

ealth Hazardous Substance Fact Sheet

CAS Number:

RTK Substance Number:

Common Name: DICHLOROFLUOROMETHANE

Synonyms: HCFC-21; CFC 21; Freon® 21; Halon® 112;

Genetron®21

Chemical Name: Methane, Dichlorofluoro-

Date: January 1999 Revision: August 2009

Description and Use

Dichlorofluoromethane is a colorless gas, or a compressed liquefied gas, with a sweet *Ether*-like odor. It is used as a refrigerant, solvent, propellant, and heat-exchange fluid.

Reasons for Citation

▶ Dichlorofluoromethane is on the Right to Know Hazardous Substance List because it is cited by OSHA, ACGIH, DOT, NIOSH, DEP and EPA.

SEE GLOSSARY ON PAGE 5.

FIRST AID

Eye Contact

▶ Immediately flush with large amounts of water for at least 15 minutes, lifting upper and lower lids. Remove contact lenses, if worn, while flushing. Seek medical attention.

Skin Contact

- Quickly remove contaminated clothing. Immediately wash contaminated skin with large amounts of soap and water.
- Immerse affected part in warm water. Seek medical attention.

Inhalation

- ▶ Remove the person from exposure.
- ▶ Begin rescue breathing (using universal precautions) if breathing has stopped and CPR if heart action has stopped.
- ▶ Transfer promptly to a medical facility.

EMERGENCY NUMBERS

Poison Control: 1-800-222-1222 CHEMTREC: 1-800-424-9300 NJDEP Hotline: 1-877-927-6337

National Response Center: 1-800-424-8802

DOT Number: UN 1029

EMERGENCY RESPONDERS >>>> SEE LAST PAGE

75-43-4

3109

Hazard Summary Hazard Rating NJDOH NFPA HEALTH 1 FLAMMABILITY 0 REACTIVITY 0 -

POISONOUS GASES ARE PRODUCED IN FIRE CONTAINERS MAY EXPLODE IN FIRE DOES NOT BURN

Hazard Rating Key: 0=minimal; 1=slight; 2=moderate; 3=serious; 4=severe

- ▶ Dichlorofluoromethane can affect you when inhaled.
- ▶ Contact can irritate the skin and eyes.
- ► Contact with the *liquefied gas* can cause frostbite.
- ► Inhaling **Dichlorofluoromethane** can irritate the nose, throat and lungs.
- ► Exposure to **Dichlorofluoromethane** can cause headache, nausea, vomiting, lightheadedness, tremors, and even unconsciousness and death.
- ► Dichlorofluoromethane may cause abnormal heart rhythm (arrhythmia).
- ▶ Dichlorofluoromethane may damage the liver.

Workplace Exposure Limits

OSHA: The legal airborne permissible exposure limit (PEL) is **1,000 ppm** averaged over an 8-hour workshift.

NIOSH: The recommended airborne exposure limit (REL) is **10 ppm** averaged over a 10-hour workshift.

ACGIH: The threshold limit value (TLV) is **10 ppm** averaged over an 8-hour workshift.

▶ Dichlorofluoromethane, at high concentrations, decreases the amount of available Oxygen. Routinely measure Oxygen content to make sure it is at least 19.5% by volume.

Determining Your Exposure

- ▶ Read the product manufacturer's Material Safety Data Sheet (MSDS) and the label to determine product ingredients and important safety and health information about the product mixture.
- ► For each individual hazardous ingredient, read the New Jersey Department of Health Hazardous Substance Fact Sheet, available on the RTK website (www.nj.gov/health/eoh/rtkweb) or in your facility's RTK Central File or Hazard Communication Standard file.
- ➤ You have a right to this information under the New Jersey Worker and Community Right to Know Act, the Public Employees Occupational Safety and Health (PEOSH) Act if you are a public worker in New Jersey, and under the federal Occupational Safety and Health Act (OSHA) if you are a private worker.
- ▶ The New Jersey Right to Know Act requires most employers to label chemicals in the workplace and requires public employers to provide their employees with information concerning chemical hazards and controls. The federal OSHA Hazard Communication Standard (29 CFR 1910.1200) and the PEOSH Hazard Communication Standard (N.J.A.C. 12:100-7) require employers to provide similar information and training to their employees.

