



Right to Know Hazardous Substance Fact Sheet

Common Name: **METHYL ETHYL KETONE PEROXIDE**

Synonyms: MEKP; MEK Peroxide

Chemical Name: 2-Butanone, Peroxide

Date: October 1999 Revision: March 2009

CAS Number: 1338-23-4

RTK Substance Number: 1259

DOT Number: UN 3101

Description and Use

Methyl Ethyl Ketone Peroxide is a colorless to yellow liquid with a fragrant, mint-like odor. It is used in making plastic monomers, as a curing agent in polyester resins, and as a cross-linking agent and catalyst in making polymers. The commercial product is sold in a solution of 60% **Methyl Ethyl Ketone Peroxide** and 40% diluent to prevent explosions.

Reasons for Citation

- ▶ **Methyl Ethyl Ketone Peroxide** is on the Right to Know Hazardous Substance List because it is cited by ACGIH, DOT, NIOSH and EPA.
- ▶ This chemical is on the Special Health Hazard Substance List.

[SEE GLOSSARY ON PAGE 5.](#)

FIRST AID

Eye Contact

- ▶ Immediately flush with large amounts of water for at least 15 minutes, lifting upper and lower lids. Remove contact lenses, if worn, while flushing. Seek medical attention immediately.

Skin Contact

- ▶ Quickly remove contaminated clothing. Immediately wash contaminated skin with large amounts of soap and water.

Inhalation

- ▶ Remove the person from exposure.
- ▶ Begin rescue breathing (using universal precautions) if breathing has stopped and CPR if heart action has stopped.
- ▶ Transfer promptly to a medical facility.

EMERGENCY NUMBERS

Poison Control: 1-800-222-1222

CHEMTREC: 1-800-424-9300

NJDEP Hotline: 1-877-927-6337

National Response Center: 1-800-424-8802

EMERGENCY RESPONDERS >>>> SEE LAST PAGE

Hazard Summary

Hazard Rating	NJDOH	NFPA
HEALTH	3	-
FLAMMABILITY	2	-
REACTIVITY	4	-
STRONG OXIDIZER COMBUSTIBLE AND REACTIVE POISONOUS GASES ARE PRODUCED IN FIRE CONTAINERS MAY EXPLODE IN FIRE		

Hazard Rating Key: 0=minimal; 1=slight; 2=moderate; 3=serious; 4=severe

- ▶ **Methyl Ethyl Ketone Peroxide** can affect you when inhaled and may be absorbed through the skin.
- ▶ Contact can severely irritate and burn the skin and eyes leading to eye damage.
- ▶ Inhaling **Methyl Ethyl Ketone Peroxide** can irritate the nose, throat and lungs.
- ▶ **Methyl Ethyl Ketone Peroxide** may cause a skin allergy.
- ▶ **Methyl Ethyl Ketone Peroxide** may damage the liver and kidneys.
- ▶ **Methyl Ethyl Ketone Peroxide** is REACTIVE and a DANGEROUS EXPLOSION HAZARD.
- ▶ **Methyl Ethyl Ketone Peroxide** is a STRONG OXIDIZER that may ignite or explode on contact with other substances.

Workplace Exposure Limits

NIOSH: The recommended airborne exposure limit (REL) is **0.2 ppm**, which should not be exceeded at any time.

ACGIH: The threshold limit value (TLV) is **0.2 ppm**, which should not be exceeded at any time.

- ▶ The above exposure limits are for air levels only. When skin contact also occurs, you may be overexposed, even though air levels are less than the limits listed above.

Determining Your Exposure

- ▶ Read the product manufacturer's Material Safety Data Sheet (MSDS) and the label to determine product ingredients and important safety and health information about the product mixture.
- ▶ For each individual hazardous ingredient, read the New Jersey Department of Health Hazardous Substance Fact Sheet, available on the RTK website (www.nj.gov/health/eoh/rtkweb) or in your facility's RTK Central File or Hazard Communication Standard file.
- ▶ You have a right to this information under the New Jersey Worker and Community Right to Know Act, the Public Employees Occupational Safety and Health (PEOSH) Act if you are a public worker in New Jersey, and under the federal Occupational Safety and Health Act (OSHA) if you are a private worker.
- ▶ The New Jersey Right to Know Act requires most employers to label chemicals in the workplace and requires public employers to provide their employees with information concerning chemical hazards and controls. The federal OSHA Hazard Communication Standard (29 CFR 1910.1200) and the PEOSH Hazard Communication Standard (N.J.A.C. 12:100-7) require employers to provide similar information and training to their employees.

