

# Right to Know Hazardous Substance Fact Sheet



Common Name: SULFURYL CHLORIDE

Synonyms: Chlorosulfuric Acid; Sulfuric Dichloride; Sulfur Oxychloride

CAS No: 7791-25-5

Molecular Formula: SO<sub>2</sub>Cl<sub>2</sub> RTK Substance No: 1768

Description: Colorless liquid with a strong, irritating odor

HAZARD DATA		
Hazard Rating	Firefighting	Reactivity
3 - Health	CORROSIVE AND WATER REACTIVE Extinguish fire using an agent suitable	<b>Sulfuryl Chloride</b> reacts with WATER or MOIST AIR to form toxic and corrosive gases such as <i>Hydrogen Chloride</i> and <i>Sulfuric Acid</i> .
0 - Fire	for type of surrounding fire. Sulfuryl	Sulfuryl Chloride can react explosively with LEAD DIOXIDE and
2W - Reactivity	Chloride itself does not burn.  DO NOT USE WATER.	ETHERS (when in the presence of METAL SALTS).  Sulfuryl Chloride is not compatible with OXIDIZING AGENTS (such
DOT#: UN 1834	POISONOUS GASES ARE	as PERCHLORATES, PEROXIDES, PERMANGANATES,
ERG Guide #: 137	PRODUCED IN FIRE, including	CHLORATES, NITRATES, CHLORINE, BROMINE and FLUORINE); STRONG ACIDS (such as HYDROCHLORIC, SULFURIC and
Hazard Class: 8 (Corrosive)	Chlorine, Hydrogen Chloride, and Sulfur Oxides.  Sulfuryl Chloride may ignite combustibles (wood, paper and oil).	NITRIC); STRONG BASES (such as SODIUM HYDROXIDE and POTASSIUM HYDROXIDE); ALCOHOLS; and AMINES.
		<b>Sulfuryl Chloride</b> attacks many METALS in the presence of WATER.

### SPILL/LEAKS

#### **Isolation Distance:**

Small Spill: 30 meters (100 feet) Large Spill: 100 meters (300 feet) Fire: 800 meters (1/2 mile)

Absorb liquids in vermiculite, dry sand, earth, or a similar material and place into sealed containers for disposal.

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DO NOT USE WATER OR WET METHOD.

DO NOT wash into sewer.

**Sulfuryl Chloride** may be hazardous to the environment, especially water systems.

## PHYSICAL PROPERTIES

Odor Threshold: Strong, irritating odor

Flash Point: Nonflammable Vapor Density: 4.6 (air = 1)

Vapor Pressure: 105 mm Hg at 68°F (20°C)

Specific Gravity: 1.67 (water = 1)
Water Solubility: Decomposes/Reacts

**Boiling Point:** 156°F (69°C) **Freezing Point:** -65°F (-54°C) **Molecular Weight:** 134.96

## **EXPOSURE LIMITS**

No occupational exposure limits have been established for **Sulfuryl Chloride**.

The Protective Action Criteria values are:

 $PAC-1 = 0.3 ppm \quad PAC-2 = 3.7 ppm \quad PAC-3 = 11ppm$ 

### PROTECTIVE EQUIPMENT

Gloves: Barrier® (>8-hr breakthrough for *Inorganic Halides*)

**Coveralls:** Tychem® BR, Responder®, and TK (>8-hr breakthrough)

Respirator: SCBA

#### HEALTH EFFECTS

**Eyes:** Severe irritation, burns and possible eye

damage

**Skin:** Severe irritation and burns

Inhalation: Nose, throat and lung irritation with

coughing, and severe shortness of

breath (pulmonary edema)

# FIRST AID AND DECONTAMINATION

Remove the person from exposure.

**Flush** eyes with large amounts of water for at least 30 minutes. Remove contact lenses if worn. Seek medical attention immediately.

**Quickly** remove contaminated clothing and wash contaminated skin with large amounts of soap and water. Seek medical attention immediately.

**Begin** artificial respiration if breathing has stopped and CPR if necessary.

**Transfer** promptly to a medical facility.

**Medical** observation is recommended as symptoms may be delayed.