Common Name: TETRAETHYL LEAD

Synonyms: Tetraethylplumbane; TEL
Chemical Name: Plumbane, Tetraethyl-
Date: March 2002 Revision: November 2007

CAS Number: 78-00-2
RTK Substance Number: 1817
DOT Number: UN 1649

Description and Use
Tetraethyl Lead is a colorless, oily liquid with a sweet, musty odor. It is used as a gasoline additive to prevent “knocking” in motors.

Reasons for Citation
- Tetraethyl Lead is on the Right to Know Hazardous Substance List because it is cited by OSHA, ACGIH, DOT, NIOSH, NTP, DEP, IARC, IRIS and EPA.
- This chemical is on the Special Health Hazard Substance List.

SEE GLOSSARY ON PAGE 5.

FIRST AID

Eye Contact
- Immediately flush with large amounts of water for at least 15 minutes, lifting upper and lower lids. Remove contact lenses, if worn, while rinsing.

Skin Contact
- Remove contaminated clothing. Wash contaminated skin with soap and water.

Inhalation
- Remove the person from exposure.
- Begin rescue breathing (using universal precautions) if breathing has stopped and CPR if heart action has stopped.
- Transfer promptly to a medical facility.

EMERGENCY NUMBERS
Poison Control: 1-800-222-1222
CHEMTREC: 1-800-424-9300
NJDEP Hotline: 1-877-927-6337
National Response Center: 1-800-424-8802

EMERGENCY RESPONDERS >>>>> SEE BACK PAGE

Hazard Summary

<table>
<thead>
<tr>
<th>Hazard Rating</th>
<th>NJDOH</th>
<th>NFPA</th>
</tr>
</thead>
<tbody>
<tr>
<td>HEALTH</td>
<td></td>
<td>3</td>
</tr>
<tr>
<td>FLAMMABILITY</td>
<td></td>
<td>2</td>
</tr>
<tr>
<td>REACTIVITY</td>
<td></td>
<td>2W</td>
</tr>
</tbody>
</table>

Carcinogen
Combustible
Water Reactive
Poisonous gases are produced in fire
Containers may explode in fire

Hazard Rating Key: 0=minimal; 1=slight; 2=moderate; 3=serious; 4=severe

- Tetraethyl Lead can affect you when inhaled or swallowed and may be absorbed through the skin.
- Tetraethyl Lead is a CARCINOGEN. HANDLE WITH EXTREME CAUTION.
- Contact may irritate the skin.
- Contact can irritate the eyes with possible loss of vision.
- Exposure can cause headache, irritability, and muscle and joint pain.
- Repeated exposure can cause Lead poisoning with metallic taste, colic and muscle cramps.
- Tetraethyl Lead may damage the nervous system.
- Exposure may cause kidney and brain damage, and anemia.
- Tetraethyl Lead is REACTIVE and a DANGEROUS EXPLOSION HAZARD.

Workplace Exposure Limits

OSHA: The legal airborne permissible exposure limit (PEL) is 0.075 mg/m³ (as Lead) averaged over an 8-hour workshift.

NIOSH: The recommended airborne exposure limit (REL) is 0.075 mg/m³ (as Lead) averaged over a 10-hour workshift.

ACGIH: The threshold limit value (TLV) is 0.1 mg/m³ (as Lead) averaged over an 8-hour workshift.

- Tetraethyl Lead is a PROBABLE CARCINOGEN in humans. There may be no safe level of exposure to a carcinogen, so all contact should be reduced to the lowest possible level.
- The above exposure limits are for air levels only. When skin contact also occurs, you may be overexposed, even though air levels are less than the limits listed above.
Determining Your Exposure

- Read the product manufacturer’s Material Safety Data Sheet (MSDS) and the label to determine product ingredients and important safety and health information about the product mixture.
- For each individual hazardous ingredient, read the New Jersey Department of Health Hazardous Substance Fact Sheet, available on the RTK website (www.nj.gov/health/eoh/rtkweb) or in your facility’s RTK Central File or Hazard Communication Standard file.
- You have a right to this information under the New Jersey Worker and Community Right to Know Act, the Public Employees Occupational Safety and Health (PEOSH) Act if you are a public worker in New Jersey, and under the federal Occupational Safety and Health Act (OSHA) if you are a private worker.
- The New Jersey Right to Know Act requires most employers to label chemicals in the workplace and requires public employers to provide their employees with information concerning chemical hazards and controls. The federal OSHA Hazard Communication Standard (29 CFR 1910.1200) requires employers to provide similar information and training to their employees.