This Fact Sheet is a summary of available information regarding the health hazards that may result from exposure. Duration of exposure, concentration of the substance and other factors will affect your susceptibility to any of the potential effects described below.

Health Hazard Information

Acute Health Effects

The following acute (short-term) health effects may occur immediately or shortly after exposure to **Methyl Ethyl Ketone Peroxide**:

- ▶ Contact can severely irritate and burn the skin and eyes leading to eye damage.
- ▶ Inhaling **Methyl Ethyl Ketone Peroxide** can irritate the nose, throat and lungs causing coughing, wheezing and/or shortness of breath.

Chronic Health Effects

The following chronic (long-term) health effects can occur at some time after exposure to **Methyl Ethyl Ketone Peroxide** and can last for months or years:

Cancer Hazard

- ▶ According to the information presently available to the New Jersey Department of Health, **Methyl Ethyl Ketone Peroxide** has not been tested for its ability to cause cancer in animals.

Reproductive Hazard

- ▶ According to the information presently available to the New Jersey Department of Health, **Methyl Ethyl Ketone Peroxide** has not been tested for its ability to affect reproduction.

Other Effects

- ▶ **Methyl Ethyl Ketone Peroxide** can irritate the lungs. Repeated exposure may cause bronchitis to develop with coughing, phlegm, and/or shortness of breath.
- ▶ **Methyl Ethyl Ketone Peroxide** may cause a skin allergy. If allergy develops, very low future exposure can cause itching and a skin rash.
- ▶ **Methyl Ethyl Ketone Peroxide** may damage the liver and kidneys.

Medical

Medical Testing

Before beginning employment and at regular times thereafter, (at least annually), the following are recommended:

- ▶ Liver and kidney function tests

If symptoms develop or overexposure is suspected, the following are recommended:

- ▶ Evaluation by a qualified allergist can help diagnose skin allergy.
- ▶ Lung function tests

Any evaluation should include a careful history of past and present symptoms with an exam. Medical tests that look for damage already done are not a substitute for controlling exposure.

Request copies of your medical testing. You have a legal right to this information under the OSHA Access to Employee Exposure and Medical Records Standard (29 CFR 1910.1020).

Mixed Exposures

- ▶ Smoking can cause heart disease, lung cancer, emphysema, and other respiratory problems. It may worsen respiratory conditions caused by chemical exposure. Even if you have smoked for a long time, stopping now will reduce your risk of developing health problems.
- ▶ More than light alcohol consumption can cause liver damage. Drinking alcohol can increase the liver damage caused by **Methyl Ethyl Ketone Peroxide**.

Workplace Controls and Practices

Very toxic chemicals, or those that are reproductive hazards or sensitizers, require expert advice on control measures if a less toxic chemical cannot be substituted. Control measures include: (1) enclosing chemical processes for severely irritating and corrosive chemicals, (2) using local exhaust ventilation for chemicals that may be harmful with a single exposure, and (3) using general ventilation to control exposures to skin and eye irritants. For further information on workplace controls, consult the NIOSH document on Control Banding at www.cdc.gov/niosh/topics/ctrlbanding/.

The following work practices are also recommended:

- ▶ Label process containers.
- ▶ Provide employees with hazard information and training.
- ▶ Monitor airborne chemical concentrations.
- ▶ Use engineering controls if concentrations exceed recommended exposure levels.
- ▶ Provide eye wash fountains and emergency showers.
- ▶ Wash or shower if skin comes in contact with a hazardous material.
- ▶ Always wash at the end of the workshift.
- ▶ Change into clean clothing if clothing becomes contaminated.
- ▶ Do not take contaminated clothing home.
- ▶ Get special training to wash contaminated clothing.
- ▶ Do not eat, smoke, or drink in areas where chemicals are being handled, processed or stored.
- ▶ Wash hands carefully before eating, smoking, drinking, applying cosmetics or using the toilet.

In addition, the following may be useful or required:

- ▶ Before entering a confined space where **Methyl Ethyl Ketone Peroxide** may be present, check to make sure that an explosive concentration does not exist.

Personal Protective Equipment

The OSHA Personal Protective Equipment Standard (29 CFR 1910.132) requires employers to determine the appropriate personal protective equipment for each hazard and to train employees on how and when to use protective equipment.

The following recommendations are only guidelines and may not apply to every situation.

