Common Name:  o-PHENYLPHENOL

Synonyms:  2-Biphenylol; 2-Hydroxydiphenyl; 2-Phenylphenol
Chemical Name:  \([1,1'-\text{Biphenyl}]\)-2-ol
Date:  December 2000  Revision:  March 2010

**Cas Number:**  90-43-7
**RTK Substance Number:**  1439
**DOT Number:**  None

---

**Description and Use**

o-Phenylphenol is a white, buff, to light lavender, crystalline (sand-like) solid. It is used as a fungicide, germicide, household disinfectant, and laboratory reagent.

---

**Reasons for Citation**

- o-Phenylphenol is on the Right to Know Hazardous Substance List because it is cited by DEP, IARC, NFPA and EPA.

---

**FIRST AID**

**Eye Contact**
- Immediately flush with large amounts of water for at least 30 minutes, lifting upper and lower lids. Remove contact lenses, if worn, while flushing. Seek medical attention immediately.

**Skin Contact**
- Quickly remove contaminated clothing. Immediately wash contaminated skin with large amounts of soap and water. Seek medical attention.

**Inhalation**
- Remove the person from exposure.
- Begin rescue breathing (using universal precautions) if breathing has stopped and CPR if heart action has stopped.
- Transfer promptly to a medical facility.

---

**EMERGENCY NUMBERS**

Poison Control:  1-800-222-1222
CHEMTREC:  1-800-424-9300
NJDEP Hotline:  1-877-927-6337
National Response Center:  1-800-424-8802

---

**Hazard Summary**

<table>
<thead>
<tr>
<th>Hazard Rating</th>
<th>NJDOH</th>
<th>NFPA</th>
</tr>
</thead>
<tbody>
<tr>
<td>HEALTH</td>
<td></td>
<td>3</td>
</tr>
<tr>
<td>FLAMMABILITY</td>
<td></td>
<td>1</td>
</tr>
<tr>
<td>REACTIVITY</td>
<td></td>
<td>0</td>
</tr>
</tbody>
</table>

**Hazard Rating Key:** 0=Minimal; 1=Slight; 2=Moderate; 3=Serious; 4=Severe

- o-Phenylphenol can affect you when inhaled and by passing through the skin.
- o-Phenylphenol can cause skin irritation.
- Contact can severely irritate and burn the eyes with possible permanent damage.
- Inhaling o-Phenylphenol can irritate the nose and throat causing coughing and wheezing.
- o-Phenylphenol may affect the kidneys.

---

**Workplace Exposure Limits**

No occupational exposure limits have been established for o-Phenylphenol. However, it may pose a health risk. Always follow safe work practices.

- It should be recognized that o-Phenylphenol can be absorbed through your skin, thereby increasing your exposure.

---

**See Glossary on Page 5.**
Determining Your Exposure

- Read the product manufacturer’s Material Safety Data Sheet (MSDS) and the label to determine product ingredients and important safety and health information about the product mixture.

- For each individual hazardous ingredient, read the New Jersey Department of Health Hazardous Substance Fact Sheet, available on the RTK website (www.nj.gov/health/eoh/rtkweb) or in your facility’s RTK Central File or Hazard Communication Standard file.

- You have a right to this information under the New Jersey Worker and Community Right to Know Act and the Public Employees Occupational Safety and Health (PEOSH) Act if you are a public worker in New Jersey, and under the federal Occupational Safety and Health Act (OSHA) if you are a private worker.

- The New Jersey Right to Know Act requires most employers to label chemicals in the workplace and requires public employers to provide their employees with information concerning chemical hazards and controls. The federal OSHA Hazard Communication Standard (29 CFR 1910.1200) and the PEOSH Hazard Communication Standard (N.J.A.C. 12:100-7) require employers to provide similar information and training to their employees.

Other Effects

- o-Phenylphenol may affect the kidneys.

Medical

Medical Testing

If symptoms develop or overexposure is suspected, the following are recommended:

- Kidney function tests

Any evaluation should include a careful history of past and present symptoms with an exam. Medical tests that look for damage already done are not a substitute for controlling exposure.

Request copies of your medical testing. You have a legal right to this information under the OSHA Access to Employee Exposure and Medical Records Standard (29 CFR 1910.1020).

Health Hazard Information

Acute Health Effects

The following acute (short-term) health effects may occur immediately or shortly after exposure to o-Phenylphenol:

- o-Phenylphenol can cause skin irritation.
- Contact can severely irritate and burn the eyes with possible permanent damage.
- Inhaling o-Phenylphenol can irritate the nose and throat causing coughing and wheezing.

Chronic Health Effects

The following chronic (long-term) health effects can occur at some time after exposure to o-Phenylphenol and can last for months or years:

Cancer Hazard

- While o-Phenylphenol has been tested, it is not classifiable as to its potential to cause cancer.

