

Right to Know Hazardous Substance Fact Sheet

Common Name: HYDROGEN CHLORIDE

Synonyms: Anhydrous Hydrogen Chloride; Muriatic Acid

Chemical Name: Hydrochloric Acid

Date: August 2009 Revision: May 2016

Description and Use

Hydrogen Chloride is a colorless gas, with a pungent odor, that fumes in air. It is often found as a compressed, liquefied gas or in a water solution. It is used to make other chemicals, in metal pickling and cleaning, as a laboratory reagent, and in household cleaners.

► ODOR THRESHOLD = 0.255 to 10.06 ppm

► Odor thresholds vary greatly. Do not rely on odor alone to determine potentially hazardous exposures.

Reasons for Citation

- ► Hydrogen Chloride is on the Right to Know Hazardous Substance List because it is cited by OSHA, ACGIH, DOT, NIOSH, DEP, IARC, IRIS, NFPA and EPA.
- ► This chemical is on the Special Health Hazard Substance List.

SEE GLOSSARY ON PAGE 5.

FIRST AID

Eye Contact

▶ Immediately flush with large amounts of water for at least 30 minutes, lifting upper and lower lids. Remove contact lenses, if worn, while flushing. Seek medical attention immediately.

Skin Contact

- Quickly remove contaminated clothing. Immediately wash contaminated skin with large amounts of water. Seek medical attention.
- Immerse affected part in warm water. Seek medical attention.

Inhalation

- ▶ Remove the person from exposure.
- ▶ Begin rescue breathing (using universal precautions) if breathing has stopped and CPR if heart action has stopped.
- ▶ Medical observation is recommended for 24 to 48 hours after overexposure, as pulmonary edema may be delayed.

EMERGENCY NUMBERS

Poison Control: 1-800-222-1222 CHEMTREC: 1-800-424-9300 NJDEP Hotline: 1-877-927-6337

National Response Center: 1-800-424-8802

CAS Number: 7647-01-0

RTK Substance Number: 1012

DOT Number: UN 1050 (Anhydrous) UN 1789 (Solution)

EMERGENCY RESPONDERS >>> SEE LAST PAGE

Hazard Summary			
Hazard Rating	NJDHSS	NFPA	
HEALTH	-	3	
FLAMMABILITY	-	0	
REACTIVITY	-	1	

CORROSIVE

POISONOUS GASES ARE PRODUCED IN FIRE

Hazard Rating Key: 0=minimal; 1=slight; 2=moderate; 3=serious; 4=severe

- ▶ Hydrogen Chloride can affect you when inhaled.
- ► Hydrogen Chloride is a CORROSIVE CHEMICAL and contact can severely irritate and burn the skin and eyes with possible eye damage.
- ► Contact with liquid Hydrogen Chloride may cause frostbite.
- ▶ Inhaling **Hydrogen Chloride** can irritate the nose and throat.
- ▶ Inhaling **Hydrogen Chloride** can irritate the lungs. Higher exposures may cause a build-up of fluid in the lungs (pulmonary edema), a medical emergency.
- Long-term exposure may cause yellowing and erosion of the teeth
- ▶ Hydrogen Chloride may affect the liver and kidneys.

Workplace Exposure Limits

OSHA: The legal airborne permissible exposure limit (PEL) is **5 ppm**, not to be exceeded at any time.

NIOSH: The recommended airborne exposure limit (REL) is **5 ppm**, which should not be exceeded at any time.

ACGIH: The threshold limit value (TLV) is **2 ppm**, which should not be exceeded at any time.