Gloves and Clothing

- ▶ Avoid skin contact with **Methyl Ethyl Ketone Peroxide**. Wear personal protective equipment made from material which can not be permeated or degraded by this substance. Safety equipment suppliers and manufacturers can provide recommendations on the most protective glove and clothing material for your operation.
- ▶ Safety equipment manufacturers recommend Butyl, Neoprene and Viton for gloves, and Tychem® BR, LV, Responder®, and TK, or the equivalent, as protective clothing materials for *Peroxides*.
- ▶ All protective clothing (suits, gloves, footwear, headgear) should be clean, available each day, and put on before work.

Eye Protection

- ▶ Wear indirect-vent, impact and splash resistant goggles when working with liquids.
- ▶ Wear a face shield along with goggles when working with corrosive, highly irritating or toxic substances.

Respiratory Protection

Improper use of respirators is dangerous. Respirators should only be used if the employer has implemented a written program that takes into account workplace conditions, requirements for worker training, respirator fit testing, and medical exams, as described in the OSHA Respiratory Protection Standard (29 CFR 1910.134).

- ▶ Where the potential exists for exposure over **0.2 ppm**, use a NIOSH approved supplied-air respirator with a full facepiece operated in a pressure-demand or other positive-pressure mode. For increased protection use in combination with an auxiliary self-contained breathing apparatus operated in a pressure-demand or other positive-pressure mode.

Fire Hazards

If employees are expected to fight fires, they must be trained and equipped as stated in the OSHA Fire Brigades Standard (29 CFR 1910.156).

- ▶ **Methyl Ethyl Ketone Peroxide** is a COMBUSTIBLE and REACTIVE LIQUID that may burn slowly at first, and after heating can burn violently and explosively.
- ▶ Use dry chemical, CO₂, water spray or alcohol-resistant foam as extinguishing agents.
- ▶ POISONOUS GASES ARE PRODUCED IN FIRE.
- ▶ CONTAINERS MAY EXPLODE IN FIRE.
- ▶ Use water spray to keep fire-exposed containers cool.
- ▶ **Methyl Ethyl Ketone Peroxide** is a STRONG OXIDIZER that may ignite or explode on contact with other substances.

Spills and Emergencies

If employees are required to clean-up spills, they must be properly trained and equipped. The OSHA Hazardous Waste Operations and Emergency Response Standard (29 CFR 1910.120) may apply.

If **Methyl Ethyl Ketone Peroxide** is spilled or leaked, take the following steps:

- ▶ Evacuate personnel and secure and control entrance to the area.
- ▶ Eliminate all ignition sources.
- ▶ Absorb liquid with fly ash, cement powder or commercial sorbents and place into sealed containers for disposal.
- ▶ Ventilate and wash area after clean-up is complete.
- ▶ Keep **Methyl Ethyl Ketone Peroxide** out of confined spaces, such as sewers, because of the possibility of an explosion.
- ▶ DO NOT wash into sewer.
- ▶ It may be necessary to contain and dispose of **Methyl Ethyl Ketone Peroxide** as a HAZARDOUS WASTE. Contact your state Department of Environmental Protection (DEP) or your regional office of the federal Environmental Protection Agency (EPA) for specific recommendations.

Handling and Storage

Prior to working with **Methyl Ethyl Ketone Peroxide** you should be trained on its proper handling and storage.

- ▶ **Methyl Ethyl Ketone Peroxide** (not in a solution or diluted) decomposes explosively at temperatures above 230°F (110°C) or if exposed to SHOCK or FRICTION.
- ▶ **Methyl Ethyl Ketone Peroxide** reacts violently with OXIDIZING AGENTS (such as PERCHLORATES, PEROXIDES, PERMANGANATES, CHLORATES, NITRATES, CHLORINE, BROMINE and FLUORINE); COMBUSTIBLES; STRONG ACIDS (such as HYDROCHLORIC, SULFURIC and NITRIC); STRONG BASES (such as SODIUM HYDROXIDE and POTASSIUM HYDROXIDE); REDUCING AGENTS (such as LITHIUM, SODIUM, ALUMINUM and their HYDRIDES); ORGANICS; and METALS (such as IRON, STEEL, COPPER, and ALUMINUM and their ALLOYS).
- ▶ **Methyl Ethyl Ketone Peroxide** is not compatible with MINERAL ACIDS; COBALT COMPOUNDS; AMINES; and ACETONE.
- ▶ Store in tightly closed containers in a cool, well-ventilated area away from SUNLIGHT, HEAT, and STEAM PIPES, and protect from SHOCK, FRICTION and CONTAMINATION.
- ▶ Sources of ignition, such as smoking and open flames, are prohibited where **Methyl Ethyl Ketone Peroxide** is used, handled, or stored in a manner that could create a potential fire or explosion hazard.
- ▶ Metal containers involving the transfer of **Methyl Ethyl Ketone Peroxide** should be grounded and bonded.
- ▶ Use only non-sparking tools and equipment, especially when opening and closing containers of **Methyl Ethyl Ketone Peroxide**.