Reproductive Hazard

- While o-Phenylphenol has been tested, it is not classifiable as to its potential to cause reproductive harm.
Workplace Controls and Practices

Very toxic chemicals, or those that are reproductive hazards or sensitizers, require expert advice on control measures if a less toxic chemical cannot be substituted. Control measures include: (1) enclosing chemical processes for severely irritating and corrosive chemicals, (2) using local exhaust ventilation for chemicals that may be harmful with a single exposure, and (3) using general ventilation to control exposures to skin and eye irritants. For further information on workplace controls, consult the NIOSH document on Control Banding at [www.cdc.gov/niOSH/topics/crtrlbanding/](http://www.cdc.gov/niOSH/topics/crtrlbanding/).

The following work practices are also recommended:

- Label process containers.
- Provide employees with hazard information and training.
- Monitor airborne chemical concentrations.
- Use engineering controls if concentrations exceed recommended exposure levels.
- Provide eye wash fountains and emergency showers.
- Wash or shower if skin comes in contact with a hazardous material.
- Always wash at the end of the workshift.
- Change into clean clothing if clothing becomes contaminated.
- Do not take contaminated clothing home.
- Get special training to wash contaminated clothing.
- Do not eat, smoke, or drink in areas where chemicals are being handled, processed or stored.
- Wash hands carefully before eating, smoking, drinking, applying cosmetics or using the toilet.

In addition, the following may be useful or required:

- Use a vacuum, or a wet method using Alcohol, to reduce dust during clean-up. DO NOT DRY SWEEP.

Personal Protective Equipment

The OSHA Personal Protective Equipment Standard (29 CFR 1910.132) requires employers to determine the appropriate personal protective equipment for each hazard and to train employees on how and when to use protective equipment.

The following recommendations are only guidelines and may not apply to every situation.

Gloves and Clothing

- Avoid skin contact with o-Phenylphenol. Wear personal protective equipment made from material which can not be permeated or degraded by this substance. Safety equipment suppliers and manufacturers can provide recommendations on the most protective glove and clothing material for your operation.
- Safety equipment manufacturers recommend Butyl and Viton as glove materials for Hydroxyl compounds, aromatic, and Tyvek®, or the equivalent, as protective materials for clothing.
- All protective clothing (suits, gloves, footwear, headgear) should be clean, available each day, and put on before work.

Eye Protection

- Wear eye protection with side shields or goggles.
- If additional protection is needed for the entire face, use in combination with a face shield. A face shield should not be used without another type of eye protection.

Respiratory Protection

**Improper use of respirators is dangerous.** Respirators should only be used if the employer has implemented a written program that takes into account workplace conditions, requirements for worker training, respirator fit testing, and medical exams, as described in the OSHA Respiratory Protection Standard (29 CFR 1910.134).

- At any detectable concentration, use a NIOSH approved negative pressure, air-purifying, particulate filter respirator with an N, R or P95 filter. More protection is provided by a full facepiece respirator than by a half-mask respirator, and even greater protection is provided by a powered-air purifying respirator.
- Leave the area immediately if (1) while wearing a filter or cartridge respirator you can smell, taste, or otherwise detect o-Phenylphenol, (2) while wearing particulate filters abnormal resistance to breathing is experienced, or (3) eye irritation occurs while wearing a full facepiece respirator. Check to make sure the respirator-to-face seal is still good. If it is, replace the filter or cartridge. If the seal is no longer good, you may need a new respirator.
- Consider all potential sources of exposure in your workplace. You may need a combination of filters, prefilters or cartridges to protect against different forms of a chemical (such as vapor and mist) or against a mixture of chemicals.
- Where the potential for high exposure exists, use a NIOSH approved supplied-air respirator with a full facepiece operated in a pressure-demand or other positive-pressure mode. For increased protection use in combination with an auxiliary self-contained breathing apparatus or an emergency escape air cylinder.

Fire Hazards

If employees are expected to fight fires, they must be trained and equipped as stated in the OSHA Fire Brigades Standard (29 CFR 1910.156).

- o-Phenylphenol is a COMBUSTIBLE SOLID.
- Use dry chemical, CO₂, water spray or alcohol-resistant foam as extinguishing agents.
- POISONOUS GASES ARE PRODUCED IN FIRE.
- Use water spray to keep fire-exposed containers cool.
- o-Phenylphenol, powder or granules may form an ignitable vapor/air mixture in closed tanks or containers.
Spills and Emergencies

If employees are required to clean-up spills, they must be properly trained and equipped. The OSHA Hazardous Waste Operations and Emergency Response Standard (29 CFR 1910.120) may apply.