HYDROGEN CHLORIDE Page 2 of 7

Determining Your Exposure

- ▶ Read the product manufacturer's Material Safety Data Sheet (MSDS) and the label to determine product ingredients and important safety and health information about the product mixture.
- ► For each individual hazardous ingredient, read the New Jersey Department of Health Hazardous Substance Fact Sheet, available on the RTK website (http://nj.gov/health/workplacehealthandsafety/right-to-know) or in your facility's RTK Central File or Hazard Communication Standard file.
- ➤ You have a right to this information under the New Jersey Worker and Community Right to Know Act, the Public Employees Occupational Safety and Health (PEOSH) Act if you are a public worker in New Jersey, and under the federal Occupational Safety and Health Act (OSHA) if you are a private worker.
- ► The New Jersey Right to Know Act requires most employers to label chemicals in the workplace and requires public employers to provide their employees with information concerning chemical hazards and controls. The federal OSHA Hazard Communication Standard (29 CFR 1910.1200) and the PEOSH Hazard Communication Standard (N.J.A.C. 12:100-7) require employers to provide similar information and training to their employees.

This Fact Sheet is a summary of available information regarding the health hazards that may result from exposure. Duration of exposure, concentration of the substance and other factors will affect your susceptibility to any of the potential effects described below.

Health Hazard Information

Acute Health Effects

The following acute (short-term) health effects may occur immediately or shortly after exposure to **Hydrogen Chloride**:

- ► Contact can severely irritate and burn the skin and eyes with possible eye damage.
- ► Contact with liquid **Hydrogen Chloride** may cause frostbite.
- ▶ Inhaling Hydrogen Chloride can irritate the nose and throat.
- ► Inhaling Hydrogen Chloride can irritate the lungs causing coughing and/or shortness of breath. Higher exposures may cause a build-up of fluid in the lungs (pulmonary edema), a medical emergency, with severe shortness of breath.

Chronic Health Effects

The following chronic (long-term) health effects can occur at some time after exposure to **Hydrogen Chloride** and can last for months or years:

Cancer Hazard

► While **Hydrogen Chloride** has been tested, it is not classifiable as to its potential to cause cancer.

Reproductive Hazard

▶ While **Hydrogen Chloride** has been tested, further testing is required to assess its potential to cause reproductive harm.

Other Effects

- ► Hydrogen Chloride can irritate the lungs. Repeated exposure may cause bronchitis to develop with coughing, phlegm, and/or shortness of breath.
- ► Long-term exposure may cause yellowing and erosion of the teeth.
- ▶ Hydrogen Chloride may affect the liver and kidneys.

Medical

Medical Testing

For frequent or potentially high exposure (half the TLV or greater), the following are recommended before beginning work and at regular times after that:

▶ Lung function tests

If symptoms develop or overexposure is suspected, the following are recommended:

- ▶ Consider chest x-ray after acute overexposure
- ▶ Liver and kidney function tests

Any evaluation should include a careful history of past and present symptoms with an exam. Medical tests that look for damage already done are <u>not</u> a substitute for controlling exposure.

Request copies of your medical testing. You have a legal right to this information under the OSHA Access to Employee Exposure and Medical Records Standard (29 CFR 1910.1020).

Mixed Exposures

- ► Smoking can cause heart disease, lung cancer, emphysema, and other respiratory problems. It may worsen respiratory conditions caused by chemical exposure. Even if you have smoked for a long time, stopping now will reduce your risk of developing health problems.
- More than light alcohol consumption can cause liver damage. Drinking alcohol can increase the liver damage caused by Hydrogen Chloride.

HYDROGEN CHLORIDE Page 3 of 7

Workplace Controls and Practices

Very toxic chemicals, or those that are reproductive hazards or sensitizers, require expert advice on control measures if a less toxic chemical cannot be substituted. Control measures include: (1) enclosing chemical processes for severely irritating and corrosive chemicals, (2) using local exhaust ventilation for chemicals that may be harmful with a single exposure, and (3) using general ventilation to control exposures to skin and eye irritants. For further information on workplace controls, consult the NIOSH document on Control Banding at www.cdc.gov/niosh/topics/ctrlbanding/.

