Common Name: PROPIONIC ACID

Synonyms: Ethylformic Acid; Methylacetic Acid
Chemical Name: Propanoic Acid
Date: April 2001 Revision: April 2010

CAS Number: 79-09-4
RTK Substance Number: 1599
DOT Number: UN 1848

Description and Use

Propionic Acid is a colorless, oily liquid with a strong, unpleasant odor. It is used as a preservative, fungicide, and antimicrobial agent.

- ODOR THRESHOLD = 0.026 to 0.17 ppm
- Odor thresholds vary greatly. Do not rely on odor alone to determine potentially hazardous exposures.

Reasons for Citation

- Propionic Acid is on the Right to Know Hazardous Substance List because it is cited by ACGIH, DOT, NIOSH, NFPA and EPA.
- This chemical is on the Special Health Hazard Substance List.

SEE GLOSSARY ON PAGE 5.

FIRST AID

Eye Contact
- Immediately flush with large amounts of water for at least 30 minutes, lifting upper and lower lids. Remove contact lenses, if worn, while flushing. Seek medical attention.

Skin Contact
- Quickly remove contaminated clothing. Immediately wash contaminated skin with large amounts of water. Seek medical attention.

Inhalation
- Remove the person from exposure.
- Begin rescue breathing (using universal precautions) if breathing has stopped and CPR if heart action has stopped.
- Transfer promptly to a medical facility.

EMERGENCY NUMBERS

Poison Control: 1-800-222-1222
CHEMTREC: 1-800-424-9300
NJDEP Hotline: 1-877-927-6337
National Response Center: 1-800-424-8802

Hazard Summary

<table>
<thead>
<tr>
<th>Hazard Rating</th>
<th>NJDOH</th>
<th>NFPA</th>
</tr>
</thead>
<tbody>
<tr>
<td>HEALTH</td>
<td>-</td>
<td>3</td>
</tr>
<tr>
<td>FLAMMABILITY</td>
<td>-</td>
<td>2</td>
</tr>
<tr>
<td>REACTIVITY</td>
<td>-</td>
<td>0</td>
</tr>
<tr>
<td>CORROSIVE</td>
<td></td>
<td></td>
</tr>
<tr>
<td>COMBUSTIBLE</td>
<td></td>
<td></td>
</tr>
<tr>
<td>POISONOUS GASES ARE PRODUCED IN FIRE</td>
<td></td>
<td></td>
</tr>
</tbody>
</table>

Hazard Rating Key: 0=minimal; 1=slight; 2=moderate; 3=serious; 4=severe

- Propionic Acid can affect you when inhaled and by passing through the skin.
- Propionic Acid is a CORROSIVE CHEMICAL and contact can severely irritate and burn the skin and eyes with possible eye damage.
- Inhaling Propionic Acid can irritate the nose, throat and lungs causing coughing, wheezing and/or shortness of breath.
- Propionic Acid can cause headache, nausea, vomiting, diarrhea and abdominal pain.

Workplace Exposure Limits

NIOSH: The recommended airborne exposure limit (REL) is 10 ppm averaged over a 10-hour workshift and 15 ppm, not to be exceeded during any 15-minute work period.

ACGIH: The threshold limit value (TLV) is 10 ppm averaged over an 8-hour workshift.

- The above exposure limits are for air levels only. When skin contact also occurs, you may be overexposed, even though air levels are less than the limits listed above.
**Determining Your Exposure**

- Read the product manufacturer's Material Safety Data Sheet (MSDS) and the label to determine product ingredients and important safety and health information about the product mixture.
- For each individual hazardous ingredient, read the New Jersey Department of Health Hazardous Substance Fact Sheet, available on the RTK website (www.nj.gov/health/eoh/rtkweb) or in your facility's RTK Central File or Hazard Communication Standard file.
- You have a right to this information under the New Jersey Worker and Community Right to Know Act and the Public Employees Occupational Safety and Health (PEOSH) Act if you are a public worker in New Jersey, and under the federal Occupational Safety and Health Act (OSHA) if you are a private worker.
- The New Jersey Right to Know Act requires most employers to label chemicals in the workplace and requires public employers to provide their employees with information concerning chemical hazards and controls. The federal OSHA Hazard Communication Standard (29 CFR 1910.1200) and the PEOSH Hazard Communication Standard (N.J.A.C. 12:100-7) require employers to provide similar information and training to their employees.

