Common Name: OZONE

CAS Number: 10028-15-6
DOT Number: None

HAZARD SUMMARY
* Ozone can affect you when breathed in.
* Ozone may cause mutations. Handle with extreme caution.
* Ozone can cause reproductive damage. Handle with extreme caution.
* Liquefied Ozone, on contact with skin or eyes, can cause irritation and severe burns.
* Breathing Ozone can irritate the nose and throat.
* Higher exposure to Ozone can cause headache, upset stomach, vomiting, and pain or tightness in the chest.
* Breathing Ozone can irritate the lungs causing coughing and/or shortness of breath. Higher exposures can cause a build-up of fluid in the lungs (pulmonary edema), a medical emergency, with severe shortness of breath.
* Repeated exposure may cause lung damage.

IDENTIFICATION
Ozone is a colorless gas with a pungent odor. It is used for purifying air and drinking water, in industrial waste treatment, oils, bleaching and waxes, and to make other chemicals.

REASON FOR CITATION
* Ozone is on the Hazardous Substance List because it is regulated by OSHA and cited by ACGIH, NIOSH, DEP and EPA.
* Definitions are provided on page 5.

HOW TO DETERMINE IF YOU ARE BEING EXPOSED
The New Jersey Right to Know Act requires most employers to label chemicals in the workplace and requires public employers to provide their employees with information and training concerning chemical hazards and controls. The federal OSHA Hazard Communication Standard, 1910.1200, requires private employers to provide similar training and information to their employees.

* Exposure to hazardous substances should be routinely evaluated. This may include collecting personal and area air samples. You can obtain copies of sampling results from your employer. You have a legal right to this information under OSHA 1910.1020.

WORKPLACE EXPOSURE LIMITS
OSHA: The legal airborne permissible exposure limit (PEL) is 0.1 ppm averaged over an 8-hour workshift.

NIOSH: The recommended airborne exposure limit is 0.1 ppm, which should not be exceeded at any time.

ACGIH: The recommended airborne exposure limits are for heavy work, 0.05 ppm; moderate work, 0.08 ppm; light work, 0.1 ppm; and workloads of less than 2 hours, 0.20 ppm; averaged over an 8-hour workshift.

* Ozone may cause mutations. All contact with this chemical should be reduced to the lowest possible level.

WAYS OF REDUCING EXPOSURE
* Where possible, enclose operations and use local exhaust ventilation at the site of chemical release. If local exhaust ventilation or enclosure is not used, respirators should be worn.
* Wear protective work clothing.
* Wash thoroughly immediately after exposure to Ozone.
* Post hazard and warning information in the work area. In addition, as part of an ongoing education and training effort, communicate all information on the health and safety hazards of Ozone to potentially exposed workers.
This Fact Sheet is a summary source of information of all potential and most severe health hazards that may result from exposure. Duration of exposure, concentration of the substance and other factors will affect your susceptibility to any of the potential effects described below.

----------------------------------------

HEALTH HAZARD INFORMATION

Acute Health Effects
The following acute (short-term) health effects may occur immediately or shortly after exposure to Ozone:

* **Liquefied Ozone**, on contact with skin or eyes, can cause irritation and severe burns.
* Breathing **Ozone** can irritate the nose and throat.
* Higher exposure to **Ozone** can cause headache, upset stomach, vomiting, and pain or tightness in the chest.
* Breathing **Ozone** can irritate the lungs causing coughing and/or shortness of breath. Higher exposures can cause a build-up of fluid in the lungs (pulmonary edema), a medical emergency, with severe shortness of breath.

Chronic Health Effects
The following chronic (long-term) health effects can occur at some time after exposure to **Ozone** and can last for months or years:

Cancer Hazard
* There is limited evidence that **Ozone** causes cancer in animals. It may cause cancer of the lungs.
* **Ozone** may cause mutations (genetic changes).

Reproductive Hazard
* **Ozone** may damage the developing fetus.

Other Long-Term Effects
* Repeated exposure may cause lung damage.

MEDICAL

Medical Testing
For those with frequent or potentially high exposure (half the PEL or greater), the following are recommended before beginning work and at regular times after that:

* Pulmonary function tests.
* Consider chest x-ray after acute overexposure.

Any evaluation should include a careful history of past and present symptoms with an exam. Medical tests that look for damage already done are not a substitute for controlling exposure.

Request copies of your medical testing. You have a legal right to this information under OSHA 1910.1020.

Mixed Exposures
* Because smoking can cause heart disease, as well as lung cancer, emphysema, and other respiratory problems, it may worsen respiratory conditions caused by chemical exposure. Even if you have smoked for a long time, stopping now will reduce your risk of developing health problems.