The following work practices are also recommended:

- ▶ Label process containers.
- ▶ Provide employees with hazard information and training.
- ▶ Monitor airborne chemical concentrations.
- ► Use engineering controls if concentrations exceed recommended exposure levels.
- ▶ Provide eye wash fountains and emergency showers.
- Wash or shower if skin comes in contact with a hazardous material
- ▶ Always wash at the end of the workshift.
- Change into clean clothing if clothing becomes contaminated.
- ▶ Do not take contaminated clothing home.
- ▶ Get special training to wash contaminated clothing.
- ▶ Do not eat, smoke, or drink in areas where chemicals are being handled, processed or stored.
- Wash hands carefully before eating, smoking, drinking, applying cosmetics or using the toilet.

In addition, the following may be useful or required:

- Specific actions are required for this chemical by OSHA.
 Refer to the OSHA Compressed gases Standard (29 CFR 1910.101).
- Where possible, transfer Hydrogen Chloride from cylinders or other containers to process containers in an enclosed system.

Personal Protective Equipment

The OSHA Personal Protective Equipment Standard (29 CFR 1910.132) requires employers to determine the appropriate personal protective equipment for each hazard and to train employees on how and when to use protective equipment.

The following recommendations are only guidelines and may not apply to every situation.

Gloves and Clothing

- Avoid skin contact with Hydrogen Chloride. Wear personal protective equipment made from material which can not be permeated or degraded by this substance. Safety equipment suppliers and manufacturers can provide recommendations on the most protective glove and clothing material for your operation.
- ► Safety equipment manufacturers recommend Butyl, Neoprene and Viton for gloves and Tychem® SL, BR, Responder® and TK; ONESuit®TEC; and Trellchem® HPS

- and VPS, or the equivalent, as protective materials for clothing.
- ▶ All protective clothing (suits, gloves, footwear, headgear) should be clean, available each day, and put on before work.

Eye Protection

- ► Wear non-vented, impact resistant goggles when working with fumes, gases, or vapors.
- Wear indirect-vent, impact and splash resistant goggles when working with liquids.
- Wear a face shield along with goggles when working with corrosive, highly irritating or toxic substances.

Respiratory Protection

Improper use of respirators is dangerous. Respirators should only be used if the employer has implemented a written program that takes into account workplace conditions, requirements for worker training, respirator fit testing, and medical exams, as described in the OSHA Respiratory Protection Standard (29 CFR 1910.134).

- Where the potential exists for exposure over 2 ppm, use a NIOSH approved full facepiece respirator with an acid gas cartridge which is specifically approved for Hydrogen Chloride. Increased protection is obtained from full facepiece powered-air purifying respirators.
- ▶ Leave the area immediately if (1) while wearing a filter or cartridge respirator you can smell, taste, or otherwise detect **Hydrogen Chloride**, (2) while wearing particulate filters abnormal resistance to breathing is experienced, or (3) eye irritation occurs while wearing a full facepiece respirator. Check to make sure the respirator-to-face seal is still good. If it is, replace the filter or cartridge. If the seal is no longer good, you may need a new respirator.
- ➤ Consider all potential sources of exposure in your workplace. You may need a combination of filters, prefilters or cartridges to protect against different forms of a chemical (such as vapor and mist) or against a mixture of chemicals.
- ▶ Where the potential exists for exposure over **20 ppm**, use a NIOSH approved supplied-air respirator with a full facepiece operated in a pressure-demand or other positive-pressure mode. For increased protection use in combination with an auxiliary self-contained breathing apparatus or an emergency escape air cylinder.
- ▶ Exposure to **50 ppm** is immediately dangerous to life and health. If the possibility of exposure above **50 ppm** exists, use a NIOSH approved self-contained breathing apparatus with a full facepiece operated in a pressure-demand or other positive-pressure mode equipped with an emergency escape air cylinder.

Fire Hazards

If employees are expected to fight fires, they must be trained and equipped as stated in the OSHA Fire Brigades Standard (29 CFR 1910.156).

- ► Extinguish fire using an agent suitable for type of surrounding fire. **Hydrogen Chloride** itself does not burn.
- ► POISONOUS GASES ARE PRODUCED IN FIRE, including Chlorine.
- Use water spray to keep fire-exposed containers cool, but DO NOT get water into containers.