Occupational Health Information Resources

The New Jersey Department of Health offers multiple services in occupational health. These services include providing informational resources, educational materials, public presentations, and industrial hygiene and medical investigations and evaluations.

For more information, please contact:

New Jersey Department of Health
Right to Know
PO Box 368
Trenton, NJ 08625-0368
Phone: 609-984-2202
Fax: 609-984-7407
E-mail: rtk@doh.state.nj.us
Web address: <http://www.nj.gov/health/eoh/rtkweb>

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for commercial purposes.*

GLOSSARY

ACGIH is the American Conference of Governmental Industrial Hygienists. They publish guidelines called Threshold Limit Values (TLVs) for exposure to workplace chemicals.

Acute Exposure Guideline Levels (AEGs) are established by the EPA. They describe the risk to humans resulting from once-in-a-lifetime, or rare, exposure to airborne chemicals.

Boiling point is the temperature at which a substance can change its physical state from a liquid to a gas.

A **carcinogen** is a substance that causes cancer.

The **CAS number** is unique, identifying number, assigned by the Chemical Abstracts Service, to a specific chemical.

CFR is the Code of Federal Regulations, which are the regulations of the United States government.

A **combustible** substance is a solid, liquid or gas that will burn.

A **corrosive** substance is a gas, liquid or solid that causes destruction of human skin or severe corrosion of containers.

DEP is the New Jersey Department of Environmental Protection.

DOT is the Department of Transportation, the federal agency that regulates the transportation of chemicals.

EPA is the Environmental Protection Agency, the federal agency responsible for regulating environmental hazards.

ERG is the Emergency Response Guidebook. It is a guide for emergency responders for transportation emergencies involving hazardous substances.

Emergency Response Planning Guideline (ERPG) values provide estimates of concentration ranges where one reasonably might anticipate observing adverse effects.

A **fetus** is an unborn human or animal.

A **flammable** substance is a solid, liquid, vapor or gas that will ignite easily and burn rapidly.

The **flash point** is the temperature at which a liquid or solid gives off vapor that can form a flammable mixture with air.

IARC is the International Agency for Research on Cancer, a scientific group.

Ionization Potential is the amount of energy needed to remove an electron from an atom or molecule. It is measured in electron volts.

IRIS is the Integrated Risk Information System database on human health effects that may result from exposure to various chemicals, maintained by federal EPA.

LEL or Lower Explosive Limit, is the lowest concentration of a combustible substance (gas or vapor) in the air capable of continuing an explosion.

mg/m³ means milligrams of a chemical in a cubic meter of air. It is a measure of concentration (weight/volume).

A **mutagen** is a substance that causes mutations. A **mutation** is a change in the genetic material in a body cell. Mutations can lead to birth defects, miscarriages, or cancer.

NFPA is the National Fire Protection Association. It classifies substances according to their fire and explosion hazard.

NIOSH is the National Institute for Occupational Safety and Health. It tests equipment, evaluates and approves respirators, conducts studies of workplace hazards, and proposes standards to OSHA.

NTP is the National Toxicology Program which tests chemicals and reviews evidence for cancer.

OSHA is the federal Occupational Safety and Health Administration, which adopts and enforces health and safety standards.

PEOSHA is the New Jersey Public Employees Occupational Safety and Health Act, which adopts and enforces health and safety standards in public workplaces.

Permeated is the movement of chemicals through protective materials.

ppm means parts of a substance per million parts of air. It is a measure of concentration by volume in air.

Protective Action Criteria (PAC) are values established by the Department of Energy and are based on AEGs and ERPGs. They are used for emergency planning of chemical release events.

A **reactive** substance is a solid, liquid or gas that releases energy under certain conditions.

STEL is a Short Term Exposure Limit which is usually a 15-minute exposure that should not be exceeded at any time during a work day.

A **teratogen** is a substance that causes birth defects by damaging the fetus.

UEL or Upper Explosive Limit is the highest concentration in air above which there is too much fuel (gas or vapor) to begin a reaction or explosion.

Vapor Density is the ratio of the weight of a given volume of one gas to the weight of another (usually *Hydrogen*), at the same temperature and pressure.