WORKPLACE CONTROLS AND PRACTICES

Unless a less toxic chemical can be substituted for a hazardous substance, **ENGINEERING CONTROLS** are the most effective way of reducing exposure. The best protection is to enclose operations and/or provide local exhaust ventilation at the site of chemical release. Isolating operations can also reduce exposure. Using respirators or protective equipment is less effective than the controls mentioned above, but is sometimes necessary.

In evaluating the controls present in your workplace, consider: (1) how hazardous the substance is, (2) how much of the substance is released into the workplace and (3) whether harmful skin or eye contact could occur. Special controls should be in place for highly toxic chemicals or when significant skin, eye, or breathing exposures are possible.

In addition, the following control is recommended:

* Where possible, automatically transfer **Ozone** from cylinders or other storage containers to process containers.

Good **WORK PRACTICES** can help to reduce hazardous exposures. The following work practices are recommended:

* Workers whose clothing has been contaminated by *liquefied Ozone* should change into clean clothing promptly.
* Contaminated work clothes should be laundered by individuals who have been informed of the hazards of exposure to **Ozone**.
* Eye wash fountains should be provided in the immediate work area for emergency use.
* If there is the possibility of skin exposure, emergency shower facilities should be provided.
* On skin contact with *liquefied Ozone*, immediately wash or shower to remove the chemical.

PERSONAL PROTECTIVE EQUIPMENT

**WORKPLACE CONTROLS ARE BETTER THAN PERSONAL PROTECTIVE EQUIPMENT.** However, for some jobs (such as outside work, confined space entry, jobs done only once in a while, or jobs done while workplace controls are being installed), personal protective equipment may be appropriate.

OSHA 1910.132 requires employers to determine the appropriate personal protective equipment for each hazard and to train employees on how and when to use protective equipment.
The following recommendations are only guidelines and may not apply to every situation.

**Clothing**
- Avoid skin contact with Ozone. Wear protective gloves and clothing. Safety equipment suppliers/manufacturers can provide recommendations on the most protective glove/clothing material for your operation.
- All protective clothing (suits, gloves, footwear, headgear) should be clean, available each day, and put on before work.
- Non-absorbent materials are recommended.

**Eye Protection**
- Wear non-vented, impact resistant goggles when working with fumes, gases, or vapors.
- For liquefied Ozone, wear indirect-vent, impact and splash resistant goggles when working with liquids.
- Wear a face shield along with goggles when working with corrosive, highly irritating or toxic substances.

**Respiratory Protection**

**IMPROPER USE OF RESPIRATORS IS DANGEROUS.**
Such equipment should only be used if the employer has a written program that takes into account workplace conditions, requirements for worker training, respirator fit testing and medical exams, as described in OSHA 1910.134.

- Where the potential exists for exposure over 0.05 ppm, use a NIOSH approved full facepiece respirator with a gas cartridge which is specifically approved for Ozone. Increased protection is obtained from full facepiece powered-air purifying respirators.
- Exposure to 5 ppm is immediately dangerous to life and health. If the possibility of exposure above 5 ppm exists, use a NIOSH approved self-contained breathing apparatus with a full facepiece operated in a pressure-demand or other positive-pressure mode.

**HANDLING AND STORAGE**

- Prior to working with Ozone you should be trained on its proper handling and storage.
- Ozone must be stored to avoid contact with ORGANICS; COMBUSTIBLES (such as WOOD, PAPER and OIL); ALKENES; DIETHYL ETHER; NITROGEN COMPOUNDS; OXIDIZING AGENTS (such as PERCHLORATES, PEROXIDES, PERMANGANATES, CHLORATES, NITRATES, CHLORINE, BROMINE and FLUORINE); ORGANICS and AROMATICS (such as BENZENE, ANILINE and ETHYLENE); and STRONG ACIDS (such as HYDROCHLORIC, SULFURIC and NITRIC) since violent reactions occur.
- Liquefied Ozone is a SEVERE EXPLOSION HAZARD when shocked or exposed to heat or flame.
- Store in a cool, dry, well-ventilated place away from STRONG BASES (such as SODIUM HYDROXIDE and POTASSIUM HYDROXIDE) and ORGANIC MATERIALS.

**QUESTIONS AND ANSWERS**

Q: If I have acute health effects, will I later get chronic health effects?
A: Not always. Most chronic (long-term) effects result from repeated exposures to a chemical.

Q: Can I get long-term effects without ever having short-term effects?
A: Yes, because long-term effects can occur from repeated exposures to a chemical at levels not high enough to make you immediately sick.