This Fact Sheet is a summary of available information regarding the health hazards that may result from exposure. Duration of exposure, concentration of the substance and other factors will affect your susceptibility to any of the potential effects described below.

Health Hazard Information

Acute Health Effects
The following acute (short-term) health effects may occur immediately or shortly after exposure to Tetraethyl Lead:

- Contact may irritate the skin.
- Contact can irritate the eyes with possible loss of vision.
- Exposure can cause headache, irritability, reduced memory, disturbed sleep, and mood and personality changes.
- Contact can cause upset stomach, poor appetite, weakness and fatigue.

Chronic Health Effects
The following chronic (long-term) health effects can occur at some time after exposure to Tetraethyl Lead and can last for months or years:

Cancer Hazard
- Tetraethyl Lead is a PROBABLE CARCINOGEN in humans. There is some evidence that Lead compounds cause lung cancer in humans and they have been shown to cause kidney cancer in animals.
- Many scientists believe there is no safe level of exposure to a carcinogen.

Reproductive Hazard
- There is limited evidence that Tetraethyl Lead may damage the male reproductive system (including decreasing the sperm count).

Other Effects
- Repeated exposure to Tetraethyl Lead can cause Lead poisoning. Symptoms include metallic taste, poor appetite, weight loss, colic, nausea, vomiting, and muscle cramps.
- Higher levels can cause muscle and joint pain, and weakness.
- High or repeated exposure may damage the nerves causing weakness, "pins and needles," and poor coordination in the arms and legs.
- Lead exposure increases the risk of high blood pressure.
- Tetraethyl Lead may cause kidney and brain damage, and damage to the blood cells causing anemia.
- Repeated exposure causes Lead to accumulate in the body. It can take years for the body to get rid of excess Lead.

Medical Testing
For frequent or potentially high exposure (half the PEL or greater), the following are recommended before beginning work and at regular times after that:

- Urine test for Lead (should be less than 150 micrograms per deciliter)

If symptoms develop or overexposure is suspected, the following are recommended:

- Exam of the nervous system
- Kidney function tests
- Complete blood count

Any evaluation should include a careful history of past and present symptoms with an exam. Medical tests that look for damage already done are not a substitute for controlling exposure.

Request copies of your medical testing. You have a legal right to this information under the OSHA Access to Employee Exposure and Medical Records Standard (29 CFR 1910.1020).

Mixed Exposures
- Body exposures to Lead from hobbies using Lead solder or pigments, target practice, and drinking moonshine made in Leaded containers will increase Lead levels. Repeated breathing or handling of Leaded gasoline may also add to body Lead levels.
Workplace Controls and Practices

Very toxic chemicals, or those that are reproductive hazards or sensitizers, require expert advice on control measures if a less toxic chemical cannot be substituted. Control measures include: (1) enclosing chemical processes for severely irritating and corrosive chemicals, (2) using local exhaust ventilation for chemicals that may be harmful with a single exposure, and (3) using general ventilation to control exposures to skin and eye irritants. For further information on workplace controls, consult the NIOSH document on Control Banding at www.cdc.gov/niosh/topics/ctrlbanding/.

The following work practices are also recommended:

- Label process containers.
- Provide employees with hazard information and training.
- Monitor airborne chemical concentrations.
- Use engineering controls if concentrations exceed recommended exposure levels.
- Provide eye wash fountains and emergency showers.
- Wash or shower if skin comes in contact with a hazardous material.
- Always wash at the end of the workshift.
- Change into clean clothing if clothing becomes contaminated.
- Do not take contaminated clothing home.
- Get special training to wash contaminated clothing.
- Do not eat, smoke, or drink in areas where chemicals are being handled, processed or stored.
- Wash hands carefully before eating, smoking, drinking, applying cosmetics or using the toilet.

In addition, the following may be useful or required:

- Before entering a confined space where Tetraethyl Lead may be present, check to make sure that an explosive concentration does not exist.

Personal Protective Equipment

The OSHA Personal Protective Equipment Standard (29 CFR 1910.132) requires employers to determine the appropriate personal protective equipment for each hazard and to train employees on how and when to use protective equipment.

The following recommendations are only guidelines and may not apply to every situation.

Gloves and Clothing

- Avoid skin contact with Tetraethyl Lead. Wear personal protective equipment made from material which can not be permeated and/or degraded by this substance. Safety equipment suppliers/manufacturers can provide recommendations on the most protective glove/clothing material for your operation.
- Safety equipment manufacturers recommend DuPont Tychem® CPF-3, BR and LV, and TK as protective materials for clothing.
- All protective clothing (suits, gloves, footwear, headgear) should be clean, available each day, and put on before work.