This Fact Sheet is a summary of available information regarding the health hazards that may result from exposure. Duration of exposure, concentration of the substance and other factors will affect your susceptibility to any of the potential effects described below.

Health Hazard Information

Acute Health Effects

The following acute (short-term) health effects may occur immediately or shortly after exposure to **Dichlorofluoromethane**:

Dicinior official children

- ▶ Contact can irritate the skin and eyes.
- ▶ Contact with the *liquefied gas* can cause frostbite.
- ▶ Inhaling **Dichlorofluoromethane** can irritate the nose, throat and lungs causing coughing, wheezing and/or shortness of breath.
- ► Exposure to **Dichlorofluoromethane** can cause headache, dizziness, nausea, vomiting, lightheadedness, confusion, tremors, and even unconsciousness and death.

Chronic Health Effects

The following chronic (long-term) health effects can occur at some time after exposure to **Dichlorofluoromethane** and can last for months or years:

Cancer Hazard

► According to the information presently available to the New Jersey Department of Health, **Dichlorofluoromethane** has not been tested for its ability to cause cancer in animals.

Reproductive Hazard

► There is limited evidence that Dichlorofluoromethane may damage the developing fetus.

Other Effects

- ➤ Dichlorofluoromethane may cause abnormal heart rhythm (arrhythmia).
- ▶ Dichlorofluoromethane may damage the liver.

Medical

Medical Testing

For frequent or potentially high exposure (half the TLV or greater), the following are recommended before beginning work and at regular times after that:

▶ Liver function tests

If symptoms develop or overexposure is suspected, the following is recommended:

► Special 24-48 hours EKG (Holter monitor) to observe and record abnormal heart rhythms

Any evaluation should include a careful history of past and present symptoms with an exam. Medical tests that look for damage already done are <u>not</u> a substitute for controlling exposure.

Request copies of your medical testing. You have a legal right to this information under the OSHA Access to Employee Exposure and Medical Records Standard (29 CFR 1910.1020).

Mixed Exposures

▶ More than light alcohol consumption can cause liver damage. Drinking alcohol can increase the liver damage caused by **Dichlorofluoromethane**.

Workplace Controls and Practices

Very toxic chemicals, or those that are reproductive hazards or sensitizers, require expert advice on control measures if a less toxic chemical cannot be substituted. Control measures include: (1) enclosing chemical processes for severely irritating and corrosive chemicals, (2) using local exhaust ventilation for chemicals that may be harmful with a single exposure, and (3) using general ventilation to control exposures to skin and eye irritants. For further information on workplace controls, consult the NIOSH document on Control Banding at www.cdc.gov/niosh/topics/ctrlbanding/.

The following work practices are also recommended:

- ▶ Label process containers.
- ▶ Provide employees with hazard information and training.
- ▶ Monitor airborne chemical concentrations.
- Use engineering controls if concentrations exceed recommended exposure levels.
- ▶ Provide eye wash fountains and emergency showers.
- ► Wash or shower if skin comes in contact with a hazardous material
- ▶ Always wash at the end of the workshift.
- Change into clean clothing if clothing becomes contaminated.
- ▶ Do not take contaminated clothing home.
- ▶ Get special training to wash contaminated clothing.
- ▶ Do not eat, smoke, or drink in areas where chemicals are being handled, processed or stored.
- Wash hands carefully before eating, smoking, drinking, applying cosmetics or using the toilet.

In addition, the following may be useful or required:

- Specific actions are required for this chemical by OSHA.
 Refer to the OSHA Compressed gases Standard (29 CFR 1910.101).
- ▶ Before entering a confined space where Dichlorofluoromethane is present, check to make sure sufficient Oxygen (19.5%) exists.
- Where possible, transfer Dichlorofluoromethane from cylinders or other containers to process containers in an enclosed system.

Personal Protective Equipment

The OSHA Personal Protective Equipment Standard (29 CFR 1910.132) requires employers to determine the appropriate personal protective equipment for each hazard and to train employees on how and when to use protective equipment.

The following recommendations are only guidelines and may not apply to every situation.

Gloves and Clothing

Avoid skin contact with Dichlorofluoromethane. Wear personal protective equipment made from material which can not be permeated or degraded by this substance. Safety equipment suppliers and manufacturers can provide recommendations on the most protective glove and clothing material for your operation.