Q: What are my chances of getting sick when I have been exposed to chemicals?
A: The likelihood of becoming sick from chemicals is increased as the amount of exposure increases. This is determined by the length of time and the amount of material to which someone is exposed.

Q: When are higher exposures more likely?
A: Conditions which increase risk of exposure include physical and mechanical processes (heating, pouring, spraying, spills and evaporation from large surface areas such as open containers), and "confined space" exposures (working inside vats, reactors, boilers, small rooms, etc.).

Q: Is the risk of getting sick higher for workers than for community residents?
A: Yes. Exposures in the community, except possibly in cases of fires or spills, are usually much lower than those found in the workplace. However, people in the community may be exposed to contaminated water as well as to chemicals in the air over long periods. This may be a problem for children or people who are already ill.

Q: What are the likely health problems from chemicals which cause mutations?
A: There are two primary health concerns associated with mutagens: (1) cancers can result from changes induced in cells and, (2) adverse reproductive and developmental outcomes can result from damage to the egg and sperm cells.

Q: Can men as well as women be affected by chemicals that cause reproductive system damage?
A: Yes. Some chemicals reduce potency or fertility in both men and women. Some damage sperm and eggs, possibly leading to birth defects.

Q: Who is at the greatest risk from reproductive hazards?
A: Pregnant women are at greatest risk from chemicals that harm the developing fetus. However, chemicals may affect the ability to have children, so both men and women of childbearing age are at high risk.
The following information is available from:

New Jersey Department of Health and Senior Services
Occupational Health Service
PO Box 360
Trenton, NJ 08625-0360
(609) 984-1863
(609) 984-7407 (fax)

Web address:   http://www.state.nj.us/health/eho/odisweb/

**Industrial Hygiene Information**
Industrial hygienists are available to answer your questions regarding the control of chemical exposures using exhaust ventilation, special work practices, good housekeeping, good hygiene practices, and personal protective equipment including respirators. In addition, they can help to interpret the results of industrial hygiene survey data.

**Medical Evaluation**
If you think you are becoming sick because of exposure to chemicals at your workplace, you may call personnel at the Department of Health and Senior Services, Occupational Health Service, who can help you find the information you need.

**Public Presentations**
Presentations and educational programs on occupational health or the Right to Know Act can be organized for labor unions, trade associations and other groups.

**Right to Know Information Resources**
The Right to Know Infoline (609) 984-2202 can answer questions about the identity and potential health effects of chemicals, list of educational materials in occupational health, references used to prepare the Fact Sheets, preparation of the Right to Know Survey, education and training programs, labeling requirements, and general information regarding the Right to Know Act. Violations of the law should be reported to (609) 984-2202.
DEFINITIONS

ACGIH is the American Conference of Governmental Industrial Hygienists. It recommends upper limits (called TLVs) for exposure to workplace chemicals.

A **carcinogen** is a substance that causes cancer.

The **CAS number** is assigned by the Chemical Abstracts Service to identify a specific chemical.

A **combustible** substance is a solid, liquid or gas that will burn.

A **corrosive** substance is a gas, liquid or solid that causes irreversible damage to human tissue or containers.

**DEP** is the New Jersey Department of Environmental Protection.

**DOT** is the Department of Transportation, the federal agency that regulates the transportation of chemicals.

**EPA** is the Environmental Protection Agency, the federal agency responsible for regulating environmental hazards.

A **fetus** is an unborn human or animal.

A **flammable** substance is a solid, liquid, vapor or gas that will ignite easily and burn rapidly.

The **flash point** is the temperature at which a liquid or solid gives off vapor that can form a flammable mixture with air.

**HHAG** is the Human Health Assessment Group of the federal EPA.

**IARC** is the International Agency for Research on Cancer, a scientific group that classifies chemicals according to their cancer-causing potential.

A **miscible** substance is a liquid or gas that will evenly dissolve in another.

**mg/m³** means milligrams of a chemical in a cubic meter of air. It is a measure of concentration (weight/volume).

A **mutagen** is a substance that causes mutations. A **mutation** is a change in the genetic material in a body cell. Mutations can lead to birth defects, miscarriages, or cancer.

**NTP** is the National Toxicology Program which tests chemicals and reviews evidence for cancer.

**OSHA** is the Occupational Safety and Health Administration, which adopts and enforces health and safety standards.

**PEL** is the Permissible Exposure Limit which is enforceable by the Occupational Safety and Health Administration.

**PIH** is a DOT designation for chemicals which are Poison Inhalation Hazards.

**ppm** means parts of a substance per million parts of air. It is a measure of concentration by volume in air.

A **reactive** substance is a solid, liquid or gas that releases energy under certain conditions.