This Fact Sheet is a summary of available information regarding the health hazards that may result from exposure. Duration of exposure, concentration of the substance and other factors will affect your susceptibility to any of the potential effects described below.

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**Health Hazard Information**

**Acute Health Effects**
The following acute (short-term) health effects may occur immediately or shortly after exposure to Propionic Acid:

- Contact can severely irritate and burn the skin and eyes with possible eye damage.
- Inhaling Propionic Acid can irritate the nose, throat and lungs causing coughing, wheezing and/or shortness of breath.
- Propionic Acid can cause headache, nausea, vomiting, diarrhea and abdominal pain.

**Chronic Health Effects**
The following chronic (long-term) health effects can occur at some time after exposure to Propionic Acid and can last for months or years:

**Cancer Hazard**
- While Propionic Acid has been tested, further testing is required to assess its potential to cause cancer.

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**Reproductive Hazard**
- There is no evidence that Propionic Acid affects reproduction. This is based on test results presently available to the NJDOH from published studies.

**Other Effects**
- Propionic Acid can irritate the lungs. Repeated exposure may cause bronchitis to develop with coughing, phlegm, and/or shortness of breath.

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**Medical**

**Medical Testing**
If symptoms develop or overexposure is suspected, the following are recommended:

- **Lung function tests**

Any evaluation should include a careful history of past and present symptoms with an exam. Medical tests that look for damage already done are not a substitute for controlling exposure.

Request copies of your medical testing. You have a legal right to this information under the OSHA Access to Employee Exposure and Medical Records Standard (29 CFR 1910.1020).

**Mixed Exposures**
- Smoking can cause heart disease, lung cancer, emphysema, and other respiratory problems. It may worsen respiratory conditions caused by chemical exposure. Even if you have smoked for a long time, stopping now will reduce your risk of developing health problems.
**Workplace Controls and Practices**

Very toxic chemicals, or those that are reproductive hazards or sensitizers, require expert advice on control measures if a less toxic chemical cannot be substituted. Control measures include: (1) enclosing chemical processes for severely irritating and corrosive chemicals, (2) using local exhaust ventilation for chemicals that may be harmful with a single exposure, and (3) using general ventilation to control exposures to skin and eye irritants. For further information on workplace controls, consult the NIOSH document on Control Banding at [www.cdc.gov/niosh/topics/ctrlbanding/](http://www.cdc.gov/niosh/topics/ctrlbanding/).

The following work practices are also recommended:

- Label process containers.
- Provide employees with hazard information and training.
- Monitor airborne chemical concentrations.
- Use engineering controls if concentrations exceed recommended exposure levels.
- Provide eye wash fountains and emergency showers.
- Wash or shower if skin comes in contact with a hazardous material.
- Always wash at the end of the workshift.
- Change into clean clothing if clothing becomes contaminated.
- Do not take contaminated clothing home.
- Get special training to wash contaminated clothing.
- Do not eat, smoke, or drink in areas where chemicals are being handled, processed or stored.
- Wash hands carefully before eating, smoking, drinking, applying cosmetics or using the toilet.

In addition, the following may be useful or required:

- Where possible, transfer *Propionic Acid* from drums or other containers to process containers in an enclosed system.
- Before entering a confined space where *Propionic Acid* may be present, check to make sure that an explosive concentration does not exist.

**Personal Protective Equipment**

The OSHA Personal Protective Equipment Standard (29 CFR 1910.132) requires employers to determine the appropriate personal protective equipment for each hazard and to train employees on how and when to use protective equipment.

The following recommendations are only guidelines and may not apply to every situation.

**Gloves and Clothing**

- Avoid skin contact with *Propionic Acid*. Wear personal protective equipment made from material which can not be permeated or degraded by this substance. Safety equipment suppliers and manufacturers can provide recommendations on the most protective glove and clothing material for your operation.
- Safety equipment manufacturers recommend Butyl and Teflon® for gloves, and Tychem® Responder®, or the equivalent, as a protective material for clothing.
- All protective clothing (suits, gloves, footwear, headgear) should be clean, available each day, and put on before work.

**Eye Protection**

- Wear indirect-vent, impact and splash resistant goggles when working with liquids.
- Wear non-vented, impact resistant goggles when working with fumes, gases, or vapors.
- Wear a face shield along with goggles when working with corrosive, highly irritating or toxic substances.