- ► Safety equipment manufacturers recommend Neoprene and Barrier® as glove materials, and Tychem®, or the equivalent, as protective clothing materials.
- Where exposure to cold equipment, vapors, or liquid may occur, employees should be provided with insulated gloves and special clothing designed to prevent the freezing of body tissues.
- All protective clothing (suits, gloves, footwear, headgear) should be clean, available each day, and put on before work.

Eye Protection

- ► Wear non-vented, impact resistant goggles when working with fumes, gases, or vapors.
- ▶ If additional protection is needed for the entire face, use in combination with a face shield. A face shield should not be used without another type of eye protection.
- ► Do not wear contact lenses when working with this substance.

Respiratory Protection

Improper use of respirators is dangerous. Respirators should only be used if the employer has implemented a written program that takes into account workplace conditions, requirements for worker training, respirator fit testing, and medical exams, as described in the OSHA Respiratory Protection Standard (29 CFR 1910.134).

- ▶ Where the potential exists for exposure over **10 ppm**, use a NIOSH approved supplied-air respirator with a full facepiece operated in a pressure-demand or other positive-pressure mode. For increased protection use in combination with an auxiliary self-contained breathing apparatus or an emergency escape air cylinder.
- ▶ Exposure to **5,000 ppm** is immediately dangerous to life and health. If the possibility of exposure above **5,000 ppm** exists, use a NIOSH approved self-contained breathing apparatus with a full facepiece operated in a pressuredemand or other positive-pressure mode equipped with an emergency escape air cylinder.

Fire Hazards

If employees are expected to fight fires, they must be trained and equipped as stated in the OSHA Fire Brigades Standard (29 CFR 1910.156).

- ► Extinguish fire using an agent suitable for type of surrounding fire. **Dichlorofluoromethane** itself does not burn
- ▶ POISONOUS GASES ARE PRODUCED IN FIRE, including Hydrogen Fluoride, Carbonyl Fluoride, Hydrogen Chloride, and Phosgene.
- ► CONTAINERS MAY EXPLODE IN FIRE.
- ▶ Use water spray to keep fire-exposed containers cool.

DICHLOROFLUOROMETHANE

Spills and Emergencies

If employees are required to clean-up spills, they must be properly trained and equipped. The OSHA Hazardous Waste Operations and Emergency Response Standard (29 CFR 1910.120) may apply.

If **Dichlorofluoromethane** is leaked, take the following steps:

- Evacuate personnel and secure and control entrance to the area.
- ▶ Eliminate ignition sources.
- ▶ Ventilate area of leak to disperse the gas.
- ▶ Stop flow of gas. If source of leak is a cylinder and the leak cannot be stopped in place, remove the leaking cylinder to a safe place in the open air, and repair leak or allow cylinder to empty.
- ► Turn leaking cylinder with leak up to prevent escape of gas in the liquid state.
- ▶ DO NOT DIRECT water iet on liquid.
- ▶ DO NOT wash into sewer.
- ▶ It may be necessary to contain and dispose of **Dichlorofluoromethane** as a HAZARDOUS WASTE. Contact your state Department of Environmental Protection (DEP) or your regional office of the federal Environmental Protection Agency (EPA) for specific recommendations.

Handling and Storage

Prior to working with **Dichlorofluoromethane** you should be trained on its proper handling and storage.

- ➤ Dichlorofluoromethane reacts with STRONG ACIDS (such as HYDROCHLORIC, SULFURIC and NITRIC) and ACID FUMES to produce toxic and corrosive *Hydrogen Fluoride* and *Hydrogen Chloride gases*.
- ▶ Dichlorofluoromethane is not compatible with POWDERED ALUMINUM; MAGNESIUM; ZINC; SODIUM; POTASSIUM; CALCIUM; and OXIDIZING AGENTS (such as PERCHLORATES, PEROXIDES, PERMANGANATES, CHLORATES, NITRATES, CHLORINE, BROMINE and FLUORINE).
- Store in tightly closed containers in a cool, well-ventilated area.
- Dichlorofluoromethane attacks some forms of PLASTIC, RUBBER and COATINGS.