HYDROGEN CHLORIDE Page 4 of 7

Spills and Emergencies

If employees are required to clean-up spills, they must be properly trained and equipped. The OSHA Hazardous Waste Operations and Emergency Response Standard (29 CFR 1910.120) may apply.

If **Hydrogen Chloride** in *solution* is spilled or leaked, take the following steps:

- Evacuate personnel and secure and control entrance to the area.
- ▶ Eliminate all ignition sources.
- ► Cover with dry lime, sand or soda ash and place into sealed containers for disposal.

If **Hydrogen Chloride** *gas* is leaked, take the following steps:

- Evacuate personnel and secure and control entrance to the area.
- ▶ Eliminate ignition sources.
- ▶ Ventilate area of leak to disperse the gas.
- Stop flow of gas. If source of leak is a cylinder and the leak cannot be stopped in place, remove the leaking cylinder to a safe place in the open air, and repair leak or allow cylinder to empty.
- ▶ DO NOT SPRAY water on leaking cylinders.
- ► Turn leaking cylinder with leak up to prevent escape of gas in liquid state.
- ▶ It may be necessary to contain and dispose of **Hydrogen Chloride** as a HAZARDOUS WASTE. Contact your state
 Department of Environmental Protection (DEP) or your
 regional office of the federal Environmental Protection
 Agency (EPA) for specific recommendations.

Handling and Storage

Prior to working with **Hydrogen Chloride** you should be trained on its proper handling and storage.

- ► Hydrogen Chloride may react explosively with ALCOHOLS; HYDROGEN CYANIDE; POTASSIUM PERMANGANATE; SODIUM; and TETRASELENIUM TETRANITRIDE, and may ignite on contact with FLUORINE; HEXALITHIUM DISILICIDE; METAL ACETYLIDES and CARBIDES.
- ▶ Hydrogen Chloride reacts with OXIDIZING AGENTS (such as PERCHLORATES, PEROXIDES, PERMANGANATES, CHLORATES, NITRATES, CHLORINE, BROMINE and FLUORINE) to form toxic Chlorine gas and reacts violently with STRONG BASES (such as SODIUM HYDROXIDE and POTASSIUM HYDROXIDE).
- ► Hydrogen Chloride will attack many METALS (such as COPPER, BRASS and ZINC) to release flammable and explosive Hydrogen gas.
- ► Hydrogen Chloride will react with ALDEHYDES and EPOXIDES to cause violent polymerization (self-reaction).
- ► Store in tightly closed containers in a cool, well-ventilated area away from COMBUSTIBLES.
- ► Hydrogen Chloride corrodes steel.

Occupational Health Information Resources

The New Jersey Department of Health and Occupational Health Service, offers multiple services in occupational health. These services include providing informational resources, educational materials, public presentations, and industrial hygiene and medical investigations and evaluations.

For more information, please contact:

New Jersey Department of Health Right to Know

PO Box 368

Trenton, NJ 08625-0368 Phone: 609-984-2202 Fax: 609-984-7407

E-mail: rtk@doh.nj.gov

Web address:

http://nj.gov/health/workplacehealthandsafety/right-to-

know

The Right to Know Hazardous Substance Fact Sheets are not intended to be copied and sold for commercial purposes.

HYDROGEN CHLORIDE Page 5 of 7

GLOSSARY

ACGIH is the American Conference of Governmental Industrial Hygienists. They publish guidelines called Threshold Limit Values (TLVs) for exposure to workplace chemicals.

Acute Exposure Guideline Levels (AEGLs) are established by the EPA. They describe the risk to humans resulting from once-in-a lifetime, or rare, exposure to airborne chemicals.

Boiling point is the temperature at which a substance can change its physical state from a liquid to a gas.

A **carcinogen** is a substance that causes cancer.

The **CAS number** is unique, identifying number, assigned by the Chemical Abstracts Service, to a specific chemical.

