120) may apply.

If Lithium Chromate is spilled, take the following steps:

- Evacuate personnel and secure and control entrance to the area.
- ► Eliminate all ignition sources.
- Moisten spilled material first or use a HEPA-filter vacuum for clean-up and deposit into sealed containers.
- ► Ventilate and wash area after clean-up is complete.
- DO NOT wash into sewer.
- It may be necessary to contain and dispose of Lithium Chromate as a HAZARDOUS WASTE. Contact your state Department of Environmental Protection (DEP) or your regional office of the federal Environmental Protection Agency (EPA) for specific recommendations.

Handling and Storage

Prior to working with **Lithium Chromate** you should be trained on its proper handling and storage.

- A regulated, marked area should be established where Lithium Chromate is handled, used or stored as required by the OSHA Chromium VI Standard (29 CFR 1910.1026).
- Lithium Chromate is an OXIDIZER which can react with ORGANICS; COMBUSTIBLES; REDUCING AGENTS (such as LITHIUM, SODIUM, ALUMINUM and their HYDRIDES); and STRONG ACIDS (such as HYDROCHLORIC, SULFURIC and NITRIC).
- ► Mixtures of *Chromate* and ZIRCONIUM can be explosive.
- Store in tightly closed containers in a cool, well-ventilated area.

Occupational Health Information Resources

The New Jersey Department of Health offers multiple services in occupational health. These services include providing informational resources, educational materials, public presentations, and industrial hygiene and medical investigations and evaluations.

For more information, please contact:

New Jersey Department of Health Right to Know PO Box 368 Trenton, NJ 08625-0368 Phone: 609-984-2202 Fax: 609-984-7407 E-mail: rtk@doh.state.nj.us Web address: http://www.nj.gov/health/eoh/rtkweb

The Right to Know Hazardous Substance Fact Sheets are not intended to be copied and sold for commercial purposes.

LITHIUM CHROMATE

GLOSSARY

ACGIH is the American Conference of Governmental Industrial Hygienists. They publish guidelines called Threshold Limit Values (TLVs) for exposure to workplace chemicals.

Acute Exposure Guideline Levels (AEGLs) are established by the EPA. They describe the risk to humans resulting from once-in-a lifetime, or rare, exposure to airborne chemicals.

Boiling point is the temperature at which a substance can change its physical state from a liquid to a gas.

A carcinogen is a substance that causes cancer.

The **CAS number** is unique, identifying number, assigned by the Chemical Abstracts Service, to a specific chemical.

CFR is the Code of Federal Regulations, which are the regulations of the United States government.

A combustible substance is a solid, liquid or gas that will burn.

A **corrosive** substance is a gas, liquid or solid that causes destruction of human skin or severe corrosion of containers.

DEP is the New Jersey Department of Environmental Protection.

DOT is the Department of Transportation, the federal agency that regulates the transportation of chemicals.

EPA is the Environmental Protection Agency, the federal agency responsible for regulating environmental hazards.

ERG is the Emergency Response Guidebook. It is a guide for emergency responders for transportation emergencies involving hazardous substances.

Emergency Response Planning Guideline (ERPG) values are intended to provide estimates of concentration ranges where one reasonably might anticipate observing adverse effects.

A fetus is an unborn human or animal.

A **flammable** substance is a solid, liquid, vapor or gas that will ignite easily and burn rapidly.

The **flash point** is the temperature at which a liquid or solid gives off vapor that can form a flammable mixture with air.

IARC is the International Agency for Research on Cancer, a scientific group.

Ionization Potential is the amount of energy needed to remove an electron from an atom or molecule. It is measured in electron volts.

IRIS is the Integrated Risk Information System database maintained by federal EPA. The database contains information on human health effects that may result from exposure to various chemicals in the environment.

LEL or **Lower Explosive Limit**, is the lowest concentration of a combustible substance (gas or vapor) in the air capable of continuing an explosion.

mg/m³ means milligrams of a chemical in a cubic meter of air. It is a measure of concentration (weight/volume).

A **mutagen** is a substance that causes mutations. A **mutation** is a change in the genetic material in a body cell. Mutations can lead to birth defects, miscarriages, or cancer.

NFPA is the National Fire Protection Association. It classifies substances according to their fire and explosion hazard.

NIOSH is the National Institute for Occupational Safety and Health. It tests equipment, evaluates and approves respirators, conducts studies of workplace hazards, and proposes standards to OSHA.