**Respiratory Protection**

*Improper use of respirators is dangerous.* Respirators should only be used if the employer has implemented a written program that takes into account workplace conditions, requirements for worker training, respirator fit testing, and medical exams, as described in the OSHA Respiratory Protection Standard (29 CFR 1910.134).

- Where the potential exists for exposure over 10 ppm, use a NIOSH approved full facepiece respirator with an organic vapor cartridge. Increased protection is obtained from full facepiece powered-air purifying respirators.
- Leave the area immediately if (1) while wearing a filter or cartridge respirator you can smell, taste, or otherwise detect *Propionic Acid*, (2) while wearing particulate filters abnormal resistance to breathing is experienced, or (3) eye irritation occurs while wearing a full facepiece respirator. Check to make sure the respirator-to-face seal is still good. If it is, replace the filter or cartridge. If the seal is no longer good, you may need a new respirator.
- Consider all potential sources of exposure in your workplace. You may need a combination of filters, prefilters or cartridges to protect against different forms of a chemical (such as vapor and mist) or against a mixture of chemicals.
- Where the potential exists for exposure over 100 ppm, use a NIOSH approved supplied-air respirator with a full facepiece operated in a pressure-demand or other positive-pressure mode. For increased protection use in combination with an auxiliary self-contained breathing apparatus or an emergency escape air cylinder.

**Fire Hazards**

If employees are expected to fight fires, they must be trained and equipped as stated in the OSHA Fire Brigades Standard (29 CFR 1910.156).

- *Propionic Acid* is a COMBUSTIBLE LIQUID.
- Use dry chemical, CO₂, water spray or alcohol-resistant foam as extinguishing agents.
- POISONOUS GASES ARE PRODUCED IN FIRE.
- Use water spray to keep fire-exposed containers cool.
- Flow or agitation may generate electrostatic charges.
- *Propionic Acid* may form an ignitable vapor/air mixture in closed tanks or containers.
**Spills and Emergencies**

If employees are required to clean-up spills, they must be properly trained and equipped. The OSHA Hazardous Waste Operations and Emergency Response Standard (29 CFR 1910.120) may apply.

If **Propionic Acid** is spilled or leaked, take the following steps:

- Evacuate personnel and secure and control entrance to the area.
- Eliminate all ignition sources.
- Absorb liquids in dry sand, earth, or a similar material and place into sealed containers for disposal.
- Ventilate area of spill or leak.
- Keep **Propionic Acid** out of confined spaces, such as sewers, because of the possibility of an explosion.
- DO NOT wash into sewer.
- It may be necessary to contain and dispose of **Propionic Acid** as a HAZARDOUS WASTE. Contact your state Department of Environmental Protection (DEP) or your regional office of the federal Environmental Protection Agency (EPA) for specific recommendations.

**Handling and Storage**

Prior to working with **Propionic Acid** you should be trained on its proper handling and storage.

- **Propionic Acid** reacts violently and explosively with OXIDIZING AGENTS (such as PERCHLORATES, PEROXIDES, PERMANGANATES, CHLORATES, NITRATES, CHLORINE, BROMINE and FLUORINE) and PHOSPHORUS TRICHLORIDE.
- **Propionic Acid** may react violently with STRONG BASES (such as SODIUM HYDROXIDE and POTASSIUM HYDROXIDE); REDUCING AGENTS (such as LITHIUM, SODIUM, ALUMINUM and their HYDRIDES); and AMINES.
- **Propionic Acid** reacts with POWDERED METALS (such as ALUMINUM and ZINC) to produce flammable and explosive Hydrogen gas.
- Store in tightly closed containers in a cool, well-ventilated area away from METALS as **Propionic Acid** is corrosive to LEAD, STEEL and NICKEL.
- **Propionic Acid** may accumulate static electricity.

**Occupational Health Information Resources**

The New Jersey Department of Health offers multiple services in occupational health. These services include providing informational resources, educational materials, public presentations, and industrial hygiene and medical investigations and evaluations.

For more information, please contact:

New Jersey Department of Health
Right to Know
PO Box 368
Trenton, NJ 08625-0368
Phone: 609-984-2202
Fax: 609-984-7407
E-mail: rtk@doh.state.nj.us
Web address: http://www.nj.gov/health/eoh/rtkweb

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**ACGIH**: is the American Conference of Governmental Industrial Hygienists. They publish guidelines called Threshold Limit Values (TLVs) for exposure to workplace chemicals.