Eye Protection

- Wear indirect-vent, impact and splash resistant goggles when working with liquids.
- Wear a face shield along with goggles when working with corrosive, highly irritating or toxic substances.
- Do not wear contact lenses when working with this substance.

Respiratory Protection

**Improper use of respirators is dangerous.** Respirators should only be used if the employer has a written program that takes into account workplace conditions, requirements for worker training, respirator fit testing, and medical exams, as described in the OSHA Respiratory Protection Standard (29 CFR 1910.134).

- Where the potential exists for exposure over 0.075 mg/m$^3$ (as Lead), use a NIOSH approved supplied-air respirator with a full facepiece operated in a pressure-demand or other positive-pressure mode. For increased protection use in combination with an auxiliary self-contained breathing apparatus operated in a pressure-demand or other positive-pressure mode.
- Exposure to 40 mg/m$^3$ (as Lead) is immediately dangerous to life and health. If the possibility of exposure above 40 mg/m$^3$ exists, use a NIOSH approved self-contained breathing apparatus with a full facepiece operated in a pressure-demand or other positive-pressure mode equipped with an emergency escape air cylinder.

Fire Hazards

If employees are expected to fight fires, they must be trained and equipped as stated in the OSHA Fire Brigades Standard (29 CFR 1910.156).

- Tetraethyl Lead is a COMBUSTIBLE LIQUID.
- Use dry chemical, CO$_2$, alcohol-resistant foam or other foaming agent as extinguishing agents, as water may not be effective in fighting fires.
- POISONOUS GASES ARE PRODUCED IN FIRE, including Lead Oxides.
- CONTAINERS MAY EXPLODE IN FIRE.
- Use water spray to keep fire-exposed containers cool.
- Vapors may travel to a source of ignition and flash back.
- Vapor is heavier than air and may travel a distance to cause a fire or explosion far from the source.
Spills and Emergencies

If employees are required to clean-up spills, they must be properly trained and equipped. The OSHA Hazardous Waste Operations and Emergency Response Standard (29 CFR 1910.120) may apply.

If **Tetraethyl Lead** is spilled, take the following steps:

- Evacuate personnel and secure and control entrance to the area.
- Eliminate all ignition sources.
- Absorb liquids in vermiculite, dry sand, earth, or a similar material and deposit in sealed containers.
- Ventilate and wash area after clean-up is complete.
- It may be necessary to contain and dispose of **Tetraethyl Lead** as a HAZARDOUS WASTE. Contact your state Department of Environmental Protection (DEP) or your regional office of the federal Environmental Protection Agency (EPA) for specific recommendations.

Handling and Storage

Prior to working with **Tetraethyl Lead**, you should be trained on its proper handling and storage.

- **Tetraethyl Lead** will react with OXIDIZING AGENTS (such as PERCHLORATES, PEROXIDES, PERMANGANATES, CHLORATES, NITRATES, CHLORINE, BROMINE and FLUORINE) and STRONG ACIDS (such as HYDROCHLORIC, SULFURIC and NITRIC) to cause fires and explosions.
- **Tetraethyl Lead** is not compatible with RUST; SULFURYL CHLORIDE; POTASSIUM PERMANGANATE; METALS; METAL OXIDES; and COMBUSTIBLES.
- **Tetraethyl Lead** will attack RUBBER, some PLASTICS, and COATINGS.
- Forms explosive mixtures in air above 200°F (93°C).
- Store in tightly closed containers in a cool, well-ventilated area away from SUNLIGHT as **Tetraethyl Lead** will decompose.
- Sources of ignition, such as smoking and open flames, are prohibited where **Tetraethyl Lead** is used, handled, or stored in a manner that could create a potential fire or explosion hazard.

Occupational Health Information Resources

The New Jersey Department of Health offers multiple services in occupational health. These services include providing informational resources, educational materials, public presentations, and industrial hygiene and medical investigations and evaluations.

For more information, please contact:

New Jersey Department of Health
Right to Know
PO Box 368
Trenton, NJ 08625-0368
Phone: 609-984-2202
Fax: 609-984-7407
E-mail: rtk@doh.state.nj.us
Web address: http://www.nj.gov/health/eoh/rtkweb

*The Right to Know Hazardous Substance Fact Sheets are not intended to be copied and sold for commercial purposes.*
GLOSSARY

ACGIH is the American Conference of Governmental Industrial Hygienists. They publish guidelines called Threshold Limit Values (TLVs) for exposure to workplace chemicals.