The **vapor pressure** is a force exerted by the vapor in equilibrium with the solid or liquid phase of the same substance. The higher the vapor pressure the higher concentration of the substance in air.

Common Name: **METHYL ETHYL KETONE PEROXIDE**

Synonyms: 2-Butanone Peroxide; MEKP; MEK Peroxide

CAS No: 1338-23-4

 Molecular Formula: C₈H₁₆O₄

RTK Substance No: 1259

 Description: Colorless to yellow liquid with a fragrant, mint-like odor, usually sold in 60% solution (with a *Phthalate* diluent) to prevent explosions

HAZARD DATA

Hazard Rating	Firefighting	Reactivity
3 - Health 2 - Fire 4 - Reactivity DOT#: UN 3101 ERG Guide #: 146 Hazard Class: 5.2 (Organic Peroxide)	Methyl Ethyl Ketone Peroxide is a COMBUSTIBLE and REACTIVE LIQUID that may burn slowly at first, and after heating can burn violently and explosively. Use dry chemical, CO ₂ , water spray or alcohol-resistant foam as extinguishing agents. POISONOUS GASES ARE PRODUCED IN FIRE. CONTAINERS MAY EXPLODE IN FIRE. Use water spray to keep fire-exposed containers cool. Methyl Ethyl Ketone Peroxide is a STRONG OXIDIZER that may ignite or explode on contact with other substances.	Methyl Ethyl Ketone Peroxide (not in solution or diluted) decomposes explosively at temperatures above 230°F (110°C) or if exposed to SHOCK or FRICTION. Methyl Ethyl Ketone Peroxide reacts violently with OXIDIZING AGENTS (such as PERCHLORATES, PEROXIDES, PERMANGANATES, CHLORATES, NITRATES, CHLORINE, BROMINE and FLUORINE); COMBUSTIBLES; STRONG ACIDS (such as HYDROCHLORIC, SULFURIC and NITRIC); STRONG BASES (such as SODIUM HYDROXIDE and POTASSIUM HYDROXIDE); REDUCING AGENTS (such as LITHIUM, SODIUM, ALUMINUM and their HYDRIDES); ORGANICS; METALS (such as IRON, STEEL, COPPER and ALUMINUM, and their ALLOYS). Methyl Ethyl Ketone Peroxide is not compatible with MINERAL ACIDS; COBALT COMPOUNDS; AMINES; and ACETONE.

SPILL/LEAKS
Isolation Distance:

Spill: 50 meters (150 feet)

Fire: 800 meters (1/2 mile)

Absorb liquid with fly ash, cement powder or commercial sorbents and place into sealed containers for disposal.

 Use only non-sparking tools and equipment, especially when opening and closing containers of **Methyl Ethyl Ketone Peroxide**.

 Keep **Methyl Ethyl Ketone Peroxide** out of confined spaces, such as sewers, because of the possibility of an explosion.

DO NOT wash into sewer.

PHYSICAL PROPERTIES
Odor Threshold: Fragrant, mint-like odor

Flash Point: 125° to 200°F (52° to 93°C)

Auto Ignition Temp: 1,051°F (566°C)

Vapor Density: 6.7 (air = 1)

Vapor Pressure: <0.01 mm Hg at 68°F (20°C)

Specific Gravity: 1.12 (water = 1)

Water Solubility: Soluble

Boiling Point: 244°F (118°C), violent decomposition at 230°F (110°C)

Molecular Weight: 176.2

EXPOSURE LIMITS
NIOSH: 0.2 ppm, Ceiling

ACGIH: 0.2 ppm, Ceiling

The Protective Action Criteria values are:

PAC-1 = 3 ppm PAC-3 = 20 ppm

PAC-2 = 20 ppm

PROTECTIVE EQUIPMENT
Gloves: Butyl, Neoprene and Viton (>4-hr breakthrough)

Coveralls: Tychem® BR, LV, Responder®, and TK (>8-hr breakthrough for *Peroxides*)

Respirator: >0.2 ppm - SCBA

HEALTH EFFECTS
Eyes: Severe irritation, and burns with possible damage

Skin: Severe irritation and burns

Inhalation: Nose, throat and lung irritation, with coughing, wheezing and shortness of breath

FIRST AID AND DECONTAMINATION
Remove the person from exposure.

Flush eyes with large amounts of water for at least 15 minutes. Remove contact lenses if worn. Seek medical attention immediately.

Quickly remove contaminated clothing and wash contaminated skin with large amounts of soap and water.

Begin artificial respiration if breathing has stopped and CPR if necessary.

Transfer promptly to a medical facility.