**Acute Exposure Guideline Levels (AEGLs)** are established by the EPA. They describe the risk to humans resulting from once-in-a lifetime, or rare, exposure to airborne chemicals.

**Boiling point** is the temperature at which a substance can change its physical state from a liquid to a gas.

A **carcinogen** is a substance that causes cancer.

The **CAS number** is unique, identifying number, assigned by the Chemical Abstracts Service, to a specific chemical.

**CFR** is the Code of Federal Regulations, which are the regulations of the United States government.

A **combustible** substance is a solid, liquid or gas that will burn.

A **corrosive** substance is a gas, liquid or solid that causes destruction of human skin or severe corrosion of containers.

The **critical temperature** is the temperature above which a gas cannot be liquefied, regardless of the pressure applied.

**DEP** is the New Jersey Department of Environmental Protection.

**DOT** is the Department of Transportation, the federal agency that regulates the transportation of chemicals.

**EPA** is the Environmental Protection Agency, the federal agency responsible for regulating environmental hazards.

**ERG** is the Emergency Response Guidebook. It is a guide for emergency responders for transportation emergencies involving hazardous substances.

**Emergency Response Planning Guideline (ERPG) values** provide estimates of concentration ranges where one reasonably might anticipate observing adverse effects.

A **fetus** is an unborn human or animal.

A **flammable** substance is a solid, liquid, vapor or gas that will ignite easily and burn rapidly.

The **flash point** is the temperature at which a liquid or solid gives off vapor that can form a flammable mixture with air.

**IARC** is the International Agency for Research on Cancer, a scientific group.

**Ionization Potential** is the amount of energy needed to remove an electron from an atom or molecule. It is measured in electron volts.

**IRIS** is the Integrated Risk Information System database on human health effects that may result from exposure to various chemicals, maintained by federal EPA.

**LEL or Lower Explosive Limit**, is the lowest concentration of a combustible substance (gas or vapor) in the air capable of continuing an explosion.

**mg/m³** means milligrams of a chemical in a cubic meter of air. It is a measure of concentration (weight/volume).

A **mutagen** is a substance that causes mutations. A **mutation** is a change in the genetic material in a body cell. Mutations can lead to birth defects, miscarriages, or cancer.

**NFPA** is the National Fire Protection Association. It classifies substances according to their fire and explosion hazard.

**NIOSH** is the National Institute for Occupational Safety and Health. It tests equipment, evaluates and approves respirators, conducts studies of workplace hazards, and proposes standards to OSHA.

**NTP** is the National Toxicology Program which tests chemicals and reviews evidence for cancer.

**OSHA** is the federal Occupational Safety and Health Administration, which adopts and enforces health and safety standards.

**PEOSHA** is the New Jersey Public Employees Occupational Safety and Health Act, which adopts and enforces health and safety standards in public workplaces.

**Permeated** is the movement of chemicals through protective materials.

**ppm** means parts of a substance per million parts of air. It is a measure of concentration by volume in air.

**Protective Action Criteria (PAC)** are values established by the Department of Energy and are based on AEGLs and ERPGs. They are used for emergency planning of chemical release events.

A **reactive** substance is a solid, liquid or gas that releases energy under certain conditions.

**STEL** is a Short Term Exposure Limit which is usually a 15-minute exposure that should not be exceeded at any time during a work day.

A **teratogen** is a substance that causes birth defects by damaging the fetus.

**UEL or Upper Explosive Limit** is the highest concentration in air above which there is too much fuel (gas or vapor) to begin a reaction or explosion.

**Vapor Density** is the ratio of the weight of a given volume of one gas to the weight of another (usually Air), at the same temperature and pressure.

The **vapor pressure** is a force exerted by the vapor in equilibrium with the solid or liquid phase of the same substance. The higher the vapor pressure the higher concentration of the substance in air.
**Common Name:** PROPIONIC ACID