NTP is the National Toxicology Program which tests chemicals and reviews evidence for cancer.

OSHA is the federal Occupational Safety and Health Administration, which adopts and enforces health and safety standards.

PEOSHA is the New Jersey Public Employees Occupational Safety and Health Act, which adopts and enforces health and safety standards in public workplaces.

Permeated is the movement of chemicals through protective materials.

PIH is a DOT designation for chemicals which are Poison Inhalation Hazards.

ppm means parts of a substance per million parts of air. It is a measure of concentration by volume in air.

A **reactive** substance is a solid, liquid or gas that releases energy under certain conditions.

STEL is a Short Term Exposure Limit which is usually a 15minute exposure that should not be exceeded at any time during a work day.

A **teratogen** is a substance that causes birth defects by damaging the fetus.

UEL or **Upper Explosive Limit** is the highest concentration in air above which there is too much fuel (gas or vapor) to begin a reaction or explosion.

Vapor Density is the ratio of the weight of a given volume of one gas to the weight of another (usually *Hydrogen*), at the same temperature and pressure.

The **vapor pressure** is a measure of how readily a liquid or a solid mixes with air at its surface. A higher vapor pressure indicates a higher concentration of the substance in air and therefore increases the likelihood of breathing it in.



Common Name: LITHIUM CHROMATE

Synonyms: Dilithium Chromate; Chromium Lithium Oxide CAS No: 14307-35-8 Molecular Formula: Li_2CrO_4 RTK Substance No: 1125 Description: Yellow, odorless powder

HAZARD DATA

Hazard Rating	Firefighting	Reactivity
4 - Health	Extinguish fire using an agent suitable for type of surrounding fire. Lithium Chromate itself does	Lithium Chromate is an OXIDIZER which can react with ORGANICS; COMBUSTIBLES; REDUCING AGENTS
0 - Fire	not burn.	(such as LITHIUM, SODIUM, ALUMINUM and their
0 - Reactivity	POISONOUS GASES ARE PRODUCED IN FIRE, including <i>Lithium Oxides and Chromium Oxides</i> .	HYDRIDES); and STRONG ACIDS (such as HYDROCHLORIC, SULFURIC and NITRIC).
DOT#: UN 3077	Use water spray to keep fire-exposed containers	Mixtures of <i>Chromate</i> and ZIRCONIUM can be explosive.
ERG Guide #: 171	cool.	
Hazard Class: 5.1	Lithium Chromate may ignite combustibles (wood, paper and oil).	
(Oxidizer)	paper and onj.	

SPILL/LEAKS

Isolation Distance:

Spill: 25 meters (75 feet)Fire: 800 meters (1/2 mile)Moisten spilled material first or use a HEPA-filter vacuum for clean-up and deposit into sealed containers.DO NOT wash into sewer.

May be detrimental to aquatic life.

EXPOSURE LIMITS

OSHA: 0.005 mg/m³, 8-hr TWA NIOSH: 0.001 mg/m³, 10-hr TWA ACGIH: 0.05 mg/m³, 8-hr TWA IDLH: 15 mg/m³ (as *Chromates*) All of the above are for *hexavalent Chromium* (*Cr VI*)

HEALTH EFFECTS

Eyes: Skin:	Severe irritation and burns Severe irritation and burns
Inhalation:	Nose, throat and lung irritation with coughing and wheezing
	Nausea, muscle cramps and convulsions
Chronic:	Hexavalent Chromium or Chromium VI compounds cause lung cancer in humans

PHYSICAL PROPERTIES

Odor Threshold:	Odorless
Flash Point:	Nonflammable
Vapor Density:	5.72 (air = 1)
Specific Gravity:	2.2 (water = 1)
Water Solubility:	Soluble
Melting Point:	166°F (75°C)
Molecular Weight:	130

PROTECTIVE EQUIPMENT

Gloves:	Neoprene and Natural Rubber
Coveralls:	DuPont Tyvek®
Respirator:	>0.001 mg/m ³ - APR with High efficiency filters >0.01 mg/m ³ - Supplied air

FIRST AID AND DECONTAMINATION

Remove the person from exposure.

Flush eyes with large amounts of water for at least 30 minutes. Remove contact lenses if worn. Seek medical attention immediately.

Quickly remove contaminated clothing and wash contaminated skin with large amounts of water. Seek medical attention immediately.

Begin artificial respiration if breathing has stopped and CPR if necessary. **Transfer** promptly to a medical facility.