Acute Exposure Guideline Levels (AEGLs) are established by the EPA. They describe the risk to humans resulting from once-in-a lifetime, or rare, exposure to airborne chemicals.

Boiling point is the temperature at which a substance can change its physical state from a liquid to a gas.

A carcinogen is a substance that causes cancer.

The CAS number is unique, identifying number, assigned by the Chemical Abstracts Service, to a specific chemical.

CFR is the Code of Federal Regulations, which are the regulations of the United States government.

A combustible substance is a solid, liquid or gas that will burn.

A corrosive substance is a gas, liquid or solid that causes destruction of human skin or severe corrosion of containers.

DEP is the New Jersey Department of Environmental Protection.

DOT is the Department of Transportation, the federal agency that regulates the transportation of chemicals.

EPA is the Environmental Protection Agency, the federal agency responsible for regulating environmental hazards.

ERG is the Emergency Response Guidebook. It is a guide for emergency responders for transportation emergencies involving hazardous substances.

Emergency Response Planning Guideline (ERPG) values are intended to provide estimates of concentration ranges where one reasonably might anticipate observing adverse effects.

A fetus is an unborn human or animal.

A flammable substance is a solid, liquid, vapor or gas that will ignite easily and burn rapidly.

The flash point is the temperature at which a liquid or solid gives off vapor that can form a flammable mixture with air.

IARC is the International Agency for Research on Cancer, a scientific group.

Ionization Potential is the amount of energy needed to remove an electron from an atom or molecule. It is measured in electron volts.

IRIS is the Integrated Risk Information System database maintained by federal EPA. The database contains information on human health effects that may result from exposure to various chemicals in the environment.

LEL or Lower Explosive Limit, is the lowest concentration of a combustible substance (gas or vapor) in the air capable of continuing an explosion.

mg/m³ means milligrams of a chemical in a cubic meter of air. It is a measure of concentration (weight/volume).

A mutagen is a substance that causes mutations. A mutation is a change in the genetic material in a body cell. Mutations can lead to birth defects, miscarriages, or cancer.

NFPA is the National Fire Protection Association. It classifies substances according to their fire and explosion hazard.

NIOSH is the National Institute for Occupational Safety and Health. It tests equipment, evaluates and approves respirators, conducts studies of workplace hazards, and proposes standards to OSHA.

NTP is the National Toxicology Program which tests chemicals and reviews evidence for cancer.

OSHA is the federal Occupational Safety and Health Administration, which adopts and enforces health and safety standards.

PEOSHA is the New Jersey Public Employees Occupational Safety and Health Act, which adopts and enforces health and safety standards in public workplaces.

Permeated is the movement of chemicals through protective materials.

PIH is a DOT designation for chemicals which are Poison Inhalation Hazards.

ppm means parts of a substance per million parts of air. It is a measure of concentration by volume in air.

A reactive substance is a solid, liquid or gas that releases energy under certain conditions.

STEL is a Short Term Exposure Limit which is usually a 15-minute exposure that should not be exceeded at any time during a work day.

A teratogen is a substance that causes birth defects by damaging the fetus.

UEL or Upper Explosive Limit is the highest concentration in air above which there is too much fuel (gas or vapor) to begin a reaction or explosion.

Vapor Density is the ratio of the weight of a given volume of one gas to the weight of another (usually Hydrogen), at the same temperature and pressure.

The vapor pressure is a measure of how readily a liquid or a solid mixes with air at its surface. A higher vapor pressure indicates a higher concentration of the substance in air and therefore increases the likelihood of breathing it in.
Common Name: TETRAETHYL LEAD

Synonyms: Tetraethylplumbane; TEL
CAS No: 78-00-2
Molecular Formula: C₈H₂₀Pb
RTK Substance No: 1817
Description: Colorless, oily liquid with a sweet, musty odor