**Synonyms:** Ethylformic Acid; Methylacetic Acid; Propanoic Acid

**CAS No:** 79-09-4

**Molecular Formula:** C₃H₆O₂

**RTK Substance No:** 1599

**Description:** Colorless, oily liquid with a strong, unpleasant odor

### HAZARD DATA

<table>
<thead>
<tr>
<th>Hazard Rating</th>
<th>Firefighting</th>
<th>Reactivity</th>
</tr>
</thead>
<tbody>
<tr>
<td>3 - Health</td>
<td>COMBUSTIBLE LIQUID</td>
<td>Propionic Acid reacts violently and explosively with OXIDIZING AGENTS (such as PERCHLORATES, PEROXIDES, PERMANGANATES, CHLORATES, NITRATES, CHLORINE, BROMINE and FLUORINE) and PHOSPHORUS TRICHLORIDE.</td>
</tr>
<tr>
<td>2 - Fire</td>
<td>Use dry chemical, CO₂, water spray or alcohol-resistant foam as extinguishing agents. POISONOUS GASES ARE PRODUCED IN FIRE. Use water spray to keep fire-exposed containers cool. Flow or agitation may generate electrostatic charges. Propionic Acid may form an ignitable vapor/air mixture in closed tanks or containers.</td>
<td></td>
</tr>
<tr>
<td>0 - Reactivity</td>
<td></td>
<td>Propionic Acid may react violently with STRONG BASES (such as SODIUM HYDROXIDE and POTASSIUM HYDROXIDE); REDUCING AGENTS (such as LITHIUM, SODIUM, ALUMINUM and their HYDRIDES); and AMINES.</td>
</tr>
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</table>

**DOT#: UN 1848**

**ERG Guide #: 132**

**Hazard Class:** 8 (Corrosive)

### SPILL/LEAKS

<table>
<thead>
<tr>
<th>Isolation Distance:</th>
</tr>
</thead>
<tbody>
<tr>
<td>Spill: 50 meters (150 feet)</td>
</tr>
<tr>
<td>Fire: 800 meters (1/2 mile)</td>
</tr>
</tbody>
</table>

Absorb liquids in dry sand, earth, or a similar material and place into sealed containers for disposal. Keep Propionic Acid out of confined spaces, such as sewers, because of the possibility of an explosion. DO NOT wash into sewer. Neutralize water spills with lime or soda ash. Propionic Acid is harmful to aquatic organisms.

### PHYSICAL PROPERTIES

<table>
<thead>
<tr>
<th>Property</th>
<th>Value</th>
</tr>
</thead>
<tbody>
<tr>
<td>Odor Threshold</td>
<td>0.026 to 0.17 ppm</td>
</tr>
<tr>
<td>Flash Point</td>
<td>126°F (52°C)</td>
</tr>
<tr>
<td>LEL</td>
<td>2.9%</td>
</tr>
<tr>
<td>UEL</td>
<td>12.1%</td>
</tr>
<tr>
<td>Auto Ignition Temp</td>
<td>869°F to 955°F (465°C to 513°C)</td>
</tr>
<tr>
<td>Vapor Density</td>
<td>2.6 (air = 1)</td>
</tr>
<tr>
<td>Vapor Pressure</td>
<td>2 mm Hg at 68°F (20°C)</td>
</tr>
<tr>
<td>Specific Gravity</td>
<td>1 (water = 1)</td>
</tr>
<tr>
<td>Water Solubility</td>
<td>Soluble</td>
</tr>
<tr>
<td>Boiling Point</td>
<td>286°F (141°C)</td>
</tr>
<tr>
<td>Freezing Point</td>
<td>-60°F (-21°C)</td>
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<tr>
<td>Ionization Potential</td>
<td>10.24 eV</td>
</tr>
<tr>
<td>Molecular Weight</td>
<td>74.08</td>
</tr>
</tbody>
</table>

### PROTECTIVE EQUIPMENT

| Gloves:          | Butyl and Teflon® (>4-hr breakthrough) |
| Coveralls:       | Tychem® Responder® (>8-hr breakthrough) |
| Respirator:      | >10 ppm - full facepiece APR with Organic Vapor filters |
|                  | >100 ppm - SCBA |

### HEALTH EFFECTS

| Eyes:            | Severe irritation, burns and possible eye damage |
| Skin:            | Severe irritation and burns |
| Inhalation:      | Nose, throat and lung irritation, with coughing, wheezing and shortness of breath Headache, nausea and abdominal pain |

### FIRST AID AND DECONTAMINATION

Remove the person from exposure. Flush eyes with large amounts of water for at least 30 minutes. Remove contact lenses if worn. Seek medical attention. Quickly remove contaminated clothing and wash contaminated skin with large amounts of water. Seek medical attention. Begin artificial respiration if breathing has stopped and CPR if necessary. Transfer promptly to a medical facility.