CFR is the Code of Federal Regulations, which are the regulations of the United States government.

A **combustible** substance is a solid, liquid or gas that will burn.

A **corrosive** substance is a gas, liquid or solid that causes destruction of human skin or severe corrosion of containers.

The **critical temperature** is the temperature above which a gas cannot be liquefied, regardless of the pressure applied.

DEP is the New Jersey Department of Environmental Protection.

DOT is the Department of Transportation, the federal agency that regulates the transportation of chemicals.

EPA is the Environmental Protection Agency, the federal agency responsible for regulating environmental hazards.

ERG is the Emergency Response Guidebook. It is a guide for emergency responders for transportation emergencies involving hazardous substances.

Emergency Response Planning Guideline (ERPG) values provide estimates of concentration ranges where one reasonably might anticipate observing adverse effects.

A fetus is an unborn human or animal.

A **flammable** substance is a solid, liquid, vapor or gas that will ignite easily and burn rapidly.

The **flash point** is the temperature at which a liquid or solid gives off vapor that can form a flammable mixture with air.

IARC is the International Agency for Research on Cancer, a scientific group.

Ionization Potential is the amount of energy needed to remove an electron from an atom or molecule. It is measured in electron volts.

IRIS is the Integrated Risk Information System database on human health effects that may result from exposure to various chemicals, maintained by federal EPA.

LEL or **Lower Explosive Limit**, is the lowest concentration of a combustible substance (gas or vapor) in the air capable of continuing an explosion.

mg/m³ means milligrams of a chemical in a cubic meter of air. It is a measure of concentration (weight/volume).

A **mutagen** is a substance that causes mutations. A **mutation** is a change in the genetic material in a body cell. Mutations can lead to birth defects, miscarriages, or cancer.

NFPA is the National Fire Protection Association. It classifies substances according to their fire and explosion hazard.

NIOSH is the National Institute for Occupational Safety and Health. It tests equipment, evaluates and approves respirators, conducts studies of workplace hazards, and proposes standards to OSHA.

NTP is the National Toxicology Program which tests chemicals and reviews evidence for cancer.

OSHA is the federal Occupational Safety and Health Administration, which adopts and enforces health and safety standards.

PEOSHA is the New Jersey Public Employees Occupational Safety and Health Act, which adopts and enforces health and safety standards in public workplaces.

Permeated is the movement of chemicals through protective materials.

ppm means parts of a substance per million parts of air. It is a measure of concentration by volume in air.

Protective Action Criteria (PAC) are values established by the Department of Energy and are based on AEGLs and ERPGs. They are used for emergency planning of chemical release events.

A **reactive** substance is a solid, liquid or gas that releases energy under certain conditions.

STEL is a Short Term Exposure Limit which is usually a 15-minute exposure that should not be exceeded at any time during a work day.

A **teratogen** is a substance that causes birth defects by damaging the fetus.

UEL or **Upper Explosive Limit** is the highest concentration in air above which there is too much fuel (gas or vapor) to begin a reaction or explosion.

Vapor Density is the ratio of the weight of a given volume of one gas to the weight of another (usually *Air*), at the same temperature and pressure.

HYDROGEN CHLORIDE Page 6 of 7

The **vapor pressure** is a force exerted by the vapor in equilibrium with the solid or liquid phase of the same substance. The higher the vapor pressure the higher concentration of the substance in air.



Right to Know Hazardous Substance Fact Sheet



Common Name: HYDROGEN CHLORIDE

Synonyms: Anhydrous Hydrogen Chloride; Muriatic Acid

CAS No: 7647-01-0 Molecular Formula: HCl RTK Substance No: 1012

Description: Colorless gas with a pungent odor that fumes in air, and is often found as a compressed,