Occupational Health Information Resources

The New Jersey Department of Health offers multiple services in occupational health. These services include providing informational resources, educational materials, public presentations, and industrial hygiene and medical investigations and evaluations.

For more information, please contact:

New Jersey Department of Health

Right to Know

PO Box 368

Trenton, NJ 08625-0368 Phone: 609-984-2202 Fax: 609-984-7407

E-mail: rtk@doh.state.nj.us

Web address: http://www.nj.gov/health/eoh/rtkweb

The Right to Know Hazardous Substance Fact Sheets are not intended to be copied and sold for commercial purposes.

DICHLOROFLUOROMETHANE

GLOSSARY

ACGIH is the American Conference of Governmental Industrial Hygienists. They publish guidelines called Threshold Limit Values (TLVs) for exposure to workplace chemicals.

Acute Exposure Guideline Levels (AEGLs) are established by the EPA. They describe the risk to humans resulting from once-in-a lifetime, or rare, exposure to airborne chemicals.

Boiling point is the temperature at which a substance can change its physical state from a liquid to a gas.

A **carcinogen** is a substance that causes cancer.

The **CAS number** is unique, identifying number, assigned by the Chemical Abstracts Service, to a specific chemical.

CFR is the Code of Federal Regulations, which are the regulations of the United States government.

A combustible substance is a solid, liquid or gas that will burn.

A **corrosive** substance is a gas, liquid or solid that causes destruction of human skin or severe corrosion of containers.

The **critical temperature** is the temperature above which a gas cannot be liquefied, regardless of the pressure applied.

DEP is the New Jersey Department of Environmental Protection.

DOT is the Department of Transportation, the federal agency that regulates the transportation of chemicals.

EPA is the Environmental Protection Agency, the federal agency responsible for regulating environmental hazards.

ERG is the Emergency Response Guidebook. It is a guide for emergency responders for transportation emergencies involving hazardous substances.

Emergency Response Planning Guideline (ERPG) values provide estimates of concentration ranges where one reasonably might anticipate observing adverse effects.

A fetus is an unborn human or animal.

A **flammable** substance is a solid, liquid, vapor or gas that will ignite easily and burn rapidly.

The **flash point** is the temperature at which a liquid or solid gives off vapor that can form a flammable mixture with air.

IARC is the International Agency for Research on Cancer, a scientific group.

Ionization Potential is the amount of energy needed to remove an electron from an atom or molecule. It is measured in electron volts.

IRIS is the Integrated Risk Information System database on human health effects that may result from exposure to various chemicals, maintained by federal EPA. **LEL** or **Lower Explosive Limit**, is the lowest concentration of a combustible substance (gas or vapor) in the air capable of continuing an explosion.

mg/m³ means milligrams of a chemical in a cubic meter of air. It is a measure of concentration (weight/volume).

A **mutagen** is a substance that causes mutations. A **mutation** is a change in the genetic material in a body cell. Mutations can lead to birth defects, miscarriages, or cancer.

NFPA is the National Fire Protection Association. It classifies substances according to their fire and explosion hazard.

NIOSH is the National Institute for Occupational Safety and Health. It tests equipment, evaluates and approves respirators, conducts studies of workplace hazards, and proposes standards to OSHA.

NTP is the National Toxicology Program which tests chemicals and reviews evidence for cancer.

OSHA is the federal Occupational Safety and Health Administration, which adopts and enforces health and safety standards.

PEOSHA is the New Jersey Public Employees Occupational Safety and Health Act, which adopts and enforces health and safety standards in public workplaces.

Permeated is the movement of chemicals through protective materials.

ppm means parts of a substance per million parts of air. It is a measure of concentration by volume in air.

Protective Action Criteria (PAC) are values established by the Department of Energy and are based on AEGLs and ERPGs. They are used for emergency planning of chemical release events.

A **reactive** substance is a solid, liquid or gas that releases energy under certain conditions.

STEL is a Short Term Exposure Limit which is usually a 15-minute exposure that should not be exceeded at any time during a work day.

A **teratogen** is a substance that causes birth defects by damaging the fetus.

UEL or **Upper Explosive Limit** is the highest concentration in air above which there is too much fuel (gas or vapor) to begin a reaction or explosion.

Vapor Density is the ratio of the weight of a given volume of one gas to the weight of another (usually *Air*), at the same temperature and pressure.