A **teratogen** is a substance that causes birth defects by damaging the fetus.

**TLV** is the Threshold Limit Value, the workplace exposure limit recommended by ACGIH.

The **vapor pressure** is a measure of how readily a liquid or a solid mixes with air at its surface. A higher vapor pressure indicates a higher concentration of the substance in air and therefore increases the likelihood of breathing it in.

**NAERG** is the North American Emergency Response Guidebook. It was jointly developed by Transport Canada, the United States Department of Transportation and the Secretariat of Communications and Transportation of Mexico. It is a guide for first responders to quickly identify the specific or generic hazards of material involved in a transportation incident, and to protect themselves and the general public during the initial response phase of the incident.

**NCI** is the National Cancer Institute, a federal agency that determines the cancer-causing potential of chemicals.

**NFPA** is the National Fire Protection Association. It classifies substances according to their fire and explosion hazard.

**NIOSH** is the National Institute for Occupational Safety and Health. It tests equipment, evaluates and approves respirators, conducts studies of workplace hazards, and proposes standards to OSHA.
EMERGENCY INFORMATION

Common Name: **OZONE**  
DOT Number: **None**  
NAERG Code: **No Citation**  
CAS Number: **10028-15-6**

<table>
<thead>
<tr>
<th>Hazard rating</th>
<th>NJDHSS</th>
<th>NFPA</th>
</tr>
</thead>
<tbody>
<tr>
<td>FLAMMABILITY</td>
<td>Not Found</td>
<td>Not Rated</td>
</tr>
<tr>
<td>REACTIVITY</td>
<td>Not Found</td>
<td>Not Rated</td>
</tr>
</tbody>
</table>

CONTAINERS MAY EXPLODE IN FIRE

Hazard Rating Key: 0=minimal; 1=slight; 2=moderate; 3=serious; 4=severe

**FIRE HAZARDS**

* Use dry chemical, CO₂, water spray, or alcohol foam extinguishers.  
* CONTAINERS MAY EXPLODE IN FIRE.  
* Use water spray to keep fire-exposed containers cool.  
* Ozone may ignite or explode on contact with combustibles (wood, paper and oil).  
* If employees are expected to fight fires, they must be trained and equipped as stated in OSHA 1910.156.

**SPILLS AND EMERGENCIES**

If Ozone is leaked, take the following steps:

* Evacuate persons not wearing protective equipment from area of leak until clean-up is complete.  
* Remove all ignition sources.  
* Ventilate area of leak to disperse the gas.  
* Stop flow of gas. If source of leak is a cylinder and the leak cannot be stopped in place, remove the leaking cylinder to a safe place in the open air, and repair leak or allow cylinder to empty.  
* It may be necessary to contain and dispose of Ozone as a HAZARDOUS WASTE. Contact your state Department of Environmental Protection (DEP) or your regional office of the federal Environmental Protection Agency (EPA) for specific recommendations.  
* If employees are required to clean-up spills, they must be properly trained and equipped. OSHA 1910.120(q) may be applicable.

**HANDLING AND STORAGE** (See page 3)

**FIRST AID**

For POISON INFORMATION call 1-800-222-1222

**Eye Contact**
* Immediately flush with large amounts of water for at least 15 minutes, occasionally lifting upper and lower lids. Seek medical attention immediately.

**Skin Contact**
* Immerse affected part in warm water. Seek medical attention.

**Breathing**
* Remove the person from exposure.  
* Begin rescue breathing (using universal precautions) if breathing has stopped and CPR if heart action has stopped.  
* Transfer promptly to a medical facility.  
* Medical observation is recommended for 24 to 48 hours after breathing overexposure, as pulmonary edema may be delayed.

**PHYSICAL DATA**

Vapor Pressure: 1,254 mm Hg at 68°F (20°C)  
Water Solubility: Very slightly soluble

**OTHER COMMONLY USED NAMES**

Chemical Name: Ozone  
Other Names: Oxygen Mol(O₃); Triatomic Oxygen

Not intended to be copied and sold for commercial purposes.

NEW JERSEY DEPARTMENT OF HEALTH AND SENIOR SERVICES
Right to Know Program  
PO Box 368, Trenton, NJ 08625-0368  
(609) 984-2202

FOR LARGE SPILLS AND FIRES immediately call your fire department. You can request emergency information from the following:

CHEMTREC: (800) 424-9300  
NJDEP HOTLINE: 1-877-WARN-DEP

---------------------------------------------------------------------------

Not intended to be copied and sold for commercial purposes.