If o-Phenylphenol is spilled, take the following steps:

▶ Evacuate personnel and secure and control entrance to the area.
▶ Eliminate all ignition sources.
▶ Moisten spilled material with Alcohol first, or use a HEPA-filter vacuum for clean-up, and place into sealed containers for disposal.
▶ Ventilate and wash area with Alcohol and then with a strong soap and water solution.
▶ DO NOT wash into sewer.
▶ It may be necessary to contain and dispose of o-Phenylphenol as a HAZARDOUS WASTE. Contact your state Department of Environmental Protection (DEP) or your regional office of the federal Environmental Protection Agency (EPA) for specific recommendations.

Handling and Storage

Prior to working with o-Phenylphenol you should be trained on its proper handling and storage.

▶ o-Phenylphenol is not compatible with OXIDIZING AGENTS (such as PERCHLORATES, PEROXIDES, PERMANGANATES, CHLORATES, NITRATES, CHLORINE, BROMINE and FLUORINE) and STRONG BASES (such as SODIUM HYDROXIDE and POTASSIUM HYDROXIDE).
▶ Store under refrigeration in tightly closed containers in a cool, well-ventilated area.
▶ Sources of ignition, such as smoking and open flames, are prohibited where o-Phenylphenol is used, handled, or stored in a manner that could create a potential fire or explosion hazard.

Occupational Health Information Resources

The New Jersey Department of Health offers multiple services in occupational health. These services include providing informational resources, educational materials, public presentations, and industrial hygiene and medical investigations and evaluations.

For more information, please contact:

New Jersey Department of Health
Right to Know
PO Box 368
Trenton, NJ 08625-0368
Phone: 609-984-2202
Fax: 609-984-7407
E-mail: rtk@doh.state.nj.us
Web address: http://www.nj.gov/health/eoh/rtkweb

The Right to Know Hazardous Substance Fact Sheets are not intended to be copied and sold for commercial purposes.
GLOSSARY

**ACGIH** is the American Conference of Governmental Industrial Hygienists. They publish guidelines called Threshold Limit Values (TLVs) for exposure to workplace chemicals.

**Acute Exposure Guideline Levels (AEGLs)** are established by the EPA. They describe the risk to humans resulting from once-in-a lifetime, or rare, exposure to airborne chemicals.

**Boiling point** is the temperature at which a substance can change its physical state from a liquid to a gas.

A **carcinogen** is a substance that causes cancer.

The **CAS number** is unique, identifying number, assigned by the Chemical Abstracts Service, to a specific chemical.

**CFR** is the Code of Federal Regulations, which are the regulations of the United States government.

A **combustible** substance is a solid, liquid or gas that will burn.

A **corrosive** substance is a gas, liquid or solid that causes destruction of human skin or severe corrosion of containers.

The **critical temperature** is the temperature above which a gas cannot be liquefied, regardless of the pressure applied.

**DEP** is the New Jersey Department of Environmental Protection.

**DOT** is the Department of Transportation, the federal agency that regulates the transportation of chemicals.

**EPA** is the Environmental Protection Agency, the federal agency responsible for regulating environmental hazards.

**ERG** is the Emergency Response Guidebook. It is a guide for emergency responders for transportation emergencies involving hazardous substances.

**Emergency Response Planning Guideline (ERPG) values** provide estimates of concentration ranges where one reasonably might anticipate observing adverse effects.

A **fetus** is an unborn human or animal.

A **flammable** substance is a solid, liquid, vapor or gas that will ignite easily and burn rapidly.

The **flash point** is the temperature at which a liquid or solid gives off vapor that can form a flammable mixture with air.

**IARC** is the International Agency for Research on Cancer, a scientific group.

**Ionization Potential** is the amount of energy needed to remove an electron from an atom or molecule. It is measured in electron volts.

**IRIS** is the Integrated Risk Information System database on human health effects that may result from exposure to various chemicals, maintained by federal EPA.

**LEL or Lower Explosive Limit**, is the lowest concentration of a combustible substance (gas or vapor) in the air capable of continuing an explosion.

**mg/m³** means milligrams of a chemical in a cubic meter of air. It is a measure of concentration (weight/volume).

A **mutagen** is a substance that causes mutations. A **mutation** is a change in the genetic material in a body cell. Mutations can lead to birth defects, miscarriages, or cancer.

**NFPA** is the National Fire Protection Association. It classifies substances according to their fire and explosion hazard.

**NIOSH** is the National Institute for Occupational Safety and Health. It tests equipment, evaluates and approves respirators, conducts studies of workplace hazards, and proposes standards to OSHA.

**NTP** is the National Toxicology Program which tests chemicals and reviews evidence for cancer.

**OSHA** is the federal Occupational Safety and Health Administration, which adopts and enforces health and safety standards.

**PEOSH** is the New Jersey Public Employees Occupational Safety and Health Act, which adopts and enforces health and safety standards in public workplaces.