HAZARD DATA

<table>
<thead>
<tr>
<th>Hazard Rating</th>
<th>Firefighting</th>
<th>Reactivity</th>
</tr>
</thead>
<tbody>
<tr>
<td>3 - Health</td>
<td>Tetraethyl Lead is a COMBUSTIBLE LIQUID. Use dry chemical, CO₂, alcohol-resistant foam or other foaming agent as extinguishing agents, as water may not be effective in fighting fires. POISONOUS GASES ARE PRODUCED IN FIRE, including Lead Oxides. CONTAINERS MAY EXPLODE IN FIRE. Use water spray to keep fire-exposed containers cool. Vapors may travel to a source of ignition and flash back. Vapor is heavier than air and may travel a distance to cause a fire or explosion far from the source.</td>
<td>Tetraethyl Lead will react with OXIDIZING AGENTS (such as PERCHLORATES, PEROXIDES, PERMANGANATES, CHLORATES, NITRATES, CHLORINE, BROMINE and FLUORINE) and STRONG ACIDS (such as HYDROCHLORIC, SULFURIC and NITRIC) to cause fires and explosions. Tetraethyl Lead is not compatible with RUST; SULFURLY CHLORIDE; POTASSIUM PERMANGANATE; METALS; METAL OXIDES; and COMBUSTIBLES. Tetraethyl Lead will attack RUBBER, some PLASTICS, and COATINGS. Forms explosive mixtures in air above 200°F (93°C).</td>
</tr>
<tr>
<td>2 - Fire</td>
<td></td>
<td></td>
</tr>
<tr>
<td>2 - Reactivity</td>
<td></td>
<td></td>
</tr>
</tbody>
</table>

DOT#: UN 1649
ERG Guide #: 131
Hazard Class: 6.1 (Poison)

SPILL/LEAKS

Isolation Distance:
Small spills: 60 meters (200 feet)
Large spills: 270 meters (900 feet)
Absorb liquids in vermiculite, dry sand, earth, or a similar material and deposit in sealed containers. Toxic to aquatic organisms. Hazardous to the environment and persists in the environment.

EXPOSURE LIMITS

<table>
<thead>
<tr>
<th>OSHA</th>
<th>0.075 mg/m³, 8-hr TWA (as Lead)</th>
</tr>
</thead>
<tbody>
<tr>
<td>NIOSH</td>
<td>0.075 mg/m³, 10-hr TWA (as Lead)</td>
</tr>
<tr>
<td>ACGIH</td>
<td>0.1 mg/m³, 8-hr TWA (as Lead)</td>
</tr>
<tr>
<td>IDLH LEVEL</td>
<td>40 mg/m³ (as Lead)</td>
</tr>
</tbody>
</table>

PHYSICAL PROPERTIES

<table>
<thead>
<tr>
<th>Odor</th>
<th>Sweet</th>
</tr>
</thead>
<tbody>
<tr>
<td>Flash Point</td>
<td>200°F (93°C)</td>
</tr>
<tr>
<td>LEL:</td>
<td>1.8%</td>
</tr>
<tr>
<td>Vapor Density</td>
<td>8.6 (air = 1)</td>
</tr>
<tr>
<td>Vapor Pressure</td>
<td>0.2 mm Hg at 68°F (20°C)</td>
</tr>
<tr>
<td>Specific Gravity</td>
<td>1.66 (water = 1)</td>
</tr>
<tr>
<td>Water Solubility</td>
<td>Insoluble</td>
</tr>
<tr>
<td>Boiling Point</td>
<td>228°F (109°C)</td>
</tr>
<tr>
<td>Ionization Potential</td>
<td>11.1 eV</td>
</tr>
<tr>
<td>Molecular Weight</td>
<td>323.5</td>
</tr>
</tbody>
</table>

PROTECTIVE EQUIPMENT

<table>
<thead>
<tr>
<th>Gloves</th>
<th>No information</th>
</tr>
</thead>
<tbody>
<tr>
<td>Coveralls</td>
<td>DuPont Tychem® CPF-3, BR and LV, and TK (&gt;8-hr breakthrough)</td>
</tr>
<tr>
<td>Boots</td>
<td>No information</td>
</tr>
<tr>
<td>Respirator</td>
<td>&gt;0.075 mg/m³ - Supplied air</td>
</tr>
</tbody>
</table>

HEALTH EFFECTS

<table>
<thead>
<tr>
<th>Eyes</th>
<th>Irritation, possible loss of vision</th>
</tr>
</thead>
<tbody>
<tr>
<td>Skin</td>
<td>Irritation</td>
</tr>
<tr>
<td>Acute</td>
<td>Headache, irritability, upset stomach, and weakness</td>
</tr>
<tr>
<td>Chronic</td>
<td>Lead compounds may cause lung cancer in humans Limited evidence of damage to male reproductive system Metallic taste, colic, muscle cramps Damage to the nervous system</td>
</tr>
</tbody>
</table>

FIRST AID AND DECONTAMINATION

Remove the person from exposure.
Flush eyes with large amounts of water for at least 15 minutes. Remove contact lenses if worn.
Remove contaminated clothing and wash contaminated skin with soap and water.
Begin rescue breathing (using universal precautions) if breathing has stopped and CPR if heart action has stopped.
Transfer to a medical facility.

November 2007