The **vapor pressure** is a force exerted by the vapor in equilibrium with the solid or liquid phase of the same substance. The higher the vapor pressure the higher concentration of the substance in air.



Right to Know Hazardous Substance Fact Sheet



Common Name: DICHLOROFLUOROMETHANE

Synonyms: HCFC-21; CFC 21; Freon® 21; Halon® 112; Genetron® 21

CAS No: 75-43-4

Molecular Formula: CHCl₂F RTK Substance No: 3109

Description: Colorless gas, or a compressed liquefied gas, with a sweet Ether-like odor

HAZARD DATA		
Hazard Rating	Firefighting	Reactivity
1 - Health	Extinguish fire using an agent suitable for type of surrounding fire. Dichlorofluoromethane itself	Dichlorofluoromethane reacts with STRONG ACIDS (such as HYDROCHLORIC, SULFURIC and NITRIC)
0 - Fire	does not burn.	and ACID FUMES to produce toxic and corrosive
0 - Reactivity	POISONOUS GASES ARE PRODUCED IN FIRE,	Hydrogen Fluoride and Hydrogen Chloride gases.
DOT# : UN 1029	including Hydrogen Fluoride, Carbonyl Fluoride, Hydrogen Chloride, and Phosgene. CONTAINERS MAY EXPLODE IN FIRE. Use water spray to keep fire-exposed containers cool.	Dichlorofluoromethane is not compatible with POWDERED ALUMINUM; MAGNESIUM; ZINC; SODIUM; POTASSIUM; CALCIUM; and OXIDIZING AGENTS (such as PERCHLORATES, PEROXIDES, PERMANGANATES, CHLORATES, NITRATES, CHLORINE, BROMINE and FLUORINE).
ERG Guide # : 126		
Hazard Class: 2.2 (Nonflammable)		

SPILL/LEAKS

Isolation Distance:

Spill: 100 meters (330 feet) Fire: 800 meters (1/2 mile)

Stop flow of gas. If source of leak is a cylinder and the leak cannot be stopped in place, remove the leaking cylinder to a safe place in the open air, and repair leak or allow cylinder to empty.

Turn leaking cylinder with leak up to prevent escape of gas in the liquid state.

DO NOT DIRECT water jet on liquid.

DO NOT wash into sewer.

Dichlorofluoromethane is considered to be an Ozone

depleting substance.

EXPOSURE LIMITS

Dichlorofluoromethane is an asphyxiant at high concentrations.

NIOSH: 10 ppm, 10-hr TWA ACGIH: 10 ppm, 8-hr TWA

IDLH: 5,000 ppm

The Protective Action Criteria values are:

PAC-1 = 30 ppm PAC-2 = 100 ppm PAC-3 = 5,000 ppm

PHYSICAL PROPERTIES

Odor Threshold: Ether-like

Flash Point: Nonflammable gas **Auto Ignition Temp:** 972°F (522°C) Vapor Density: 3.8 (air = 1)

Vapor Pressure: 1,193 mm Hg at 70°F (21°C)

Specific Gravity: 1.48 (water = 1)Water Solubility: Slightly soluble **Boiling Point:** 48°F (8.9°C) Freezing Point: -211°F (-135°C) **Ionization Potential:** 12.39 eV Molecular Weight: 102.92

PROTECTIVE EQUIPMENT

Insulated Neoprene (<1-hr breakthrough) and Gloves:

Barrier® (>8-hr breakthrough)

Coveralls: Tychem® Responder® (>8-hr breakthrough)

Respirator: >30 ppm - SCBA

HEALTH EFFECTS

Irritation Eyes:

Skin: Irritation. Contact with liquid causes

Nose, throat and lung irritation with Inhalation:

coughing, wheezing and/or shortness of

breath

Headache, dizziness, lightheadedness, confusion, tremors, unconsciousness

and death

FIRST AID AND DECONTAMINATION

Remove the person from exposure.

Flush eyes with large amounts of water for at least 15 minutes. Remove contact lenses if worn. Seek medical attention.

Quickly remove contaminated clothing and wash contaminated skin with large amounts of soap and water.

Immerse affected part in warm water. Seek medical attention.

Begin artificial respiration if breathing has stopped and CPR if necessary.

Transfer promptly to a medical facility.