**Permeated** is the movement of chemicals through protective materials.

**ppm** means parts of a substance per million parts of air. It is a measure of concentration by volume in air.

**Protective Action Criteria (PAC)** are values established by the Department of Energy and are based on AEGLs and ERPGs. They are used for emergency planning of chemical release events.

A **reactive** substance is a solid, liquid or gas that releases energy under certain conditions.

**STEL** is a Short Term Exposure Limit which is usually a 15-minute exposure that should not be exceeded at any time during a work day.

A **teratogen** is a substance that causes birth defects by damaging the fetus.

**UEL or Upper Explosive Limit** is the highest concentration in air above which there is too much fuel (gas or vapor) to begin a reaction or explosion.

**Vapor Density** is the ratio of the weight of a given volume of one gas to the weight of another (usually Air), at the same temperature and pressure.

The **vapor pressure** is a force exerted by the vapor in equilibrium with the solid or liquid phase of the same substance. The higher the vapor pressure the higher concentration of the substance in air.
**Common Name:** o-PHENYLPHENOL

**Synonyms:** 2-Biphenylol; 2-Hydroxydiphenyl; 2-Phenylphenol

**CAS No:** 90-43-7

**Molecular Formula:** C₁₂H₁₀O

**RTK Substance No:** 1439

**Description:** White, buff, to light lavender, crystalline solid

### HAZARD DATA

<table>
<thead>
<tr>
<th>Hazard Rating</th>
<th>Firefighting</th>
<th>Reactivity</th>
</tr>
</thead>
<tbody>
<tr>
<td>3 - Health</td>
<td>COMBUSTIBLE SOLID</td>
<td>o-Phenylphenol is not compatible with OXIDIZING AGENTS (such as PERCHLORATES, PEROXIDES, PERMANGANATES, CHLORATES, NITRATES, CHLORINE, BROMINE and FLUORINE) and STRONG BASES (such as SODIUM HYDROXIDE and POTASSIUM HYDROXIDE).</td>
</tr>
<tr>
<td>1 - Fire</td>
<td>Use dry chemical, CO₂, water spray or alcohol-resistant foam as extinguishing agents. POISONOUS GASES ARE PRODUCED IN FIRE. Use water spray to keep fire-exposed containers cool. o-Phenylphenol in powder or granular form may form an ignitable vapor/air mixture in closed tanks or containers.</td>
<td></td>
</tr>
<tr>
<td>0 - Reactivity</td>
<td></td>
<td></td>
</tr>
</tbody>
</table>

**DOT#:** None  
**ERG Guide #:** None  
**Hazard Class:** None

### PHYSICAL PROPERTIES

- **Flash Point:** 255°F (124°C)
- **Auto Ignition Temp:** 986°F (530°C)
- **Vapor Pressure:** 1 mm Hg at 212°F (100°C)
- **Specific Gravity:** 1.2 (water = 1)
- **Water Solubility:** Very slightly soluble
- **Boiling Point:** 527° to 547°F (275° to 286°C)
- **Melting Point:** 132° to 135°F (56° to 57°C)
- **pH:** 11.2 to 11.6 (1% solution)
- **Molecular Weight:** 170.2

### EXPOSURE LIMITS

No occupational exposure limits have been established for o-Phenylphenol.

The Protective Action Criteria values are:
- PAC-1 = 75 mg/m³
- PAC-2 = 500 mg/m³
- PAC-3 = 500 mg/m³

### PROTECTIVE EQUIPMENT

- **Gloves:** Butyl and Viton (>8-hr breakthrough for Hydroxyl compounds, aromatic)
- **Coveralls:** Tyvek®
- **Respirator:** Small Spill: Full facepiece APR with High efficiency filters  
Large Spill or Fire: SCBA

### SPILL/LEAKS

**Isolation Distance:** Spill: 25 meters (75 feet)

Moisten spilled material with Alcohol first, or use a HEPA-filter vacuum for clean-up, and place into sealed containers for disposal.

Wash area with Alcohol and then with a strong soap and water solution.

DO NOT wash into sewer.

o-Phenylphenol is toxic to aquatic organisms.

### HEALTH EFFECTS

- **Eyes:** Irritation
- **Skin:** Irritation
- **Inhalation:** Nose and throat irritation with coughing and wheezing

### FIRST AID AND DECONTAMINATION

**Remove** the person from exposure.

**Flush** eyes with large amounts of water for at least 30 minutes. Remove contact lenses if worn. Seek medical attention immediately.

**Quickly** remove contaminated clothing and wash contaminated skin with large amounts of soap and water. Seek medical attention.

**Begin** artificial respiration if breathing has stopped and CPR if necessary.

**Transfer** promptly to a medical facility.