liquefied gas or in a water solution

HAZARD DATA			
Hazard Rating	Firefighting	Reactivity	
3 - Health 0 - Fire 1 - Reactivity DOT#:	Extinguish fire using an agent suitable for type of surrounding fire. Hydrogen Chloride itself does not burn. POISONOUS GASES ARE PRODUCED IN FIRE, including	Hydrogen Chloride may react explosively with ALCOHOLS; HYDROGEN CYANIDE; POTASSIUM PERMANGANATE; SODIUM; and TETRASELENIUM TETRANITRIDE, and may ignite on contact with FLUORINE; HEXALITHIUM DISILICIDE; METAL ACETYLIDES and CARBIDES.	
UN 1050 (Anhydrous) UN 1789 (Solution)	Chlorine. Use water spray to keep fire-exposed containers cool, but DO NOT get water into containers.	Hydrogen Chloride reacts with OXIDIZING AGENTS (such as PERCHLORATES, PEROXIDES, PERMANGANATES, CHLORATES, NITRATES, CHLORINE, BROMINE and FLUORINE) to form toxic Chlorine gas and reacts violently with STRONG BASES (such as SODIUM HYDROXIDE and POTASSIUM HYDROXIDE).	
ERG Guide #: 125 (Anhydrous) 157 (Solution)			
Hazard Class: 2.3 (Toxic Gas) (Anhydrous) 8 (Corrosive) (Solution)		Hydrogen Chloride will attack many METALS (such as COPPER, BRASS and ZINC) to release flammable and explosive Hydrogen gas. Hydrogen Chloride will react with ALDEHYDES and EPOXIDES to cause violent polymerization (self-reaction).	

SPILL/LEAKS

Isolation Distance:

Small Spill: 30 meters (100 feet) Large Spill: 60 meters (200 feet) Fire: 800 meters (1/2 mile)

Cover **Hydrogen Chloride** in *solution* with dry lime, sand or soda ash and place into sealed containers for disposal.

Stop flow of gas. If source of leak is a cylinder and the leak cannot be stopped in place, remove the leaking cylinder to a safe place in the open air, and repair leak or allow cylinder to empty.

DO NOT SPRAY water on leaking cylinder.

Turn leaking cylinder with leak up to prevent escape of gas in liquid state.

EXPOSURE LIMITS

OSHA: 5 ppm, Ceiling
NIOSH: 5 ppm, Ceiling
ACGIH: 2 ppm, Ceiling
IDLH: 50 ppm

The Protective Action Criteria values are:

PAC-1 = 1.8 ppm PAC-2 = 22 ppm PAC-3 = 100 ppm

HEALTH EFFECTS

Eyes: Severe irritation, burns and possible eye damage

Skin: Severe irritation and burns

Contact with liquid causes frostbite

Inhalation: Nose, throat and lung irritation with coughing and

severe shortness of breath (pulmonary edema)

PHYSICAL PROPERTIES

Odor Threshold: 0.255 to 10.06 ppm

Flash Point: Nonflammable

Vapor Density: 1.3 (air = 1)

Vapor Pressure: >760 mm Hg at -120°F (-84°C)

Specific Gravity: 1.27 (liquid) (water = 1)

Water Solubility: Soluble

Boiling Point: -121°F (-85°C)

Freezing Point: -174°F (-114°C)

Ionization Potential: 12.74 eV

Molecular Weight: 36.47

PROTECTIVE EQUIPMENT

Gloves: Butyl, Neoprene and Viton (>8-hr breakthrough)

Coveralls: Tychem® BR, Responder® and TK;

ONESuit®TEC; Trellchem® HPS and VPS (>8-

hr breakthrough)

Respirator: >2 ppm - full facepiece APR with *Acid gas* filters

>20 ppm - SCBA

FIRST AID AND DECONTAMINATION

Remove the person from exposure.

Flush eyes with large amounts of water for at least 30 minutes. Remove contact lenses if worn. Seek medical attention immediately.

Quickly remove contaminated clothing and wash contaminated skin with large amounts of water. Seek medical attention.

Immerse affected part in warm water. Seek medical attention.

Begin artificial respiration if breathing has stopped and CPR if necessary.

Transfer promptly to a medical facility.

Medical observation is recommended as symptoms may be delayed.